

Local Relativistic Exact Decoupling

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Abstract

We present a systematic hierarchy of approximations for *local* exact-decoupling of four-component quantum chemical Hamiltonians based on the Dirac equation. Our ansatz reaches beyond the trivial local approximation that is based on a unitary transformation of only the atomic block-diagonal part of the Hamiltonian. Systematically, off-diagonal Hamiltonian matrix blocks can be subjected to a unitary transformation to yield relativistically corrected matrix elements. The full hierarchy is investigated with respect to the accuracy reached for the electronic energy and molecular properties on a balanced test molecule set that comprises molecules with heavy elements in different bonding situations. Our atomic (local) assembly of the unitary transformation needed for exact decoupling provides an excellent local approximation for any relativistic exact-decoupling approach. Its order- N^2 scaling can be further reduced to linear scaling by employing the neighboring-atomic-blocks approximation. Therefore, it is an efficient relativistic method perfectly well suited for relativistic calculations on large molecules. If a large molecule contains many light atoms (typically hydrogen atoms), the computational costs can be further reduced by employing a well-defined non-relativistic approximation for these light atoms without significant loss of accuracy.

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I. INTRODUCTION

The consideration of relativistic effects is essential to the proper understanding of the chemistry of heavy-element containing molecules [1, 2]. The Dirac equation provides the relativistic quantum mechanical description of a single electron in the presence of external electromagnetic potentials. As the Dirac operator comprises four-dimensional operators, its eigenfunctions are four-component. The relativistic many-electron Hamiltonian is composed of the one-electron Dirac operators and two-electron interaction operators. The Coulomb interaction turned out to be sufficiently accurate for the description of chemical phenomena and the corresponding many-electron Hamiltonian is called the Dirac–Coulomb Hamiltonian.

A relativistic method based on the Dirac–Coulomb Hamiltonian and four-component one-electron states to calculate the total energy and other physical properties of a many-electron atom or molecule is called a four-component method. A four-component method also yields negative-energy solutions, which are pathologic and of no use in chemistry. Also, a large number of basis functions is required to properly describe the negative-energy states. These are two drawbacks of four-component methods that motivated the development of two-component methods, which remove the negative energy solutions and could in principle exactly reproduce the results of four-component calculation [2].

Several relativistic two-component methods were developed in the past decades. For example, one of the widely used approaches is the second-order Douglas–Kroll–Hess method (DKH2) [3, 4]. It employs the free-particle Foldy–Wouthuysen (FW) [5] transformation as well as sequential Douglas–Kroll [6] transformations to decouple the four-component operator. High order (more than 2) [7–10] and even arbitrary order [11–16] DKH methods have been developed. The zeroth-order regular approximation (ZORA) [17–19] is another highly successful relativistic two-component method. Within the ZORA framework, it is particularly easy to implement the calculation of molecular properties; see Refs. [20, 21] for very recent examples and Ref. [22] for a review.

In recent years it was shown that, if the goal is to exactly decouple the four-component Hamiltonian matrix instead of the Hamiltonian operator, the formulae of exact decoupling can be directly obtained. The Barysz–Sadlej–Snijders (BSS) [23–26] method aims to exactly decouple the free-particle FW-transformed four-component Hamiltonian matrix by a matrix operator of the form derived in Ref. [27]. The key matrix operator, which is used to construct the decoupling transformation matrix, is obtained by solving an iterative equation. However, invoking

the free-particle FW transformation turns out to be not necessary for the construction of the exact decoupling transformation which has led to the formulation of a one-step protocol. The pioneering work of this one-step transformation was provided by Dyall [28–32] in the form of the so-called normalized elimination of the small component (NESC) approach. Later it was generalized to the so-called X2C method by several groups [33–44]. Since the DKH method also employs matrix techniques to evaluate the Douglas-Kroll transformation, it is also able to exactly decouple the four-component Hamiltonian matrix. This has been shown within the arbitrary-order approach [11–14, 16, 45]. For reviews of these developments see Refs. [46–52]

Even within a scalar relativistic approximation, where the spin-dependent operators are neglected, the construction of the relativistic Hamiltonian matrix dominates the step of the one-electron integral evaluation. This is the case because the relativistic transformation is done within an un-contracted basis set and involves many matrix manipulations such as multiplication and diagonalization. For a large molecule including only one or a few heavy atoms, this would waste computational resources since the relativistic effects are highly local and thus only significant for heavy atoms. It is therefore clear that a local relativistic method is desirable. Even for a molecule containing only heavy atoms, such a local relativistic approach would be preferable as we shall see. This is evident from the relativistic effective core potential (ECP) approach or from the frozen-core approximation which turned out to be accurate approaches for the calculation of valence electron properties [53]. However, a standard ECP cannot respond to the change of environment and we cannot obtain accurate results for the properties which require the electronic wavefunction in the core region of atoms. This also holds true for the frozen-core approximation. Therefore, an all-electron local relativistic method is desired.

The relativistic Hamiltonian operator is universal in coordinate space and a local approach is unfeasible at the operator level. However, since most of quantum chemical calculations apply a linear combination of atom-centered basis functions, we may employ these basis functions to construct atomic projectors as it is done in charge and spin population analysis [54]. For those Hamiltonian matrix blocks for which relativistic corrections are important, we shall derive a local relativistic Hamiltonian matrix to evaluate it. The locality is then exploited in the basis-function space instead of coordinate space. A crucial aspect to determine will be which block of the Hamiltonian matrix requires a relativistic correction and how to evaluate it. Undoubtedly, the heavy atom diagonal blocks will need a relativistic description, while the relativistic correction of heavy atom off-diagonal blocks depends on their contribution to physical observables and on

the feasibility of applying the relativistic correction. For relativistic transformation techniques, such as DKH and X2C, it is impossible to apply the relativistic transformation to an off-diagonal block alone without information about other blocks. It requires the information of a full square block to construct the relativistic transformation. Such a difficulty disappears if one employs an operator-based relativistic methods such as ZORA. The off-diagonal block of Hamiltonian matrix can be directly computed from relativistic integrals. However, the ZORA method provides a very poor description of atomic inner-shell orbitals, which carry the largest part of relativistic effects. In this sense, the exact decoupling methods are the best candidate for applying a local relativistic scheme since they do not neglect important relativistic corrections.

All current exact decoupling methods employ the transformation technique. It is therefore not trivial to apply the relativistic correction to heavy-atom off-diagonal Hamiltonian matrix elements. However, in this work we propose that if the local approximation is applied to the decoupling transformation instead of the Hamiltonian matrix, one could obtain results which include the relativistic correction to heavy atom off-diagonal blocks. We also note that the main effort for the construction of the relativistic Hamiltonian matrix with exact-decoupling methods is due to the evaluation of the decoupling transformation matrices. Therefore, a local approximation to the decoupling transformation would also lead to a significant reduction of the computational cost.

II. RELATIVISTICALLY LOCAL STRUCTURE OF THE HAMILTONIAN MATRIX

The following notation will be used throughout our discussions. Upper case labels A, B, C denote heavy atoms which require relativistic corrections, while lower case labels a, b, c denote light atoms for which a (standard) non-relativistic description can be assumed to be accurate enough based on numerical evidence compiled within computational chemistry in the past decades. To be more precise, the former labels refer to nuclei with high nuclear charge numbers, while the latter refer to those with low nuclear charge numbers (typically with nuclear charges of less than two dozen protons). Of course, the distinction whether a nucleus is considered heavy or light bears some sort of arbitrariness and it will be made on the basis of the accuracy required in a quantum chemical calculation. In general, there are then five different types of Hamiltonian matrix blocks H_{AA} , H_{AB} , H_{Aa} , H_{aa} , and H_{ab} that need to be considered.

The evaluation of relativistic Hamiltonian matrix with exact-decoupling approaches requires

a set of square matrices. They are the non-relativistic kinetic energy matrix T ,

$$T_{ij} = \frac{1}{2} \langle \lambda_i | \mathbf{p}^2 | \lambda_j \rangle, \quad (1)$$

the external potential matrix V ,

$$V_{ij} = \langle \lambda_i | \mathcal{V} | \lambda_j \rangle, \quad (2)$$

and the overlap matrix S ,

$$S_{ij} = \langle \lambda_i | \lambda_j \rangle, \quad (3)$$

as well as an additional relativistic matrix W

$$W_{ij} = \frac{1}{4c^2} \langle \lambda_i | \boldsymbol{\sigma} \cdot \mathbf{p} \mathcal{V} \boldsymbol{\sigma} \cdot \mathbf{p} | \lambda_j \rangle, \quad (4)$$

(note that all expressions are given in Hartree atomic units, in which the rest mass of the electron takes a numerical value of one). In the above equations, λ_i denotes the i -th 2-spinor atom-centered basis function, \mathcal{V} the external potential, c the speed of light, $\boldsymbol{\sigma}$ the vector of Pauli spin matrices. The basis functions λ_i may be grouped according to the atomic nucleus to which they are assigned, which is key to the local approaches discussed in the following. The matrix representation of the relativistic Hamiltonian is then evaluated as a function of the above-mentioned matrices

$$H = H(T, V, W, S). \quad (5)$$

A straightforward idea for a trivial local approximation is to apply the evaluation of the relativistic Hamiltonian matrix only to atomic (i.e., diagonal) blocks

$$H_{AA} = f(T_{AA}, V_{AA}, W_{AA}, S_{AA}), \quad (6)$$

(instead of to the full matrix, but where A may also represent a group of atoms) and to ignore the relativistic correction to all off-diagonal blocks

$$H_{AB} = T_{AB} + V_{AB}. \quad (7)$$

Adding more atoms to a group A will improve the accuracy, but the computational advances will be lost when a group A becomes very large. This approach was employed in Ref. [55, 56] and explored in detail in Ref. [57] for the low-order DKH method. We refer to it as the *diagonal local approximation to the Hamiltonian* (DLH) matrix. It is clear that the DLH approximation can also be applied to relativistic exact-decoupling approaches. The DLH approximation works

well at large interatomic distances, but the difference to a reference energy increases with shorter distance.

However, relativistic corrections to off-diagonal blocks are also important especially when inter-atomic distances are short. The necessity of applying the relativistic corrections to off-diagonal blocks has already been discussed in Ref. [58] with a so-called two-center approximation. In Ref. [58], the DKH transformation was applied to pairs of atoms ($A \oplus B$)

$$H_{A\oplus B} = H(T_{A\oplus B}, V_{A\oplus B}, W_{A\oplus B}, S_{A\oplus B}), \quad (8)$$

with

$$H_{A\oplus B} = \begin{pmatrix} H_{AA} & H_{AB} \\ H_{BA} & H_{BB} \end{pmatrix}, \quad (9)$$

to obtain a relativistically corrected off-diagonal Hamiltonian matrix H_{AB} . Unfortunately, such a two-center approximation introduces an inconsistent treatment to diagonal blocks. For instance, H_{AA} can be obtained from either the pair ($A \oplus B$) or from another pair ($A \oplus C$).

So far, we have only discussed the AA and AB blocks which, by definition, require a relativistic description for both two atoms. Since relativistic corrections for light atoms are very small and can be neglected, H_{aa} and H_{ab} can be approximated by the non-relativistic Hamiltonian matrices

$$H_{aa} = T_{aa} + V_{aa}, \quad (10)$$

$$H_{ab} = T_{ab} + V_{ab}. \quad (11)$$

A relativistic scheme for the heavy–light hybrid off-diagonal blocks H_{Aa} should, however, be considered explicitly.

III. LOCAL DECOMPOSITION OF THE X-OPERATOR

The general expression of exact decoupling transformations can be written as

$$U = \begin{pmatrix} Z^+ \frac{1}{\sqrt{1+X^\dagger X}} & Z^+ \frac{1}{\sqrt{1+X^\dagger X}} X^\dagger \\ -Z^- \frac{1}{\sqrt{1+XX^\dagger}} X & Z^- \frac{1}{\sqrt{1+XX^\dagger}} \end{pmatrix} \equiv \begin{pmatrix} U^{LL} & U^{LS} \\ U^{SL} & U^{SS} \end{pmatrix}. \quad (12)$$

Z^+ and Z^- are two-component unitary operators, and only Z^+ is required for the evaluation of the electrons-only Hamiltonian. The X -operator generates the electronic small-component functions φ_S^\dagger from the large-component functions φ_L^\dagger

$$\varphi_S^\dagger = X\varphi_L^\dagger. \quad (13)$$

Two-dimensional Hamiltonians are then obtained by applying the unitary transformation of Eq. (12) to blockdiagonalize ('*bd*') the four-dimensional Dirac-based Hamiltonian D

$$H^{bd} = UDU^\dagger. \quad (14)$$

It yields both the electrons-only Hamiltonian H (as the upper left block in H^{bd}) and the one for negative-energy solutions, although the latter will be discarded.

As discussed by Dyall in Ref. [29] for the NESC method, the Breit-Pauli approximation to the X matrix, which is used to construct the relativistic transformation matrix, is

$$X = I, \quad (15)$$

where I denotes the identity matrix. Such $X = I$ approximation are not variational stable and thus cannot be used for variational calculation. However, one could employ this approximation for light atoms only

$$X_{aa} = I, \quad (16)$$

since the corrections to light atoms are small and may not affect variational calculations. This idea was suggested in Ref. [42] for the X2C method but no results were presented.

Although the lowest level approximation to X Eq. (15) did not provides useful results, the atomic approximation to the X matrix gave results with very small errors. As discussed by Dyall in Ref. [30, 31], the X matrix can be approximate as the direct sum of atomic blocks

$$X = X_{AA} \oplus X_{BB} \oplus \dots \quad (17)$$

This approximation resembles the linear combination of atomic four-spinors ansatz in four-component calculations. The same approach was also employed in Ref. [59] in the sense of an atomic approximation to the projection on electronic states. If the electronic states are further transformed to a two-component picture by a renormalization matrix, this gives rise to a local X2C method. As discussed in Refs. [38, 42], the local approximation to the X matrix Eq. (17) works well for spectroscopic constants of diatomic molecules.

However, there are some drawbacks of the local- X matrix approximation to the X2C method. Firstly, it only reduced the cost for the evaluation of X matrix, while the exact-decoupling transformation matrices are still of molecular dimension as the renormalization matrix is not atomic block diagonal even if the X matrix is. The computational demands for the evaluation

of the relativistic approximation are thus still tremendous for large molecules. Secondly, the $X = I$ approximation did not provide a satisfactory treatment of the Aa blocks, since it may suffer from variational collapse and Eqs. (10) and (11) cannot be fulfilled within this approximation.

IV. LOCAL APPROXIMATIONS TO THE EXACT-DECOUPLING TRANSFORMATION

In this article, we suggest an local approximation to the exact-decoupling transformation. We may approximate the unitary transformation by taking only the 'atomic' diagonal blocks (all off-diagonal blocks are then set to zero)

$$U = U_{AA} \oplus U_{BB} \oplus \dots, \quad (18)$$

where the atomic unitary transformations U_{AA} are obtained from the diagonalization of the corresponding D_{AA} blocks of matrix operator D . We denote this approximation to U as the *diagonal local approximation to the unitary decoupling transformation* (DLU).

If we now substitute the approximate unitary transformation of Eq. (18) into Eq. (14), the diagonal blocks,

$$H_{AA}^{bd} = U_{AA} D_{AA} U_{AA}^\dagger, \quad (19)$$

turn out to be the same as in the DLH approach, while the off-diagonal blocks then read

$$H_{AB}^{bd} = U_{AA} D_{AB} U_{BB}^\dagger, \quad (20)$$

which is to be compared with the expression

$$H_{AB}^{bd} = \sum_{I,J} U_{AI} D_{IJ} U_{JB}^\dagger, \quad (21)$$

where U has not been approximated (here, I and J run over all atomic blocks). Hence, the DLU approach also introduces a relativistic correction to off-diagonal 'interatomic' blocks when compared with the DLH approximation.

The cost for the assembly of the unitary transformation U within the DLU approach is then of order N , where N measures the system size, namely the number of atoms. It is linear scaling since the atomic approximation is directly applied to the unitary transformation. However, the next step of applying the unitary decoupling transformation to obtain the relativistic Hamiltonian

matrices is no longer linear scaling. If no local approximation was applied for the unitary decoupling transformation, according to Eq. (21) where the summation indices I, J run over all atomic blocks, the calculation of H_{AB} matrix will require $2N^2$ number of matrix multiplications. Since the number of matrices to be calculated is of order N^2 , the total cost of relativistic transformation without local approximation is therefore of order N^4 . If the DLU approximation is applied, no summation will be needed and only two matrix multiplications will be required for each heavy-heavy block H_{AB} . The total cost is then of order N^2 .

If the distance of two atoms A and B is sufficiently large, the relativistic correction to H_{AB} can be neglected. Thus, we may define neighboring atomic pairs according to their distances. Then, only the Hamiltonian matrix H_{AB}^{bd} of neighboring pairs requires the transformation

$$H_{AB}^{bd} = U_{AA} D_{AB} U_{BB}^\dagger, \quad \forall A, B \text{ being neighbors}, \quad (22)$$

whereas all other pairs are simply taken in their non-relativistic form. Since the number of neighboring pairs is usually a linear function of system size. The total cost is then reduced to order N .

To investigate the relativistic correction to hybrid H_{Aa} blocks of the electrons-only Hamiltonian H , we need to consider the explicit structure of the unitary decoupling transformation to yield H

$$H = \begin{pmatrix} U^{LL}, & U^{LS} \end{pmatrix} \begin{pmatrix} V & T \\ T & (W-T) \end{pmatrix} \begin{pmatrix} U^{LL,\dagger} \\ U^{LS,\dagger} \end{pmatrix}, \quad (23)$$

where only the electronic part of the exact-decoupling transformation represented by U^{LL} and U^{LS} [which are the upper part of Eq. (12)] is employed. From Eq. (23) it is easy to see that the electrons-only Hamiltonian reads

$$\begin{aligned} H = & U^{LS} T U^{LL,\dagger} + U^{LL} T U^{LS,\dagger} - U^{LS} T U^{LS,\dagger} \\ & + U^{LL} V U^{LL,\dagger} + U^{LS} W U^{LS,\dagger}. \end{aligned} \quad (24)$$

In the non-relativistic (NR) limit, where the speed of light approaches infinity, we have

$$U^{LL,\text{NR}} = I, \quad U^{LS,\text{NR}} = I, \quad \text{and} \quad W^{\text{NR}} = 0. \quad (25)$$

If we insert them into the form of relativistic electrons-only Hamiltonian (24), we arrive at the non-relativistic Hamiltonian matrix,

$$H^{\text{NR}} = T + V. \quad (26)$$

This suggests a solution for the relativistic treatment of the hybrid heavy–light blocks H_{Aa} , because we may set all light-atom diagonal blocks in the unitary transformation to identity matrices

$$U_{aa}^{LL} = I \quad \text{and} \quad U_{aa}^{LS} = I. \quad (27)$$

To reproduce the non-relativistic Hamiltonian matrices of light-atom-only blocks, we must set the W matrix of light atoms to zero,

$$W_{aa} = 0. \quad (28)$$

The W_{Aa} matrices are also set to zero

$$W_{Aa} = 0. \quad (29)$$

We denote this approximation for the treatment of hybrid blocks as the DLU(NR) approach. If we do not introduce any approximations to the W matrices, the transformation will yield a Hamiltonian that differs from the non-relativistic Hamiltonian matrix on the order of $1/c^2$, which corresponds to the order achieved in the Breit–Pauli approximation. We therefore denote it as the DLU(BP) approach.

The final local approximation to the unitary matrices are then

$$U^{LL} = U_{AA}^{LL} \oplus U_{BB}^{LL} \oplus \cdots \oplus U_{aa}^{LL} \oplus U_{bb}^{LL} \oplus \cdots, \quad (30)$$

$$U^{LS} = U_{AA}^{LS} \oplus U_{BB}^{LS} \oplus \cdots \oplus U_{aa}^{LS} \oplus U_{bb}^{LS} \oplus \cdots, \quad (31)$$

Where the atomic unitary transformation matrices of heavy atoms are evaluated by the relativistic exact decoupling approaches. The explicit expressions of Hamiltonian matrix blocks H_{AA} and H_{AB} are

$$\begin{aligned} H_{AA} = & U_{AA}^{LS} T_{AA} U_{AA}^{LL, \dagger} + U_{AA}^{LL} T_{AA} U_{AA}^{LS, \dagger} \\ & - U_{AA}^{LS} T_{AA} U_{AA}^{LS, \dagger} + U_{AA}^{LL} V_{AA} U_{AA}^{LL, \dagger} \\ & + U_{AA}^{LS} W_{AA} U_{AA}^{LS, \dagger}, \end{aligned} \quad (32)$$

$$\begin{aligned} H_{AB} = & U_{AA}^{LS} T_{AB} U_{BB}^{LL, \dagger} + U_{AA}^{LL} T_{AB} U_{BB}^{LS, \dagger} \\ & - U_{AA}^{LS} T_{AB} U_{BB}^{LS, \dagger} + U_{AA}^{LL} V_{AB} U_{BB}^{LL, \dagger} \\ & + U_{AA}^{LS} W_{AB} U_{BB}^{LS, \dagger}. \end{aligned} \quad (33)$$

If the DLU(NR) approach is employed, the unitary transform matrices of light atoms are simply set to identity matrices and the explicit form of H_{Aa} , H_{aa} , and H_{ab} is

$$H_{Aa} = U_{AA}^{LL}T_{Aa} + U_{AA}^{LL}V_{Aa}, \quad (34)$$

$$H_{aa} = T_{aa} + V_{aa}, \quad (35)$$

$$H_{ab} = T_{ab} + V_{ab}. \quad (36)$$

In the DLU(BP) approach, however, they read

$$H_{Aa} = U_{AA}^{LL}T_{Aa} + U_{AA}^{LL}V_{Aa} + U_{AA}^{LS}W_{Aa}, \quad (37)$$

$$H_{aa} = T_{aa} + V_{aa} + W_{aa}, \quad (38)$$

$$H_{ab} = T_{ab} + V_{ab} + W_{ab}. \quad (39)$$

The relativistic transformation of a property operator P [13, 14, 16, 45, 60, 61], however, requires an additional matrix Q and its matrix elements are given by

$$Q_{ij} = \frac{1}{4c^2} \langle \lambda_i | \boldsymbol{\sigma} \cdot \mathbf{p} \mathcal{P} \boldsymbol{\sigma} \cdot \mathbf{p} | \lambda_j \rangle. \quad (40)$$

And the relativistic corrected property matrix is

$$P = U^{LL} P U^{LL,\dagger} + U^{LS} Q U^{LS,\dagger}. \quad (41)$$

V. DLU EVALUATION WITHIN EXACT-DECOUPLING APPROACHES

A. The X2C approach

The X2C method employs

$$U_{\text{X2C}} = \begin{pmatrix} \frac{1}{\sqrt{1+X^\dagger X}} & \frac{1}{\sqrt{1+X^\dagger X}} X^\dagger \\ -\frac{1}{\sqrt{1+X X^\dagger}} X & \frac{1}{\sqrt{1+X X^\dagger}} \end{pmatrix}, \quad (42)$$

as the exact-decoupling unitary transformation. The unitary operators for the electronic states are then

$$U_{\text{X2C}}^{LL} = \frac{1}{\sqrt{1+X^\dagger X}}, \quad (43)$$

$$U_{\text{X2C}}^{LS} = U_{\text{X2C}}^{LL} X^\dagger = \frac{1}{\sqrt{1+X^\dagger X}} X^\dagger, \quad (44)$$

For an actual implementation, we need their matrix representations in a finite non-orthogonal space of the atom-centered basis functions. The X -matrix is evaluated by diagonalizing the following generalized eigenvalue equation

$$\begin{pmatrix} V & T \\ T & (W-T) \end{pmatrix} \begin{pmatrix} C_L^+ \\ C_S^+ \end{pmatrix} = \begin{pmatrix} S & 0 \\ 0 & \frac{1}{2c^2}T \end{pmatrix} \begin{pmatrix} C_L^+ \\ C_S^+ \end{pmatrix} \epsilon^+, \quad (45)$$

where C_L^+ and C_S^+ denote the large- and small-component molecular-orbital coefficients, respectively. We obtain the X matrix by the following equation

$$X = C_S^+ (C_L^+)^{-1}. \quad (46)$$

The matrix representation of U^{LL} and U^{LS} is then given by

$$U_{X2C}^{LL} = S^{1/2}(S^{-1/2}\tilde{S}S^{-1/2})^{-1/2}S^{-1/2}, \quad (47)$$

$$U_{X2C}^{LS} = U_{X2C}^{LL}X^\dagger, \quad (48)$$

with \tilde{S} defined as

$$\tilde{S} = S + \frac{1}{2c^2}X^\dagger T X. \quad (49)$$

In our atomic approximation to the unitary transformation U , we need to evaluate the above equation for each 'atomic' block of the original Hamiltonian in order to obtain the diagonal blocks U_{AA}^{LL} and U_{AA}^{LS} . The decoupling transformations are then assembled according to Eqs. (30) and (31) and the relativistic Hamiltonian matrix follows from Eq. (23).

B. The BSS approach

In the BSS approach, all matrices are transformed to an orthonormal basis-function space first. It is convenient to calculate the transformation matrix K by diagonalizing the non-relativistic kinetic energy matrix

$$TK = SKt, \quad (50)$$

since the eigenvalues t can be used for later evaluation of the free-particle FW transformation. The eigenvector matrix K has the following properties:

$$K^\dagger SK = I \quad \text{and} \quad K^\dagger TK = t. \quad (51)$$

The free-particle FW (fpFW) transformation features four diagonal block matrices

$$U_0 = \begin{pmatrix} U^{LL,\text{fpFW}} & U^{LS,\text{fpFW}} \\ U^{SL,\text{fpFW}} & U^{SS,\text{fpFW}} \end{pmatrix} = \begin{pmatrix} \sqrt{\frac{E_0+c^2}{2E_0}} & \sqrt{\frac{E_0-c^2}{2E_0}} f \\ -\sqrt{\frac{E_0-c^2}{2E_0}} & \sqrt{\frac{E_0+c^2}{2E_0}} f \end{pmatrix}, \quad (52)$$

with $E_0 = \sqrt{2tc^2 + c^4}$ and $f = \sqrt{2c^2/t}$. It is then applied to yield a transformed four-dimensional Hamiltonian matrix D_0

$$D_0 = U_0 \begin{pmatrix} K & 0 \\ 0 & K \end{pmatrix}^\dagger \begin{pmatrix} V & T \\ T & (W-T) \end{pmatrix} \begin{pmatrix} K & 0 \\ 0 & K \end{pmatrix} U_0^\dagger. \quad (53)$$

Next, the exact-decoupling BSS transformation,

$$U_1 = \begin{pmatrix} \frac{1}{\sqrt{1+R^\dagger R}} & \frac{1}{\sqrt{1+R^\dagger R}} R^\dagger \\ -\frac{1}{\sqrt{1+RR^\dagger}} R & \frac{1}{\sqrt{1+RR^\dagger}} \end{pmatrix}, \quad (54)$$

which has same structure as the exact-decoupling transformation U_{X2C} , is applied. The R matrix is obtained by diagonalizing the free-particle FW transformed four-component Hamiltonian matrix D_0 and employing a similar relation to the one in Eq. (46).

After the exact-decoupling BSS transformation has been carried out, the Hamiltonian matrix is back-transformed to the original (non-orthogonal) basis

$$H = \begin{pmatrix} K^{-1} & 0 \\ 0 & K^{-1} \end{pmatrix}^\dagger U_1 D_0 U_1^\dagger \begin{pmatrix} K^{-1} & 0 \\ 0 & K^{-1} \end{pmatrix}. \quad (55)$$

Consequently, we obtain the matrix form of U_{BSS}^{LL} and U_{BSS}^{LS} for the BSS approach as

$$U_{\text{BSS}}^{LL} = (K^{-1})^\dagger (1 + RR^\dagger)^{-\frac{1}{2}} (U^{LL,\text{fpFW}} + R^\dagger U^{SL,\text{fpFW}}) K^\dagger, \quad (56)$$

$$U_{\text{BSS}}^{LS} = (K^{-1})^\dagger (1 + RR^\dagger)^{-\frac{1}{2}} (U^{LS,\text{fpFW}} + R^\dagger U^{SS,\text{fpFW}}) K^\dagger. \quad (57)$$

The expressions for obtaining the relativistic Hamiltonian matrix (and property matrices) are then as given above.

C. The DKH approach

The DKH approach features the same initial transformation as the BSS approach to obtain the transformed Hamiltonian matrix D_0 . Subsequent decoupling transformations are expressed as

$$U^{(m)} = \prod_{k=m}^1 U_k \quad (58)$$

with the generalized parametrization of the U_k [9],

$$U_k = \sum_{i=0}^{\lfloor m/k \rfloor} a_{k,i} \mathbf{W}_k^i, \quad (59)$$

where \mathbf{W}_k are anti-hermitian matrix operators and m is related to the order of the DKH expansion. Since the expression of \mathbf{W}_k is too lengthy to be presented here, we refer the reader to Ref. [16] for details. If m is large (strictly, if it approaches infinity), exact decoupling is achieved. Usually, a low value for m is often sufficient for calculations of relative energies and valence-shell properties (i.e., $m=1$ for the original DKH2 approach).

With the complete decoupling DKH transformation written as

$$U_{\text{DKH}} = \begin{pmatrix} U^{LL,\text{DKH}} & U^{LS,\text{DKH}} \\ U^{SL,\text{DKH}} & U^{SS,\text{DKH}} \end{pmatrix}. \quad (60)$$

the matrix forms of U_{DKH}^{LL} and U_{DKH}^{LS} are

$$U_{\text{DKH}}^{LL} = (K^{-1})^\dagger (U^{LL,(m)} U^{LL,\text{fpFW}} + U^{LS,(m)} U^{SL,\text{fpFW}}) K^\dagger, \quad (61)$$

$$U_{\text{DKH}}^{LS} = (K^{-1})^\dagger (U^{LL,(m)} U^{LS,\text{fpFW}} + U^{LS,(m)} U^{SS,\text{fpFW}}) K^\dagger. \quad (62)$$

We should note that the finite-order DKH Hamiltonian matrices are not the result of directly applying the decoupling transformation. To give consistent results, the transformation in Eq. (23) is only used to obtain the off-diagonal blocks while the diagonal blocks are replaced by the traditional finite-order DKH Hamiltonian. However, for high-order calculations, the differences of with/without replacing the diagonal blocks are very small.

VI. RESULTS AND DISCUSSIONS

As we have shown, there exists a systematic hierarchy of several levels for the local relativistic approximation to Hamiltonian matrix elements. In this section, we study the accuracy of the different levels on a test molecule set and introduce the following notation to distinguish the results:

1. **FULL**: Full molecular relativistic transformation, no approximation.
2. **DLU**: Diagonal local approximation to decoupling transformations, U^{LL} and U^{LS} are evaluated for each atomic (diagonal) block. However, three variants are possible:

- i) **DLU(ALL)**: Calculate relativistic transformation for all atoms (all atoms are treated as 'heavy' elements). I.e., apply the 'atomic' unitary transformation to the diagonal and to all off-diagonal blocks of the Hamiltonian
 - ii) **DLU(NB)**: Calculate relativistic transformation for all atoms (all atoms are treated as 'heavy' elements), but apply the relativistic transformation only to blocks of the Hamiltonian assigned to neighboring atoms (and to the diagonal blocks, of course) to achieve linear scaling.
 - iii) **DLU(ABC,BP)**: Distinguish heavy and light atoms, where ABC denote the heavy atoms and apply the BP approximation to the light atoms.
 - iv) **DLU(ABC,NR)**: Distinguish heavy and light atoms, where ABC denote the heavy atoms and apply the NR approximation to the light atoms.
3. **DLH**: Diagonal local approximation to Hamiltonian matrix blocks. The relativistic correction for off-diagonal 'interatomic' blocks are neglected.
- i) **DLH(ALL)** : Calculate relativistic correction to all diagonal 'atomic' blocks.
 - ii) **DLH(ABC,NR/BP)** : Only heavy-atom blocks are considered for the relativistic description.

Note that this classification system does not account for all possible local relativistic approximations: i) A could also represent a group of atoms and ii) additional off-diagonal approximations may also be included.

We have been implemented the local relativistic schemes discussed so far for the X2C, BSS, and DKH exact-decoupling approaches into the MOLCAS programme package [62], into which we have recently implemented X2C, BSS, and the polynomial-cost DKH schemes [52]. In our scalar-relativistic calculations the definition of the W matrix was

$$W_{ij} = \frac{1}{4c^2} \langle \lambda_i | \mathbf{p} \cdot \mathcal{V} \mathbf{p} | \lambda_j \rangle, \quad (63)$$

and λ_i now refers to scalar basis functions instead of 2-spinor basis functions. All other expressions derived in this work are the same for the scalar-relativistic X2C, BSS, and DKH variants.

For our test molecule set, we should rely on closed-shell molecules with heavy elements in the vicinity of other heavy elements and of light elements. Four molecules were selected to test the validity of the different local schemes: I_5^+ , $\text{WH}_6(\text{PH}_3)_3$, $\text{W}(\text{CH}_3)_6$ and Pb_9^{2+} . Moreover, two

reactions



were chosen to study the relative energies. Their structures (see Fig. 1) have been optimized with the TURBOMOLE program package [63] employing the BP86 density functional [64, 65] with a valence triple-zeta basis set with polarization functions on all atoms [66] in combination with Stuttgart ECPs for W, Pb, and I as implemented in TURBOMOLE.

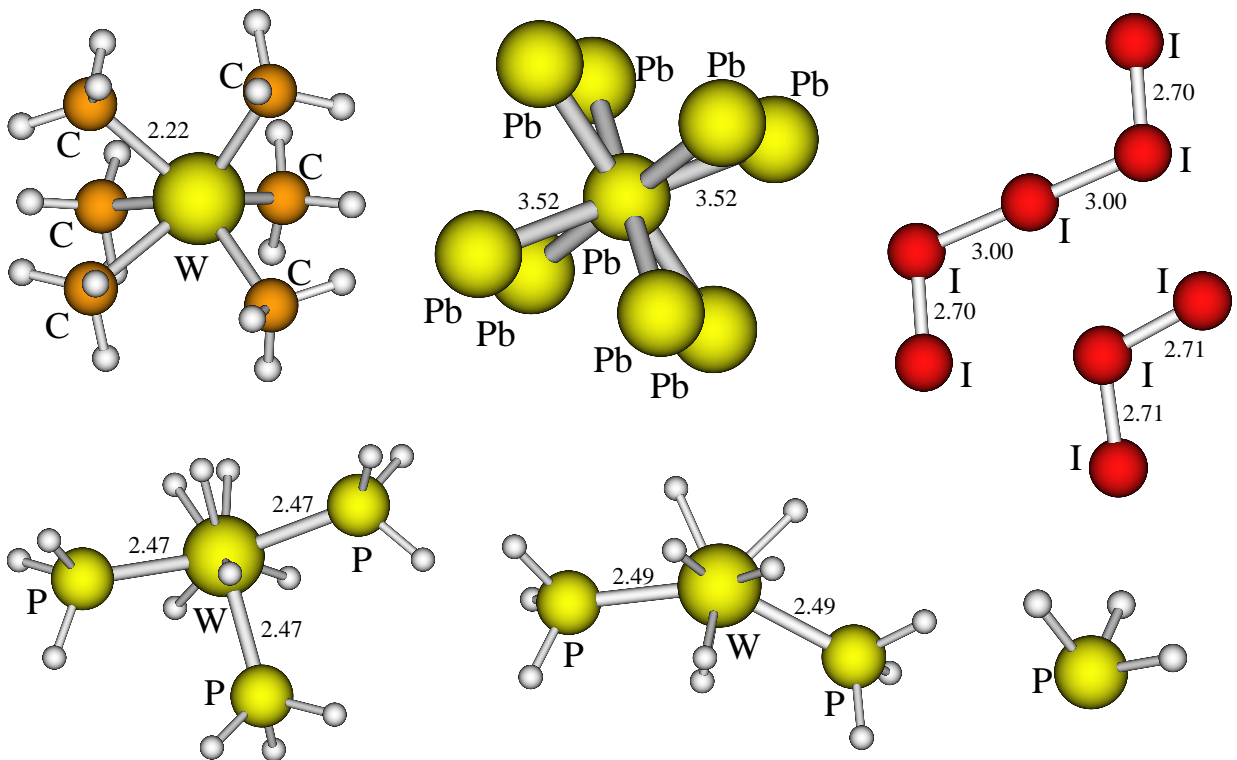


FIG. 1: Structures of the test molecule set (white spheres are hydrogen atoms; selected bond length are given in Å).

All molecules are closed-shell and for our analysis *all-electron* Hartree–Fock calculations were then performed with the MOLCAS programme package. Since our aim is to test the validity of local relativistic approximations instead of obtaining accurate results which can be compared to experiments, we do not need to consider electron-correlation effects in our calculations. The all-electron atomic natural orbital (ANO) basis sets [67–69] were employed at a double-zeta level for all atoms. The exact-decoupling DKH calculations were performed at 35-th order, since

higher order results did not improve the accuracy for our purpose so that this high DKH order can be considered exact. Results of finite-order (standard) DKH2 calculations are also given for the sake of comparison.

We performed DLU(ALL) and DLH(ALL) calculations for all four molecules in order to analyze the general error that is introduced by these local approximations. The differences of total electronic energies with respect to the FULL reference calculation without any local approximation are presented in Table I. Since the errors of total energies are rather small, all values are given in milli-Hartree (mH) atomic units. The errors in total energies can be understood as errors on atomization energies that one would observe when atomization energies are to be calculated with one of the local approaches compared to the FULL one.

TABLE I: Total energy differences introduced by local approximations DLU(ALL) and DLH(ALL) with respect to the FULL reference energy. The FULL total electronic energies are given in Hartree and all deviations are in milli-Hartree atomic units.

Method	Approximation	I_5^+	$WH_6(PH_3)_3$	$W(CH_3)_6$	Pb_9^{2+}
DKH35	FULL	-35562819.0554	-17170991.3602	-16375243.0202	-187905906.7440
	DLU(ALL)	-0.0119	-0.0126	-0.0204	0.0376
	DLH(ALL)	0.2471	-30.8580	-6.3823	2.0376
BSS	FULL	-35562831.5438	-17171044.3875	-16375296.0488	-187906906.7440
	DLU(ALL)	-0.0119	-0.0125	-0.0204	0.0371
	DLH(ALL)	0.2461	-30.8303	-6.3749	2.0065
X2C	FULL	-35561841.0814	-17171053.6272	-16375311.3620	-187911251.8227
	DLU(ALL)	-0.0133	-0.0306	-0.0367	0.0523
	DLH(ALL)	0.2629	-33.3336	-7.0463	2.8362
DKH2	FULL	-35553068.6409	-17158776.3373	-16363031.9017	-187712096.3579
	DLU(ALL)	-0.0083	0.0011	0.0143	0.0240
	DLH(ALL)	0.2897	-33.4347	-6.9555	1.6040

For the I_5^+ molecule, which possesses a pseudo-one-dimensional structure with iodine atoms connected to at most two other iodine atoms, the DLU(ALL) scheme produces an error of about 0.012 mH for both DKH and BSS, while X2C has a slightly larger error of 0.013 mH. Interestingly, it is smallest for DKH2, i.e., only 0.008 mH. The DLH(ALL) local scheme produces

relatively large errors on the total electronic energies, they are 0.24 mH for DKH and BSS, 0.26 mH for X2C, and 0.29 for DKH2. Clearly, DLU(ALL) provides more accurate total energies than DLH(ALL), as it includes the relativistic form of the atom–other-atom off-diagonal blocks, while DLH(ALL) only takes into account the atom–same-atom diagonal blocks. The errors of DLH(ALL) are about 20 times larger than those of DLU(ALL) for the I_5^+ molecule. This indicates the relativistic corrections to off-diagonal blocks are quite important. Apparently, DLU(ALL) provides a very good scheme to take into account the off-diagonal relativistic correction.

The Pb_9^{2+} molecule features a more compact structure where more than two heavy atoms bind to a central lead atom. Moreover, lead atoms are heavier than iodine atoms and relativistic effects are more pronounced. As we can see from Table I, errors of Pb_9^{2+} molecule are all larger than those observed for the I_5^+ molecule. The deviations of DLU(ALL) electronic energies from the FULL reference energy are 0.037 mH for BSS and DKH, while X2C is still slightly larger with a value 0.052 mH (again DKH2 features the smallest deviation of only 0.024 mH). However, deviations of DLH(ALL) results from the reference have become significantly larger, they are 2.03, 2.01, and 2.84 mH for DKH, BSS, and X2C, respectively (but only 1.60 mH for DKH2). Hence, the errors of DLU(ALL) are roughly 60 times smaller than those observed for DLH(ALL).

Considering now molecules that also contain light atoms, the relative accuracy of DLU(ALL) vs. DLH(ALL) changes dramatically. First of all, we note that our DLU(ALL) scheme preserves its accuracy and the deviations from the FULL reference electronic energy are in between those obtained for I_5^+ and Pb_9^{2+} with the same trends as already discussed. The errors of DLH(ALL) energies obtained for the $WH_6(PH_3)_3$ molecule are, however, larger than 30 mH. I.e., they are significantly larger than those observed for Pb_9^{2+} . For $W(CH_3)_6$ molecule, the deviations from the reference are around 6 mH, which are also still larger than Pb_9^{2+} and certainly nonnegligible. Note that the DLH(ALL) errors obtained for $WH_6(PH_3)_3$ and $W(CH_3)_6$ are not all below those of the Pb_9^{2+} molecule although the nuclear charge number of tungsten is smaller than that of lead. This is in contrast to the results obtained for our DLU(ALL) scheme. Since DLH neglects any relativistic correction to off-diagonal 'interatomic' blocks, the large errors must be due to the strong coupling between tungsten and its surrounding lighter atoms (phosphorus and carbon). Apparently, if a heavy-element-containing molecule has several strong bonds to this heavy element, the off-diagonal blocks of neighboring atom pairs can have significant contributions to the total electronic energy. Therefore, the DLH may introduce significant errors for systems including strong bond with its heavy atoms.

Since DLU(ALL) gives a balanced description for all blocks, it will still work in those cases where DLH(ALL) fails. DLU can therefore even be recommended for calculations of molecular systems in different geometries, as for example required for the calculation of potential energy surfaces. DLH would be unfit for this case as it will introduce large errors when the inter-atomic distances become short.

DLU(ALL) also works well for the finite-order (standard) DKH2 method. The errors are even smaller compare to exact-decoupling approaches. However, note that the reference energies of the different relativistic approaches are not the same. Since the DLU(ALL) scheme only discards the off-diagonal blocks of the decoupling transformation compare to the FULL reference scheme, the small errors in the total electronic energies indicate that neglecting off-diagonal terms in the decoupling transformation produces negligible errors for the total energies. This conclusion holds for both exact-decoupling and finite-order DKH approaches. With the DLH(ALL) scheme, however, the errors in the DKH2 calculation that are due to the local DLH approximation are of the same order as those introduced by the finite, i.e., second order when compared to the exact-decoupling approaches. This observation leads to the very important conclusion that none of the exact-decoupling approaches is superior to a finite-order DKH approach if one employs a DLH scheme as this would render the accuracy gained by exact decoupling meaningless! In fact, if DLH is for some reason the local approximation of choice, a standard DKH2 approach would yield a more efficient relativistic scheme and nothing would be gained by achieving exact decoupling.

Besides total energies, we also study the effects of local approximations on molecular properties. We calculated the picture change corrected quadrupole moments. The property matrices are transformed according to Eq. (41). The values for the components of the quadrupole moment depend on the orientation of the molecule. Therefore, we present the spherically averaged quadrupole moments defined as $\sqrt{(Q_{XX}^2 + Q_{YY}^2 + Q_{ZZ}^2)}/3$, in Table II. Where Q_{XX} denotes the XX (diagonal) component of the quadrupole moment. The spherically averaged quadrupole moments are isotropic and thus independent of molecular orientation.

TABLE II: Spherically averaged quadrupole moments for our test set of molecules. All values are given in Hartree atomic units.

Method	Approximation	I_5^+	$\text{WH}_6(\text{PH}_3)_3$	$\text{W}(\text{CH}_3)_6$	Pb_9^{2+}
DKH35	FULL	83.84331	57.61285	57.29508	173.68894
	DLU(ALL)–FULL	-0.00009	0.00018	0.00025	0.00416
	DLH(ALL)–FULL	0.02176	-0.02448	-0.00218	-0.20252
BSS	FULL	83.84329	57.61285	57.29507	173.68873
	DLU(ALL)–FULL	-0.00009	0.00018	0.00025	0.00417
	DLH(ALL)–FULL	0.02177	-0.02449	-0.00218	-0.20231
X2C	FULL	83.84406	57.61240	57.29499	173.68292
	DLU(ALL)–FULL	-0.00017	0.00029	0.00027	0.00590
	DLH(ALL)–FULL	0.02199	-0.02483	-0.00197	-0.21653
DKH2	FULL	83.85310	57.61393	57.29674	173.80677
	DLU(ALL)–FULL	0.00009	-0.00012	0.00018	-0.00003
	DLH(ALL)–FULL	0.02196	-0.02458	-0.00186	-0.20760

The relative errors of DLU(ALL) are all below 10^{-5} a.u., this upper limit can be further reduced to 10^{-6} a.u. if we do not include Pb_9^{2+} . Hence, DLU(ALL) is not only a good approximation for total electronic energies but also for molecular properties like quadrupole moments. The errors introduced by DLU(ALL) are all negligible, while DLH(ALL) shows large errors for quadrupole moments. While for total electronic energies, the largest error of the DLH(ALL) scheme shows up in $\text{WH}_6(\text{PH}_3)_3$ and the smallest in I_5^+ , it is, for the quadrupole moments, largest for Pb_9^{2+} and smallest for $\text{W}(\text{CH}_3)_6$. This different behavior of DLH(ALL) indicates again the important contribution of off-diagonal 'interatomic' blocks, whose importance must be different for electronic energies and for properties. In our DLU(ALL) scheme, I_5^+ has the smallest error and Pb_9^{2+} the largest one for both, energies and properties. Therefore, DLU(ALL) turns out to give a consistent description of relativistic effects, while DLH(ALL) yields an unbalanced relativistic treatment. We should note that we have investigated also other properties (radial moments, electric field gradients, contact densities) with similar conclusions, which is the reason, why we do not report those data here.

For both, energies and quadrupole moments, in the DLU(ALL) scheme, DKH and BSS yields almost the same deviations from the reference and X2C always has larger ones than DKH and BSS. This similarity of DKH and BSS can be well understood since both of them employ the free-particle FW transformation as well as the subsequent exact-decoupling transformation. The close results of DKH and BSS were also observed in Ref. [52]. The slightly larger errors of X2C indicate that the off-diagonal blocks of the X2C exact-decoupling transformation have slightly larger contributions compared to DKH and BSS. However, such differences of errors are too small to be considered a serious drawback for the X2C approach; they are smaller than the differences of the total values. We may use $\text{WH}_6(\text{PH}_3)_3$ as an example. The difference of total isotropic quadrupole moments (57.61285–57.61240) is 0.45 mH, while the difference of errors by DLU(ALL) (0.00029–0.00018) is 0.11 mH. The errors introduced by the DLU(ALL) approximation are smaller than the differences between exact-decoupling approaches. This is also the case for the differences between DKH2 and exact-decoupling approaches. As we can see, the difference between DKH2 and BSS of total isotropic quadrupole moments (57.61393-57.61285) is 1.08 mH, which are much larger than the DLU(ALL) errors.

For chemical purposes, comparisons of total electronic energies are meaningless, because chemical reaction kinetics and thermodynamics are governed by energy differences. We therefore calculated the electronic reaction energies of the two reactions mentioned above. The results are given in Table III and Table IV. The errors of relative energies are less than that of absolute total energies. For example with the DKH approach, the error of I_5^+ is reduced from 0.012 mH to 0.002 mH for DLU(ALL), 0.24 mH to 0.12 mH for DLH(ALL). As the $\text{WH}_6(\text{PH}_3)_3$ molecule, the error is decreased from 0.0126 mH to 0.0006 mH for DLU(ALL), 30.9 mH to 19.4 mH for DLH(ALL). It also holds for other relativistic approaches.

TABLE III: Electronic Hartree–Fock reaction energies for the reaction $\text{I}_5^+ \longrightarrow \text{I}_3^+ + \text{I}_2$. All values are in milli-Hartree atomic units (note that 1 mH is equivalent to about 2.6 kJ/mol).

	DKH35	BSS	X2C	DKH2
FULL	13.9797	13.9798	13.9739	13.9550
DLU(ALL)–FULL	0.0022	0.0022	0.0026	0.0011
DLU(NB)–FULL	0.0040	0.0040	0.0044	0.0027
DLH(ALL)–FULL	-0.1190	-0.1190	-0.1214	-0.1231

Next, we study the neighboring-atomic-block (NB) approximation at the example of the I_3^+ molecule. As shown in Figure 1, it is composed of a chain of iodine atoms and thus suitable for applying the neighboring approximation. Only the pairs of atoms which are connected with a bond displayed in Figure 1 are considered as neighbors. The neighboring groups could also be determined by introducing a cut-off distance parameter. Then, the pair of atoms which has shorter distance than the given parameter is counted as a neighboring pair. The relativistic transformations are now only applied to such neighboring pairs according to Eq. (22) within the DLU scheme. As we can see from Table III, the neighboring approximation DLU(NB) gives very small errors for relative energies. Therefore it is a very good (additional) approximation within our DLU scheme. Hence, with DLU(NB) the computational cost will become linear scaling, while the DLU(ALL) scheme has an order- N^2 scaling, which may be a bottleneck for calculations on very large molecules. But be aware that the success of NB approximation depends on how one defines the neighboring atoms. If the cut-off distance is too small, the DLU scheme will be reduced to the DLH scheme since no neighbors could be found.

Reduction in terms of computational costs can also be achieved by employing the BP or NR approximation to the light atoms. It should be a good approximation if a molecule has many light atoms such as hydrogen. We study these approximations for the phosphine ligand dissociation energy of $WH_6(PH_3)_3$ molecule. The results are presented in Table IV, where DLU(WP,NR) denotes the NR approximation applied to the H atoms while W and P atoms are relativistically described. By contrast, DLU(W,NR) denotes application of the relativistic transformation only to the W atom, while for P and H apply the NR approximation. From Table IV we can see that the results of the BP approximation are very bad, especially if the BP approximation was employed for the phosphorus atoms. The BP approximation to P and H atoms gives errors which are roughly seven times the value for the reactions energy. Hence, this is approximation turns out to be completely useless. Such huge errors stem from the variational collapse of the BP approximation.

TABLE IV: Relative energy difference of the reaction $\text{WH}_6(\text{PH}_3)_3 \longrightarrow \text{WH}_6(\text{PH}_3)_2 + \text{PH}_3$. All values are in milli-Hartree atomic units.

	DKH35	BSS	X2C	DKH2
FULL	35.8447	35.8448	35.8439	35.9064
DLU(ALL)–FULL	-0.0006	-0.0007	0.0040	0.0101
DLU(WP,NR)–FULL	-0.1579	-0.1579	-0.1533	-0.1476
DLU(W,NR)–FULL	0.3648	0.3648	0.3684	0.3737
DLU(WP,BP)–FULL	1.2501	1.2500	1.2547	1.2608
DLU(W,BP)–FULL	254.4914	254.4914	254.4951	254.5018
DLH(ALL)–FULL	19.4202	19.4024	21.0255	21.1635
DLH(WP,NR)–FULL	19.4229	19.4051	21.0284	21.1663
DLH(W,NR)–FULL	19.6722	19.6544	21.2789	21.4146
DLH(WP,BP)–FULL	20.3668	20.3490	21.9743	22.1116
DLH(W,BP)–FULL	273.0474	273.0296	274.6570	274.7922

However, the NR approximation is much better. The errors on the reaction energy within the DLU(NR) scheme, which are in the range of 0.1 to 0.3 mH and thus less than 1 kJ/mol, are perfectly acceptable. This success is due to the correct NR limit of the NR approximation within the DLU scheme. Not unexpectedly, the DLH(NR) results have quite larger errors, because DLH(ALL) already does not provide good results for the phosphine dissociation energy from the $\text{WH}_6(\text{PH}_3)_3$ complex. If we study the difference between DLH(NR) and DLH(ALL), we can find that the NR approximation to actually give small errors on top of DLH(ALL). However, the DLH scheme is not recommended since its accuracy is not guaranteed at all. Only in those cases for which the DLH scheme may work, we can further add the NR approximation on top of it for light atoms. As can be seen from both Table III and Table IV, different relativistic approaches behave basically identical with respect to the discussion of the NB and NR approximations. Therefore, the approximations for the reduction of computational cost discussed in this article can be applied to all relativistic transformation approaches.

VII. CONCLUSIONS

In this work, we aimed at a rigorous local approximation to relativistic transformation schemes, which makes them applicable to and efficient for calculations on large molecules. We developed a systematic hierarchy of approximations that is based on the assembly of the unitary transformation from 'atomic' contributions (DLU) rather than on a local approximation directly applied to the matrix representation of the Hamiltonian (DLH).

The straightforward DLH scheme, which has evolved in the development of the DKH2 method (see references given above), turns out to be not a good local approximation since it only covers the relativistic treatment of the atom–same-atom diagonal blocks in the Hamiltonian. It may fail if inter-atomic distances become short so that a relativistic treatment of the off-diagonal 'interatomic' blocks becomes important. As a consequence, the DLH approximation is not suitable, for example, in studies that crucially depend on accurate electronic energy surfaces. Even if the molecular geometry is fixed, the DLH approach does not provide a balanced description for different molecular properties, since the contributions of the off-diagonal blocks to total observables are quite different for different properties.

By contrast, the DLU scheme proposed by us in this paper overcomes the drawbacks of the DLH scheme. It is an excellent approach to take into account the atom–other-atom (i.e., 'interatomic') off-diagonal relativistic transformations. It does not show an instability for varying molecular structures (i.e, for electronic energy surfaces) and properties. The size of the error introduced by DLU is much smaller than that of DLH; it is typically only 1% of the latter. The errors of the DLU approach when compared to the full relativistic transformation without any local approximation turn out to be very small. They are even smaller than the difference of among the different exact-decoupling approaches, which are already very tiny.

The DLU scheme is valid for all exact-decoupling approaches, while the local X -matrix scheme only works for the X2C approach. Furthermore, the DLU scheme can also be applied to finite-order DKH approaches such as DKH2, which has been very successful in computational chemistry and whose computational costs are less than those of any exact-decoupling approach.

If one has to use the DLH scheme for some reason, such as the lack of a DLU implementation or because of its linear-scaling behavior, the selection of a relativistic approaches is not decisive as the errors introduced by the DLH scheme are already higher than the difference of different relativistic approaches.

However, the linear-scaling behavior of DLH is no good reason since the DLU scheme can

also be linear-scaling within the neighboring-atomic-blocks approximation, which produces negligible errors. One can further reduce the computational costs by employing the BP or NR approximation to all light atoms. However, the BP approximation turns out to be not suitable for variational calculations since it introduces large errors. Only the NR approximation yields acceptable results. The errors of the NR approximation will be larger if the charge of the atom for which the NR approximation is invoked becomes larger. Therefore, the application of the NR approximation will depend on the balance of accuracy and computational cost in an actual calculation.

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