

1-{2-[(取代苯基)甲氧基]-2-(取代苯基)乙基}-1H-三唑和 苯并三唑类化合物的合成及抗真菌活性*

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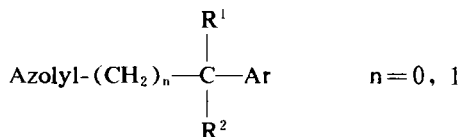
摘要 报道了 39 个新的 1-{2-[(取代苯基)甲氧基]-2-(取代苯基)乙基}-1H-三唑和苯并三唑类化合物的合成与体外抑菌试验, 结果表明, 化合物 6, 13 的抗真菌活性为益康唑的 4~10 倍, 化合物 10, 12, 14, 15, 20, 22, 23, 28, 30 和 31 等对大部分真菌活性也优于或相当于益康唑及克霉唑, 化合物 23 和 31 抗裴氏着色菌活性为益康唑、克霉唑的 30 倍以上, 化合物 22 抗白念珠菌活性为益康唑的 8 倍, 克霉唑的 4 倍。

关键词 三唑类; 苯并三唑类; 抗真菌活性

由于临床上广谱抗菌素、皮质激素和抗肿瘤药物的广泛使用, 导致真菌感染率大为提高, 特别是深部真菌感染已成为癌症病人、爱滋病人及免疫丧失病人死亡的因素之一。临床迫切需要高效低毒、广谱、选择性好的抗真菌尤其抗深部真菌的药物。

自 60 年代末, 克霉唑 (clotrimazole)^[1]、咪康唑 (miconazole)^[2] 和益康唑 (econazole)^[3] 等相继出现后, 氮唑类抗真菌药物的发展最为引人注目。近几年, 由于三唑类化合物对深部真菌作用较好被日益重视, 相继开发了如氟康唑 (fluconazole)^[4]、伊曲康唑 (itraconazole)^[5] 和 terconazole^[6] 等抗深部真菌药物, 有的已在多个国家上市。

构效关系研究表明, 绝大多数有抗真菌活性的氮唑类化合物有如下基本结构:



分子中咪唑环或三唑环是必需的, 而且 1 位氮原子必须与分子中央的碳原子相连接, 改变其它基团则影响理化性质, 从而影响其抗真菌活性。

根据氮唑类抗真菌药物的构效关系及作用机理^[7], 我们以益康唑为先导化合物, 用三唑或苯并三唑(可看作三唑取代物)替换咪唑基团, 同时变换分子中芳环上取代基, 设计合成了 39 个 1-{2-[(取代苯基)甲氧基]-2-(取代苯基)乙基}-1H-三唑和苯并三唑类化合物(见图 1 及表 1)。用二倍稀释法测定了它们对 13 种常见真菌的体外抑菌活性, 并与益康唑、克霉唑对照(见表 2)。

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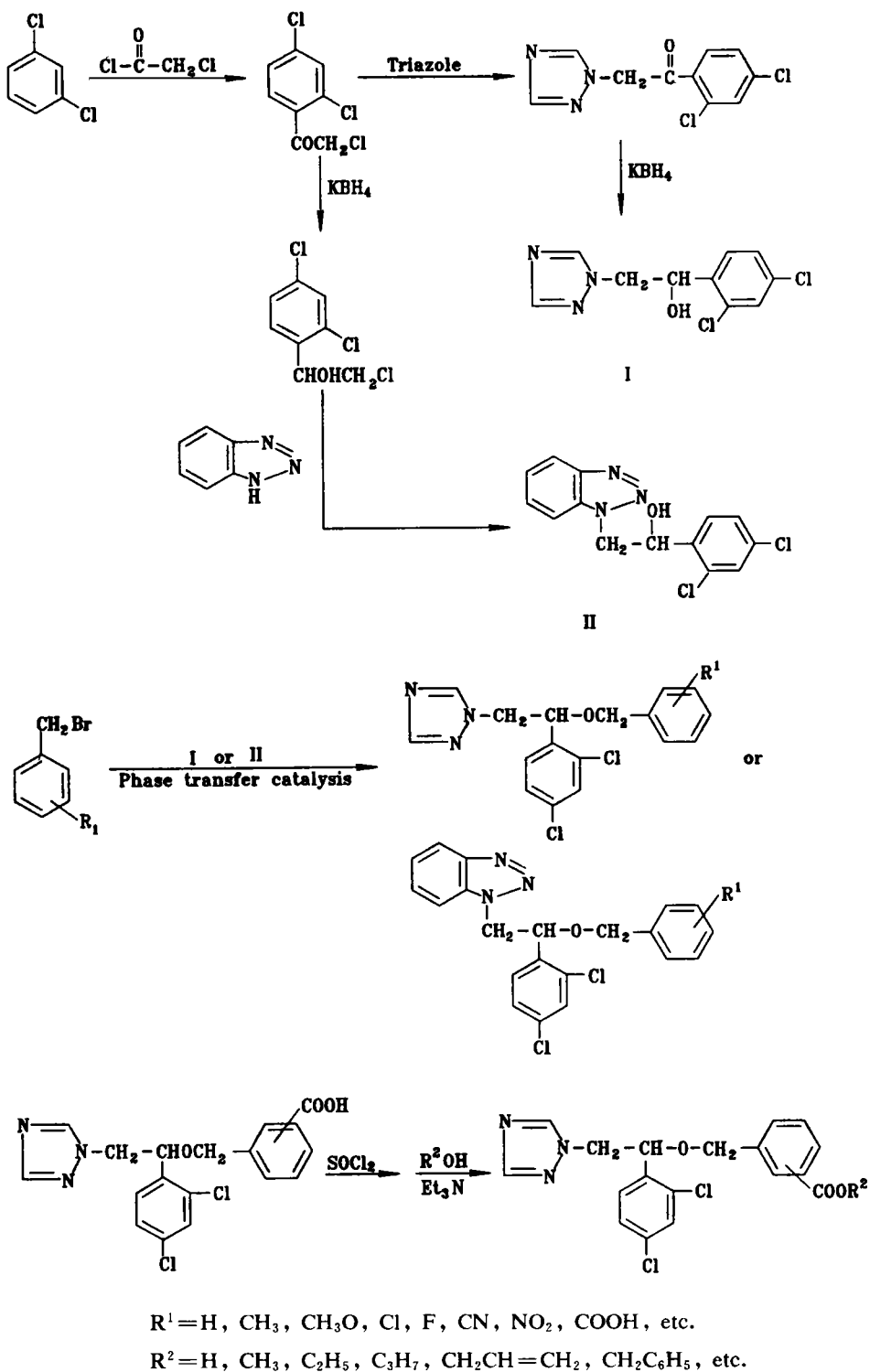
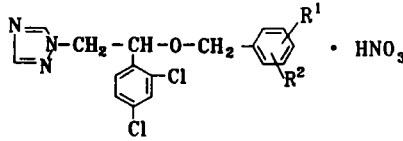







Fig 1 Route of synthesis of the title compounds.

Tab 1 Structure and physical properties of the compounds





Compd ^a	R ¹	R ²	MP C	Yield %	IR(KBr) cm ⁻¹	¹ HNMR(δppm, DMSO-d ₆)
1	H	H	159~160	50.24	860,830,790,752	7.30(m, 5H, Ph-H)
2	<i>p</i> -CH ₃	H	174~175	53.46	870,858,830,815	2.30(s, 3H, Ph-CH ₃)
3	<i>p</i> -OCH ₃	H	162~163	13.61	1210,850,824,810	
4	b		133~135	63.24	1040,860,820	
5	c		160~161	56.81	875,860,815,790	
6	d		128~129	56.81	3120,925,835,640	8.96(s, 1H, ) , 8.33(s, 1H, ) , 7.66 [d, 1H, J = 2.12 Hz, ], 7.49 [dd, 1H, J ₁ = 2.1 Hz, J ₂ = 11.8 Hz, ], 7.40 [d, 1H, J = 11.8 Hz, ] 5.64(m, 1H, Ph-CH-CH ₂ -), O 5.17(m, 2H, =CH ₂), 5.05(m, 1H, -CH=), 4.17(m, 2H, Ph-CH-CH ₂ -N), O 3.72(m, 2H, O-CH ₂ -)
7	<i>o</i> -Cl	H	159~161	53.81	860,825,635	7.70~7.25(m, 7H, Ph-H), 5.22 (dd, J ₁ = 3.66 Hz, J ₂ = 8.14 Hz, 1H, Ar-CH-CH ₂) O
8	<i>m</i> -Cl	H	149~150	58.30	865,730,635	7.75~7.02(m, 7H, Ph-H), 5.14 (dd, J ₁ = 3.58 Hz, J ₂ = 8.24 Hz, 1H, Ar-CH-CH ₂) O
9	<i>p</i> -Cl	H	165~166	40.27	845,823,828	
10	2-Cl	4-Cl	168~170	52.25	860,840,828	
11	<i>o</i> -F	H	164~166	52.84	870,828,780,763	
12	<i>m</i> -F	H	158~160	78.80	860,740,630	7.68~6.86(m, 7H, Ph-H), 4.57 (m, 2H, -CH ₂ -N)
13	<i>p</i> -F	H	160~162	69.93	845,830,635	7.67~7.12(m, 7H, Ph-H), 5.13 (dd, J ₁ = 3.64 Hz, J ₂ = 8.14 Hz, 1H, Ph-CH-CH ₂ N), O 4.55(m, 2H, -CH-CH ₂ -N), O 4.38(m, 2H, O-CH ₂ -C ₆ H ₄ F)

Continued

Compd ^a	R ¹	R ²	MP	Yield	IR (KBr)	¹ HNMR (δ ppm, DMSO-d ₆)
			C	%	cm ⁻¹	
14	<i>o</i> -NO ₃	H	131~133	43.86	1525, 1380, 860, 720	5.27 (dd, J ₁ = 3.64 Hz, J ₂ = 7.48 Hz, 1H, Ph- $\underset{\text{O}}{\text{C}}\text{H}-\text{CH}_2$)
15	<i>p</i> -NO ₃	H	168~169	51.40	890, 860, 830	
16	<i>p</i> -CN	H	173~174	74.39	2160, 870, 828, 780	
17	H	<i>o</i> -COOC ₂ H ₅	150~153	32.26	1715, 865, 830	
18 ^c	H	<i>m</i> -COOH	163~166	68.00	2520, 1950, 1690, 740	
19	H	<i>m</i> -COOCH ₃	145~147	57.69	1720, 820, 755	8.00~7.35 (m, 7H, Ph- $\underline{\text{H}}$), 3.84 (s, 3H, COOCH_3)
20	H	<i>m</i> -COOC ₂ H ₅	151~152	32.52	1720, 825, 750, 635	4.29 (q, J = 9.1 Hz, 2H, $\text{COOCH}_2-\text{CH}_3$), 1.31 (t, J = 9.1 Hz, 3H, $\text{COOCH}_2-\text{CH}_3$)
21	H	<i>m</i> -COOC ₃ H ₇	139~140	72.73	1720, 750, 735, 640	4.20 (t, J = 6.57 Hz, 2H, $\text{COOCH}_2-\text{CH}_2\text{CH}_3$), 1.73 (m, J = 7.53 Hz, 2H, $\text{COOCH}_2-\underline{\text{CH}_2}-\text{CH}_3$), 0.97 (t, J = 7.53 Hz, 3H, $\text{COOCH}_2\text{CH}_2-\text{CH}_3$)
22	H	<i>m</i> -COOCH ₂ CH=CH ₂	132~134	55.56	3110, 1710, 920, 750	6.07 (m, 1H, $-\underline{\text{C}}\text{H}=\text{CH}_2$), 5.37 (m, 2H, $-\text{COOCH}_2\text{CH}=\underline{\text{C}}\text{H}_2$), 4.81 (m, 2H, $\underline{\text{C}}\text{H}_2-\text{CH}=\text{CH}_2$)
23	H	<i>m</i> -COOCH(CH ₃) ₂	150~152	46.05	1715, 730, 640	5.10 [m, J = 6.25 Hz, 1H, $\text{COOCH}(\text{CH}_3)_2$], 1.31 [d, J = 6.25 Hz, 6H, $\text{COOCH}(\underline{\text{C}}\text{H}_3)_2$]
24	H	<i>m</i> -COOCH ₂ CH(CH ₃) ₂	138~139	76.92	1720, 1220, 740, 635	4.06 (d, J = 6.44 Hz, 2H, $\text{COOCH}_2-\text{CH}-$), 2.03 (m, J ₁ = 12.3 Hz, J ₂ = 6.44 Hz, 1H, $-\text{CH}_2-\underline{\text{C}}\text{H}-$), 0.98 [d, J = 12.3 Hz, 6H, $\text{COOCH}_2\text{CH}(\underline{\text{C}}\text{H}_3)_2$]
25	H	<i>m</i> -COOCH ₂ C ₆ H ₅	120~122	60.24	1725, 745, 740	5.35 (s, 2H, $-\text{OCH}_2-\text{Ph}$)
26 ^c	H	<i>p</i> -COOH	187~189	71.43	2500, 1940, 1690	12.9 (s, 1H, $-\text{COOH}$), 7.87~7.19 (m, 7H, Ph- $\underline{\text{H}}$), 5.17 (dd, J ₁ = 3.7 Hz, J ₂ = 8.0 Hz, 1H, Ar- $\underset{\text{O}}{\text{C}}\text{H}-\text{CH}_2-$), 4.54 (m, 2H, $-\text{C}\text{H}-\underline{\text{C}}\text{H}_2-\text{N}$), 4.38 (m, 2H, $\text{O}-\underline{\text{C}}\text{H}_2-\text{Ar}$)
27	H	<i>p</i> -COOCH ₃	145~147	55.60	1720, 830, 640	5.16 (dd, J ₁ = 3.81 Hz, J ₂ = 8.22 Hz, 1H, $-\underline{\text{C}}\text{H}-\text{CH}_2$)
28	H	<i>p</i> -COOC ₂ H ₅	140~142	41.41	1725, 875, 830	1.35 (t, J = 7.0 Hz, 3H, $-\text{OCH}_2\text{CH}_3$)
29	H	<i>p</i> -COOCH ₂ CH ₂ CH ₃	139~141	78.95	1720, 836, 645	4.20 (t, J = 6.54 Hz, 2H, $-\text{O}-\underline{\text{C}}\text{H}_2-\text{CH}_2-$), 1.69 (m, J ₁ = 6.54 Hz, J ₂ = 7.4 Hz, 2H, $-\text{CH}_2-\underline{\text{C}}\text{H}_2-\text{CH}_3$), 0.95 (t, J = 7.44 Hz, 3H, $-\text{CH}_2-\text{CH}_2-\underline{\text{C}}\text{H}_3$)

Continued

Compd ^a	R ¹	R ²	MP	Yield	IR (KBr)	¹ HNMR (δ ppm, DMSO-d ₆)
			C	%	cm ⁻¹	
30	H	<i>p</i> -COOCH ₂ CH=CH ₂	140~142	73.68	3130, 1720, 925	8.80(s, 1H,  , 8.19(s, 1H,  , 7.91~7.24(m, 7H, Ph-H), 5.18(dd, J ₁ =3.37 Hz, J ₂ =8.07 Hz, 1H, Ar-C(<u>H</u>)-CH ₂ -), 6.07(m, 1H, COOCH ₂ CH=CH ₂), 5.36(m, 2H, COOCH ₂ CH=CH ₂), 4.81(m, 2H, COOCH ₂ -CH=CH ₂)
31	H	<i>p</i> -COOCH(CH ₃) ₂	154~155	52.61	1718, 1380, 835	7.82~7.18(m, 7H, Ph-H)
32	H	<i>p</i> -COOCH ₂ CH(CH ₃) ₂	142~144	71.43	1725, 830, 650	4.39(m, 2H, O-CH ₂ -Ph)
33	H	<i>p</i> -COOCH ₂ C ₆ H ₅	125~127	48.19	1720, 835, 755	8.0~7.2(m, 12H, Ph-H), 5.35(s, 2H, COOCH ₂ -Ph)
34	H	<i>p</i> -SO ₃	258~260	65.42	2500, 1350, 1160	4.50(s, bra, 1H, -SO ₃ H)
35 ^f	H	H	184~186	16.11	860, 830, 740	
36 ^f	2-Cl	4-Cl	110~111	33.1	860, 840, 830	
37 ^f	<i>p</i> -CN	H	168~171	21.74	2230, 815, 730	
38 ^f	H	g	145~147	44.21	3140, 1715, 820	
39 ^f	H	<i>p</i> -COOH	185~188	38.30	2610, 1940, 1715	12.1(s, 1H, -COOH), 8.02~7.37(m, 11H, Ph-H)

a. C, H, N analyses were within 0.5% of calculated values. b. Compd 4: Phenyl group replaced by 1,2,3,4-tetrahydronaphthyl group. c. Compd 5: Phenyl group replaced by 1-naphthyl group. d. Compd 6: phenyl group replaced by allyl group. e. Compds 18, 26, 34, 36, 29: Free bases. f. Compds 35~39: Triazolyl group replaced by benzotriazolyl group. g. Compd 38: Phenyl group replaced by allyl group.

对1H-1,2,4-三唑烷基化文献报道在回流的乙酸乙酯中相转移催化反应^[8],后处理困难,收率低。现改为在冰水浴下二氯甲烷中反应,适当延长反应时间,收率明显提高。

Williamson 反应制备目标化合物中,使用氢化钠,要求无水,条件苛刻,我们改为相转移催化醚化,反应条件温和,反应后在溶剂中直接加硝酸成盐,重结晶纯化,避开了层析,简化了操作。

体外抑菌试验初步表明,大多数目标化合物对所测的多种真菌均有不同程度的抗菌活性。化合物6,13对所测菌种活性最高,为先导化合物益康唑的4~10倍,化合物10,12,14,15,20,22,23,28,30和31等对大部分真菌的活性也优于或相当于益康唑及克霉唑。引人注目的是部分化合物对深部真菌有较好的活性,化合物14除了新型隐球菌和熏烟色曲菌外,对其它真菌均优于克霉唑,化合物23和31抗裴氏着色菌活性为益康唑及克霉唑的30倍以上,化合物22抗白念珠菌活性为益康唑的8倍,克霉唑的4倍。

Tab 2 Antifungal activities of the compounds

Compd	MIC* ($\mu\text{g}\cdot\text{ml}^{-1}$)													
	<i>Tricho- phyton</i>	<i>T. diolaceum</i>	<i>T. rubrum</i>	<i>Micro- sporum gypseum</i>	<i>M. caniso</i>	<i>M. ferru- gucum</i>	<i>Epider- mophyton floccosum</i>	<i>Spero- trichum schenckii</i>	<i>Clado- sporium carrionii</i>	<i>Candida albicans</i>	<i>Crypto- coccus ucoformans</i>	<i>Fonseca pedrusii</i>	<i>Aspergillus fumigatus</i>	
1	5	1.25	0.31	1.25	2.50	0.63	10	>40	>40	>40	>40	>40	>40	
2	10	2.50	0.16	5	5	2.50	10	20	5	>40	>40	2.50	>40	
3	10	2.50	2.50	>40	10	10	20	>40	10	>40	40	40	10	
4	10	5	5	10	>10	5	10	10	20	40	>40	40	>10	
5	10	10	1.25	10	10	20	10	>10	10	>40	>40	>40	>10	
6	0.08	0.16	0.08	1.25	0.31	0.63	0.08	10	1.25	40	20	10	10	
7	10	2.50	0.31	10	2.50	0.15	2.50	10	5	>40	>40	2.50	>40	
8	20	5	1.25	10	5	1.25	5	10	5	>40	>40	5	>40	
9	20	0.63	0.31	20	10	0.16	1.25	20	10	20	>40	20	>40	
10	20	0.16	2.50	1.25	5	0.16	0.31	20	20	20	>40	40	>40	
11	1.25	1.25	0.16	5	5	1.25	2.50	>40	1.25	>40	>40	>40	>40	
12	2.50	0.31	0.16	5	2.50	5	0.63	20	10	>40	>40	2.50	>40	
13	0.63	0.31	0.08	5	5	1.25	0.31	1.25	0.63	20	>40	5	20	
14	2.50	2.50	2.50	20	20	10	5	5	5	10	5	5	>40	
15	2.50	0.63	0.08	5	1.25	0.63	0.31	20	2.50	40	40	5	20	
16	5	2.50	0.16	10	5	2.50	1.25	10	2.50	10	>40	5	>40	
17	10	2.50	5	10	10	10	20	20	5	>40	>40	20	>40	
18	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	>10	>40	>40	
19	10	2.50	1.25	10	40	10	2.50	10	1.25	40	>40	10	40	
20	10	0.31	0.16	2.50	20	20	5	10	2.50	40	40	10	40	
21	20	0.08	1.25	40	40	0.31	5	40	2.50	40	>40	20	>40	
22	20	10	5	20	20	2.50	5	5	2.50	5	40	10	>40	
23	20	0.63	1.25	20	5	5	0.63	10	40	10	>40	1.25	>40	
24	40	5	20	40	40	20	20	10	40	40	40	20	>40	
25	40	40	20	40	>40	>40	40	>40	40	>40	>40	40	>40	
26	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	>10	>40	>40	
27	10	1.25	0.63	10	10	2.50	20	40	10	40	40	10	>40	
28	10	0.63	0.16	1.25	10	20	5	20	2.50	>40	>40	10	>40	
29	10	0.63	0.16	5	5	1.25	20	40	5	>40	>40	5	>40	
30	5	20	5	20	10	1.25	20	5	2.50	20	20	10	>40	
31	10	0.63	1.25	20	5	0.63	10	40	5	>40	>40	1.25	20	
32	20	10	10	5	20	2.50	40	40	40	>40	>40	10	>40	
33	40	20	40	40	>40	10	>40	20	40	>40	20	10	>40	
34	40	20	>40	40	40	>40	>40	>40	>40	20	40	40	>40	
35	40	2.50	1.25	5	>40	5	>40	40	40	>40	>40	>40	>40	
36	>40	20	40	>40	>40	20	>40	>40	40	>40	>40	>40	>40	
37	40	5	5	>40	40	2.50	>40	40	1.25	40	>40	>40	1.25	
38	40	5	2.50	5	20	1.25	>40	10	>40	>40	>40	>40	>40	
39	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	>40	
**	0.63	0.63	2.50	20	20	0.31	5	25	10	20	0.63	>40	2.50	
***	0.63	0.63	0.31	20	>40	20	20	—	—	40	>40	>40	>40	

* Drugs: All compounds were dissolved in DMSO and diluted with Sabourand's dextrose agar medium; incubation at 37°C for 2~7 d. ** Clotrimazole was imported for Italy. *** Econazole was synthesized in our laboratory.

实 验 部 分

熔点用毛细管法测定,温度未校正;元素分析仪为 MOD-1106型;红外光谱仪为 Hitachi 270-50型, KBr 压片;核磁共振仪为 AC-300P 型, TMS 为内标, DMSO- d_6 为溶剂。

1-(2,4-二氯苯基)-2-(1H-1,2,4-三唑基-1-)-乙醇^[9]

将三唑 13.6 g (0.2 mole) 悬浮于 CH_2Cl_2 70 ml 中,于冰水浴冷却下滴入 α -2,4-三氯苯乙酮 20.1 g (0.09 mole) 和 CH_2Cl_2 30 ml 混合液,滴毕于 0~5 C 搅拌反应 6 h,放置过夜,减压回收溶剂,残物倾入 300 ml 冰水中,过滤,水洗,干燥,溶于 CH_3OH 100 ml,冰浴冷却下分批加入 KBH_4 10 g,加毕搅拌 1 h,加热回流 4 h,回收溶剂,残物倾入冰水中,滤集,水洗,干燥,95%乙醇重结晶,得白色固体 19.6 g,收率 83%, mp 89~90 C (90 C)^[9]。

4-[2-(1H-1,2,4-三唑基-1-)-1-(2,4-二氯苯基)乙氧基]甲基}苯甲酸(26)

将 NaOH 0.5 g 溶于水 2 ml 中,加入 THF 30 ml,正四丁基溴化铵 0.1 g 和 1-(2,4-二氯苯基)-2-(1H-1,2,4-三唑基-1-)-乙醇 1.30 g (5 mmole),冰水浴冷却并搅拌下,滴入对溴甲基苯甲酸乙酯 2.0 g 和 THF 15 ml 混合液,滴毕室温搅拌反应 12 h,再加热回流 7 h 后,加入 10% NaOH 1 ml,再回流搅拌 3 h,蒸出溶剂,向残物中加入水 50 ml,乙酸乙酯提取,水相用醋酸酸化至 pH 4~5,乙酸乙酯提取,水洗,干燥,蒸去溶剂,异丙醇重结晶得白色固体 1.40 g,收率 71%, mp 187~189 C。元素分析 $\text{C}_{18}\text{H}_{15}\text{Cl}_2\text{N}_3\text{O}_3$, 理论值%: C 55.12, H 3.85, N 10.72; 实测值%: C 55.02, H 3.81, N 10.79。

化合物 18 和 39 合成方法同 26。

4-[2-(1H-1,2,4-三唑基-1-)-1-(2,4-二氯苯基)乙氧基]甲基}苯甲酸烯丙基酯(30)

将 26 0.60 g (1.53 mmole) 溶于 CHCl_3 10 ml,滴入 SOCl_2 3 ml,加热回流 1 h,回收溶剂及多余氯化亚砷,加入烯丙醇 10 ml 及吡啶 0.5 ml,室温搅拌过夜,回收烯丙醇,向残留物中加水 50 ml,调 pH 呈碱性,二氯甲烷提取,水洗,干燥,蒸去二氯甲烷,将残物溶于异丙醇,滴入浓硝酸成盐,乙醇重结晶,得 0.56 g,收率 73%, mp 140~142 C。

化合物 17, 19~25 和 27~33 按此法合成。

1-2-[2-(4-氟苯基)甲氧基]-2-(2,4-二氯苯基)乙基]-1H-1,2,4-三唑硝酸盐(13)

取 2-(1H-1,2,4-三唑基-1-)-1-(2,4-二氯苯基)乙醇 1.3 g (5 mmole),加入 50% NaOH 水溶液 1.0 g, CH_2Cl_2 20 ml,四丁基溴化铵 0.1 g,回流搅拌 5 min,于室温滴加含对溴甲基氟苯 1.0 g 的 CH_2Cl_2 液 10 ml,滴毕,室温搅拌 1 h,再加热回流 5 h,反应完毕水洗反应液,加硝酸成盐,冷却,过滤得粗品,用乙醇-丙酮(1:1)重结晶,得无色固体 1.5 g,收率 69%, mp 160~162 C。元素分析 $\text{C}_{17}\text{H}_{14}\text{Cl}_2\text{N}_3\text{O}\cdot\text{HNO}_3$, 理论值%: C 47.57, H 3.52, N 13.06; 实测值%: C 47.49, H 3.61, N 13.06。

化合物 1~16, 35~38 均按此法合成。

4-[2-(1H-1,2,4-三唑基-1-)-1-(2,4-二氯苯基)乙氧基]甲基}苯磺酸(34)

将 NaOH 0.5 g 溶于水 2 ml 中,加入 THF 30 ml,正四丁基溴化铵 0.1 g, 1-(2,4-二氯苯基)-2-(1H-1,2,4-三唑基-1-)-乙醇 1.3 g (5 mmole),冰浴冷却,搅拌下滴入对溴甲基苯磺酸乙酯 2.0 g 和 THF 15 ml 混合液,滴毕室温搅拌反应 60 h,加热回流 5 h,蒸出溶剂,向残物中加入水 50 ml,甲苯提取 3 次,水相用浓硝酸酸化,有片状结晶析出,异丙醇重结晶,得闪亮鳞片状结晶 1.45 g,收率 65%, mp 258~260 C。元素分析 $\text{C}_{17}\text{H}_{15}\text{Cl}_2\text{N}_3\text{O}_4\text{S}$, 理论值%: C 47.67, H 3.53, N 9.81; 实测值%: C 47.44, H 3.48, N 9.84。

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SYNTHESIS AND ANTIFUNGAL ACTIVITIES OF 1-{2-[(SUBSTITUTED-PHENYL)METHOXY]-2-(SUBSTITUTED-PHENYL)ETHYL}-1H-TRIAZOLES AND BENZTRIAZOLES

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ABSTRACT Thirty-nine 1-{2-[(2,4-dichlorophenyl)methoxy]-2-(substituted-phenyl)ethyl}-1H-triazoles and benztriazoles were synthesized and all title compounds are reported for the first time.

Results of preliminary biological tests showed that the most active compounds **6** and **13** are 4~10 times more active than the lead compound econazole. The antifungal activities of compounds **10**, **12**, **14**, **15**, **20**, **22**, **23**, **28**, **30** and **31** are better than or comparable to the activities of econazole and clotrimazole against most fungi. Compounds **23** and **31** are 30 times more active than econazole and clotrimazole against *Fonseaea pedrosoi*. Compound **22** is 4 and 8 times more active than clotrimazole and econazole, respectively, against *Candide albicans*.

Key words Triazoles; Benztriazoles; Antifungal activity