

第二章 均匀物质的热力学性质

1. 基本热力学函数

2. 麦氏关系及应用

3. 气体节流和绝热膨胀

§ 2.1 内能、焓、自由能和吉布斯函数的全微分

一、热力学函数

1.基本热力学函数

温度 T : 宏观定义和微观定义

内能 U : 宏观定义和微观定义

熵 S : 宏观定义和微观定义

2.辅助热力学函数

焓: $H=U+pV$

自由能: $F=U-TS$

吉布斯函数: $G=U-TS+pV$

二、热力学函数的全微分

利用热力学基本等式和H.F.G的定义式得到

$$\left\{ \begin{aligned} dU &= TdS - pdV \\ dH &= TdS + Vdp \\ dF &= -SdT - pdV \\ dG &= -SdT + Vdp \end{aligned} \right.$$

记忆方法:

$$F = U - TS$$

$$dF = dU - TdS - SdT$$

$$= TdS - pdV - TdS - SdT$$

$$\therefore dF = -SdT - pdV$$

$$\left\{ \begin{aligned} TdS \\ -SdT \\ +Vdp \\ -pdV \end{aligned} \right.$$

$$\left\{ \begin{aligned} dU &= TdS - pdV \\ dH &= TdS + Vdp \\ dG &= -SdT + Vdp \end{aligned} \right.$$

记忆方法:

$$G = U - TS + pV$$

$$dG = dU - TdS - SdT + pdV + Vdp$$

$$= TdS - pdV - TdS - SdT + pdV + Vdp$$

$$dG = -SdT + Vdp$$

$$\left\{ \begin{aligned} +Vdp \\ -pdV \\ +Vdp \end{aligned} \right.$$

三、麦克斯韦关系推导

1. 内能

$$dU = TdS - pdV$$

$$U = U(S, V), \quad dU = \left(\frac{\partial U}{\partial S}\right)_V dS + \left(\frac{\partial U}{\partial V}\right)_S dV$$

$$T = \left(\frac{\partial U}{\partial S}\right)_V = T(S, V), \quad p = -\left(\frac{\partial U}{\partial V}\right)_S = p(S, V)$$

$$\frac{\partial^2 U}{\partial S \partial V} = \left(\frac{\partial T}{\partial V}\right)_S, \quad \frac{\partial^2 U}{\partial V \partial S} = -\left(\frac{\partial p}{\partial S}\right)_V$$

$$\frac{\partial^2 U}{\partial V \partial S} = \frac{\partial^2 U}{\partial S \partial V}$$

$$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_V$$

2. 焓 $H = U + pV$

$$\boxed{dH = TdS + Vdp}$$

$$H = H(S, p), \quad dH = \left(\frac{\partial H}{\partial S}\right)_p dS + \left(\frac{\partial H}{\partial p}\right)_S dp$$
$$T = \left(\frac{\partial H}{\partial S}\right)_p = T(S, p), \quad V = \left(\frac{\partial H}{\partial p}\right)_S = V(S, p)$$

$$\frac{\partial^2 H}{\partial S \partial p} = \frac{\partial T}{\partial p}, \quad \frac{\partial^2 H}{\partial p \partial S} = \frac{\partial V}{\partial S}$$

$$\frac{\partial^2 H}{\partial p \partial S} = \frac{\partial^2 H}{\partial S \partial p}$$

$$\boxed{\left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p}$$

3. 自由能 $F = U - TS$

$$dF = -SdT - pdV$$

$$F = F(T, V), \quad dF = \left(\frac{\partial F}{\partial T}\right)_V dT + \left(\frac{\partial F}{\partial V}\right)_T dV$$

$$S = -\left(\frac{\partial F}{\partial T}\right)_V = S(T, V), \quad p = -\left(\frac{\partial F}{\partial V}\right)_T = p(T, V)$$

$$\frac{\partial^2 F}{\partial V \partial T} = \frac{\partial^2 F}{\partial T \partial V}$$

$$\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial p}{\partial T}\right)_V$$

$$U = F + TS = F - T\left(\frac{\partial F}{\partial T}\right)_V$$

$$H = U + pV = F - T\left(\frac{\partial F}{\partial T}\right)_V - V\left(\frac{\partial F}{\partial V}\right)_T$$

4. 吉布斯函数（自由焓） $G = H - TS = F + pV$

$$dG = -SdT + Vdp$$

$$G = G(T, p), \quad dG = \left(\frac{\partial G}{\partial T} \right)_p dT + \left(\frac{\partial G}{\partial p} \right)_T dp$$

$$S = - \left(\frac{\partial G}{\partial T} \right)_p = S(T, p), \quad V = \left(\frac{\partial G}{\partial p} \right)_T = V(T, p)$$

$$\frac{\partial^2 G}{\partial p \partial T} = \frac{\partial^2 G}{\partial T \partial p}$$

$$\left(\frac{\partial S}{\partial p} \right)_T = - \left(\frac{\partial V}{\partial T} \right)_p$$

$$H = G + TS = G - T \left(\frac{\partial G}{\partial T} \right)_p$$

$$U = H - pV = G - T \left(\frac{\partial G}{\partial T} \right)_p - p \left(\frac{\partial G}{\partial p} \right)_T$$

麦克斯韦关系

$$\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial p}{\partial S}\right)_U$$

U

$$\left(\frac{\partial T}{\partial p}\right)_S = \left(\frac{\partial V}{\partial S}\right)_p$$

H

$(-T)$	S
$(-p)$	V

F

$$\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial p}{\partial T}\right)_V$$

G

$$\left(\frac{\partial S}{\partial p}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_p$$

§ 2.2 麦氏关系的简单应用

麦克斯韦关系的应用有：

- (1)用实验可测量的量（如状态方程，热容量 C_p 、 C_v 、膨胀系数 α 、压缩系数 κ_r 等）表示不能直接测量的量（如 U 、 H 、 F 、 G 等）（§ 2.2的内容）
- (2)用实验可以测量的量表示某些物理效应及物理量的变化率（§ 2.3的内容）
- (3)求基本热力学函数和特性函数，进而求出所有热力学函数（§ 2.3、§ 2.4的内容）
- (4)讨论某些物质的热力学性质（§ 2.6、§ 2.7的内容）

1. 能态方程与 C_V

选择 T, V 为独立变量，内能和熵均可写成态变量 T 和 V 的函数

内能 $dU = TdS - pdV$

$$S = S(T, V), \quad dS = \left(\frac{\partial S}{\partial T} \right)_V dT + \left(\frac{\partial S}{\partial V} \right)_T dV$$

$$dU = T \left(\frac{\partial S}{\partial T} \right)_V dT + \left[T \left(\frac{\partial S}{\partial V} \right)_T - p \right] dV$$

$$\left(\frac{\partial S}{\partial V} \right)_T = \left(\frac{\partial p}{\partial T} \right)_V$$

$$U = U(T, V), \quad dU = \left(\frac{\partial U}{\partial T} \right)_V dT + \left(\frac{\partial U}{\partial V} \right)_T dV$$

$$C_V = \left(\frac{\partial U}{\partial T} \right)_V = T \left(\frac{\partial S}{\partial T} \right)_V$$

$$\left(\frac{\partial U}{\partial V} \right)_T = T \left(\frac{\partial p}{\partial T} \right)_V - p$$

$$dU = C_V dT + \left[T \left(\frac{\partial p}{\partial T} \right)_V - p \right] dV$$

$$dS = \frac{C_V}{T} dT + \left(\frac{\partial p}{\partial T} \right)_V dV$$

讨论：

$$\left(\frac{\partial U}{\partial V}\right)_T = T\left(\frac{\partial p}{\partial T}\right)_V - p$$

(1) 对于理想气体， $pV = nRT$

显然有， $\left(\frac{\partial U}{\partial V}\right)_T = 0$ 焦耳定律的结果

(2) 对于范氏气体（1 mol）

$$\left(p + \frac{a}{v^2}\right)(v - b) = RT$$

$$\text{则有：} \left(\frac{\partial U}{\partial v}\right)_T = T \frac{R}{v - b} - p = \frac{a}{v^2}$$

实际气体的内能不仅与温度有关，而且与体积有关。

2. 焓态方程与 C_p

以 T, p 为独立变量, 则有

焓 $\boxed{dH = TdS + Vdp}$

$$S = S(T, p), \quad dS = \left(\frac{\partial S}{\partial T}\right)_p dT + \left(\frac{\partial S}{\partial p}\right)_T dp$$

$$dH = T \left(\frac{\partial S}{\partial T}\right)_p dT + \left[T \left(\frac{\partial S}{\partial p}\right)_T + V \right] dp$$

$$\boxed{\left(\frac{\partial S}{\partial p}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_p}$$

$$H = H(T, p), \quad dH = \left(\frac{\partial H}{\partial T}\right)_p dT + \left(\frac{\partial H}{\partial p}\right)_T dp$$

$$C_p = \left(\frac{\partial H}{\partial T}\right)_p = T \left(\frac{\partial S}{\partial T}\right)_p$$

$$\left(\frac{\partial H}{\partial p}\right)_T = V - T \left(\frac{\partial V}{\partial T}\right)_p$$

$$\boxed{dH = C_p dT + \left[V - T \left(\frac{\partial V}{\partial T}\right)_p \right] dp}$$

$$\boxed{dS = \frac{C_p}{T} dT - \left(\frac{\partial V}{\partial T}\right)_p dp}$$

* 雅可比行列式

设 u, v 是自变量 (x, y) 的函数 $u(x, y), v(x, y)$

雅可比行列式定义:

$$\frac{\partial(u, v)}{\partial(x, y)} = \begin{vmatrix} \left(\frac{\partial u}{\partial x}\right)_y & \left(\frac{\partial u}{\partial y}\right)_x \\ \left(\frac{\partial v}{\partial x}\right)_y & \left(\frac{\partial v}{\partial y}\right)_x \end{vmatrix} = \left(\frac{\partial u}{\partial x}\right)_y \left(\frac{\partial v}{\partial y}\right)_x - \left(\frac{\partial u}{\partial y}\right)_x \left(\frac{\partial v}{\partial x}\right)_y$$

性质:

$$\begin{aligned} 1. \quad \left(\frac{\partial u}{\partial x}\right)_y &= \frac{\partial(u, y)}{\partial(x, y)} & 2. \quad \frac{\partial(u, v)}{\partial(x, y)} &= -\frac{\partial(v, u)}{\partial(x, y)} = -\frac{\partial(u, v)}{\partial(y, x)} \\ 3. \quad \frac{\partial(u, v)}{\partial(x, y)} &= \frac{\frac{\partial(u, v)}{\partial(r, s)}}{\frac{\partial(x, y)}{\partial(r, s)}} & 4. \quad \frac{\partial(u, v)}{\partial(x, y)} &= \frac{1}{\frac{\partial(x, y)}{\partial(u, v)}} \end{aligned}$$

试求，简单系统的 $C_p - C_V = ?$

法一：用隐函数的方法

$$C_p - C_V = T \left[\left(\frac{\partial S}{\partial T} \right)_p - \left(\frac{\partial S}{\partial T} \right)_V \right] \quad \text{熵可写成 } S(T, p) = S(T, V(T, p))$$

$$S = S(T, V), \quad dS = \left(\frac{\partial S}{\partial T} \right)_V dT + \left(\frac{\partial S}{\partial V} \right)_T dV$$

$$S = S(T, p), \quad dS = \left(\frac{\partial S}{\partial T} \right)_p dT + \left(\frac{\partial S}{\partial p} \right)_T dp$$

$$V = V(T, p), \quad dV = \left(\frac{\partial V}{\partial T} \right)_p dT + \left(\frac{\partial V}{\partial p} \right)_T dp$$

$$\left(\frac{\partial S}{\partial T} \right)_p = \left(\frac{\partial S}{\partial T} \right)_V + \left(\frac{\partial S}{\partial V} \right)_T \left(\frac{\partial V}{\partial T} \right)_p$$

$$\left(\frac{\partial S}{\partial V} \right)_T = \left(\frac{\partial p}{\partial T} \right)_V$$

$$C_p - C_V = T \left(\frac{\partial p}{\partial T} \right)_V \left(\frac{\partial V}{\partial T} \right)_p$$

由物态方程决定。

法二：雅可比行列式

$$\begin{aligned}
 C_p - C_V &= T \left[\left(\frac{\partial S}{\partial T} \right)_p - \left(\frac{\partial S}{\partial T} \right)_V \right] = T \left[\left(\frac{\frac{\partial(S, p)}{\partial(T, V)}}{\frac{\partial(T, p)}{\partial(T, V)}} \right)_p - \left(\frac{\partial S}{\partial T} \right)_V \right] \\
 &= T \left[\frac{1}{\left(\frac{\partial p}{\partial V} \right)_T} \begin{vmatrix} \left(\frac{\partial S}{\partial T} \right)_V & \left(\frac{\partial S}{\partial V} \right)_T \\ \left(\frac{\partial p}{\partial T} \right)_V & \left(\frac{\partial p}{\partial V} \right)_T \end{vmatrix} - \left(\frac{\partial S}{\partial T} \right)_V \right] \\
 &= T \left\{ \frac{1}{\left(\frac{\partial p}{\partial V} \right)_T} \left[\left(\frac{\partial S}{\partial T} \right)_V \left(\frac{\partial p}{\partial V} \right)_T - \left(\frac{\partial S}{\partial V} \right)_T \left(\frac{\partial p}{\partial T} \right)_V \right] - \left(\frac{\partial S}{\partial T} \right)_V \right\} \\
 &= - \frac{T}{\left(\frac{\partial p}{\partial V} \right)_T} \left(\frac{\partial S}{\partial V} \right)_T \left(\frac{\partial p}{\partial T} \right)_V = - \frac{T}{\left(\frac{\partial p}{\partial V} \right)_T} \left(\frac{\partial p}{\partial T} \right)_V^2 \quad \text{书上例二的结果}
 \end{aligned}$$

$$C_p - C_V = -\frac{T}{\left(\frac{\partial p}{\partial V}\right)_T} \left(\frac{\partial p}{\partial T}\right)_V^2 = -\frac{T}{\left(\frac{\partial(p,T)}{\partial(V,T)}\right)} \left(\frac{\partial p}{\partial T}\right)_V \frac{\partial(p,V)}{\partial(T,V)}$$

$$= T \left(\frac{\partial p}{\partial T}\right)_V \frac{\partial(p,V)}{\partial(p,T)}$$

$$C_p - C_V = T \left(\frac{\partial p}{\partial T}\right)_V \left(\frac{\partial V}{\partial T}\right)_p$$

体胀系数

$$\alpha = \frac{1}{V} \left(\frac{\partial V}{\partial T}\right)_p$$

等温压缩系数

$$\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial p}\right)_T$$

$$\left(\frac{\partial V}{\partial p}\right)_T = -\left(\frac{\partial T}{\partial p}\right)_V / \left(\frac{\partial T}{\partial V}\right)_p$$

$$\left(\frac{\partial p}{\partial T}\right)_V = -\left(\frac{\partial V}{\partial T}\right)_p / \left(\frac{\partial V}{\partial p}\right)_T$$

$$C_p - C_V = \frac{VT\alpha^2}{\kappa_T} \geq 0$$

$$\gamma \geq 1$$

以上几式，对于任意简单系统均适用。但 $C_p - C_V = nR$ 只是理想气体的结论。

等温和绝热压缩系数(书上例一)

$$\kappa_T = -\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_T$$

$$\kappa_S = -\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_S$$

$$\frac{\kappa_T}{\kappa_S} = \frac{-\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_T}{-\frac{1}{V} \left(\frac{\partial V}{\partial p} \right)_S} = \frac{\frac{\partial(V, T)}{\partial(p, T)}}{\frac{\partial(V, S)}{\partial(p, S)}} = \frac{\frac{\partial(V, T)}{\partial(p, T)}}{\frac{\partial(V, S)}{\partial(p, S)}} = \frac{\frac{\partial(p, S)}{\partial(V, S)}}{\frac{\partial(V, T)}{\partial(V, S)}} = \frac{\left(\frac{\partial S}{\partial T} \right)_p}{\left(\frac{\partial S}{\partial T} \right)_V} = \frac{C_p}{C_V}$$

$$\frac{\kappa_T}{\kappa_S} = \frac{C_p}{C_V} = \gamma$$

平衡稳定性要求：以上四量皆为正。

作业：

P73 2.1题， 2.2题， 2.3题， 2.4题