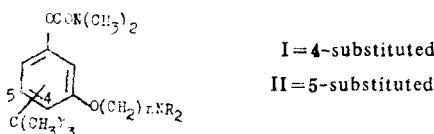


# 可逆性胆碱酯酶抑制剂二甲氨基甲酸-[3-(烷氨基)烷氧基-4(5)-叔丁基]苯酯的合成

王 林 董 永 明

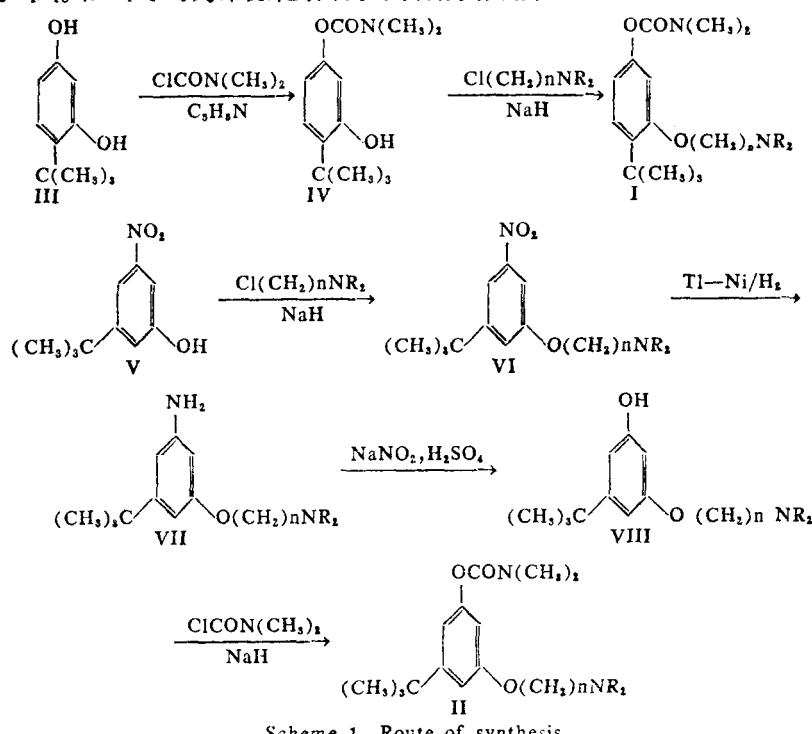
(军事医学科学院毒物药物研究所, 北京)

在催醒安的邻位和对位异构体及其同系物的苯环上引入烷基后, 可以增强抑制胆碱酯酶的活性<sup>(1,2)</sup>, 我们乃进一步在催醒安及其同系物的苯环不同位置上引入叔丁基, 合成了一系列二甲氨基甲酸-[3-(烷氨基)烷氧基-4(5)-叔丁基]苯酯(I<sub>1~13</sub> 和 II<sub>1~5</sub>) (表 2), 以探索对活性的影响。



Structures of the compounds synthesized

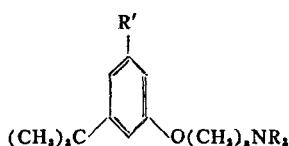
化合物 I<sub>1~13</sub> 和 II<sub>1~5</sub> 均为未知化合物。其合成路线如下:



Scheme 1. Route of synthesis

中间体 IV 参照前报<sup>(1)</sup>方法制备, 所得产物经元素分析和质谱测定, 符合单酯化产品, 但

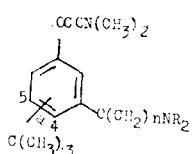
Tab 1. 3-(Alkylamino) alkoxy-5-t-butyl-1-substituted benzenes



Compd	R'	n	NR <sub>2</sub>	BP °C/kPa(mm) and/or MP °C(Solvent)	Yield (%)	Formula <sup>a</sup>	MS m/z(M <sup>+</sup> )
VI <sub>1</sub>	NO <sub>2</sub>	2	N(CH <sub>3</sub> ) <sub>2</sub>	135~140/0.07(0.5) 182~183(EtOH) <sup>b</sup>	73.3	C <sub>14</sub> H <sub>21</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	266 <sup>d</sup>
VI <sub>2</sub>	NO <sub>2</sub>	2		152~155/0.005(0.04)	75.7	C <sub>16</sub> H <sub>24</sub> N <sub>2</sub> O <sub>3</sub>	292
VI <sub>3</sub>	NO <sub>2</sub>	2		178~184/0.07(0.5)	83.3	C <sub>17</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	306
VI <sub>4</sub>	NO <sub>2</sub>	3		129/0.009(0.07)	75.2	C <sub>17</sub> H <sub>26</sub> N <sub>2</sub> O <sub>3</sub>	306
VI <sub>5</sub>	NO <sub>2</sub>	3		148~151/0.0004(0.003)	96.3	C <sub>18</sub> H <sub>28</sub> N <sub>2</sub> O <sub>3</sub>	320
VII <sub>1</sub>	NH <sub>2</sub>	2	N(CH <sub>3</sub> ) <sub>2</sub>	246(EtOH) <sup>b</sup>	84.0 <sup>c</sup>	C <sub>14</sub> H <sub>24</sub> N <sub>2</sub> O·2HCl	236 <sup>d</sup>
VII <sub>2</sub>	NH <sub>2</sub>	2		130~136/0.009(0.07) 66.5~68.5(petr ether 30~60°)	53.2	C <sub>16</sub> H <sub>24</sub> N <sub>2</sub> O	262
VII <sub>3</sub>	NH <sub>2</sub>	2		160~162/0.01(0.08) (90~91, petr ether 30~60°)	62.1	C <sub>17</sub> H <sub>26</sub> N <sub>2</sub> O	276
VII <sub>4</sub>	NH <sub>2</sub>	3		129~132/0.0001(0.001) 236~239(EtOH-Et <sub>2</sub> O) <sup>b</sup>	81.2	C <sub>17</sub> H <sub>26</sub> N <sub>2</sub> O·2HCl	276 <sup>d</sup>
VII <sub>5</sub>	NH <sub>2</sub>	3		165~167/0.007(0.05) 236~238(EtOH) <sup>b</sup>	68.4	C <sub>18</sub> H <sub>28</sub> N <sub>2</sub> O·2HCl·½H <sub>2</sub> O	290 <sup>d</sup>
VIII <sub>1</sub>	OH	2	N(CH <sub>3</sub> ) <sub>2</sub>	116~117, EtOH-petr ether 30~60°	61.4	C <sub>14</sub> H <sub>22</sub> NO <sub>4</sub>	237
VIII <sub>2</sub>	OH	2		104~105, Me <sub>2</sub> CO	65.7	C <sub>16</sub> H <sub>20</sub> NO <sub>2</sub>	263
VIII <sub>3</sub>	OH	2		132~133.5, Et <sub>2</sub> O	72.0	C <sub>17</sub> H <sub>27</sub> NO <sub>2</sub>	277
VIII <sub>4</sub>	OH	3		122~124, Me <sub>2</sub> CO	52.8	C <sub>17</sub> H <sub>27</sub> NO <sub>2</sub>	277
VIII <sub>5</sub>	OH	3		107~108, petr ether 60~90°	48.5	C <sub>18</sub> H <sub>28</sub> NO <sub>2</sub>	291

a. Analytical results for C, H, and N are within  $\pm 0.4\%$  of the calculated values; b. Melting point of hydrochloride salt; c. The yield of compound VII<sub>1</sub> is based on hydrochloride salt, others of VII are based on the corresponding free bases; d. M<sup>+</sup> of free base.

Tab 2. 3-(Alkylamino)alkoxy-4(5)-t-butylphenyl N,N-dimethylcarbamates



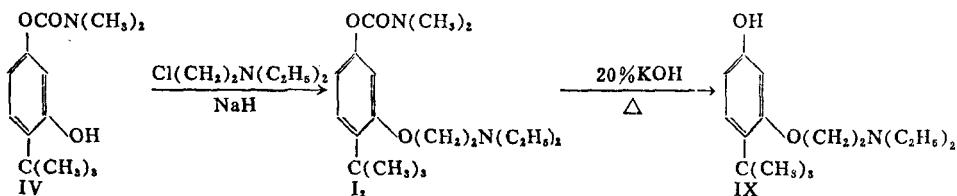
Compd <sup>a</sup>	a	n	NR <sub>2</sub>	BP °C/kPa(mm) or MP°C	Yield (%)	HCl Salt		Formula <sup>c</sup>	Pharmacological results	
						MP°C <sup>b</sup>	Yield (%)		LD <sub>50</sub> mg/kg(ip)	pI <sub>so</sub>
I <sub>1</sub>	4	2	N(CH <sub>3</sub> ) <sub>2</sub>	145~150/0.007(0.05)	68.0	198~200	95.0	C <sub>17</sub> H <sub>28</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	266	<4
I <sub>2</sub>	4	2	N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	142~145/0.0004(0.003)	81.8	185~187	84.0	C <sub>19</sub> H <sub>32</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	<4
I <sub>3</sub>	4	2	N(nC <sub>3</sub> H <sub>7</sub> ) <sub>2</sub>	148~150/0.003(0.02)	73.4	170~171	72.6	C <sub>21</sub> H <sub>38</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	<4
I <sub>4</sub>	4	2	N(iC <sub>3</sub> H <sub>7</sub> ) <sub>2</sub>	134~138/0.005(0.04)	87.7	181~183	56.6	C <sub>21</sub> H <sub>38</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	178	6.19
I <sub>5</sub>	4	2	N(nC <sub>4</sub> H <sub>9</sub> ) <sub>2</sub>			123~124	45.6	C <sub>23</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	300	>9
I <sub>6</sub>	4	2		78~80 <sup>d</sup>	88.7	171~174	57.6	C <sub>19</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	<4
I <sub>7</sub>	4	2		142~143/0.005(0.04)	74.2	171.5~172	87.9	C <sub>20</sub> H <sub>32</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	46.4	4.51
I <sub>8</sub>	4	2	N() <sub>2</sub>			205	68.0	C <sub>27</sub> H <sub>44</sub> N <sub>2</sub> O <sub>3</sub> ·HCl		
I <sub>9</sub>	4	3	N(CH <sub>3</sub> ) <sub>2</sub>	167~168/0.007(0.05)	60.3	161.5~163	84.6	C <sub>18</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	6.7
I <sub>10</sub>	4	3		178~179/0.01(0.08)	71.4	155.5~156.5	74.5	C <sub>20</sub> H <sub>32</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	7.0
I <sub>11</sub>	4	3		162~166/0.008(0.06)	82.4	183~185	79.1	C <sub>21</sub> H <sub>34</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	4
I <sub>12</sub>	4	3	N(C <sub>2</sub> H <sub>5</sub> ) <sub>2</sub>	176~183/0.004(0.03)	83.1	152~153.5	77.2	C <sub>20</sub> H <sub>34</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	147	<4
I <sub>13</sub>	4	3	N(nC <sub>3</sub> H <sub>7</sub> ) <sub>2</sub>	168~170/0.007(0.05)	60.2	119.5~120	65.5	C <sub>22</sub> H <sub>38</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	82.5	4.24
II <sub>1</sub>	5	2	N(CH <sub>3</sub> ) <sub>2</sub>			190~192	70.8	C <sub>17</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub> ·HCl	215	4.9
II <sub>2</sub>	5	2				176~178	66.2	C <sub>19</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	82.5	5.3
II <sub>3</sub>	5	2				179~181	75.3	C <sub>20</sub> H <sub>32</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	56.2	5.4
II <sub>4</sub>	5	3				167~168	50.4	C <sub>20</sub> H <sub>32</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	68.1	5.2
II <sub>5</sub>	5	3				173~175	54.4	C <sub>21</sub> H <sub>34</sub> N <sub>2</sub> O <sub>3</sub> ·HCl	68.1	6.05

3-(2-dimethylamino)ethoxyphenyl N, N-dimethylcarbamate

48 6.82

a. MS, IR and <sup>1</sup>HNMR of all compounds are consistent with the assigned structures. b. Recryst from EtOH-Et<sub>2</sub>O c. Analytical results for C, H and N are within  $\pm 0.4\%$  of the calculated values, except C of II<sub>1</sub> which was found to be 62.76%. d. Recryst from 95% EtOH

核磁共振谱不能确定其结构为中间体 IV。我们乃将所得单酯化合物先与二乙氨基氯乙烷进行醚化生成化合物 I<sub>2</sub>, 再经水解, 所得产物 IX 的熔点为 106.5~107.5°C (60~90°C 石油醚重结晶) 与文献<sup>(3)</sup>定位合成的完全一致, 从而确定了单酯化合物即为中间体 IV。



所有化合物分别经孙长荣和马秀英等进行了初步药理评选。发现引入叔丁基后多数化合物的毒性和抑酶活性均有所下降; 大多数化合物 I 经口服给药预防小鼠有机磷化合物中毒的效价明显增强。

## 实验部分

物理常数均未校正。红外光谱仪用岛津 IR-408 型, KBr 压片。核磁共振仪用 JEOL-JNM-PMX 60 S<sub>1</sub> 型, CDCl<sub>3</sub> 为溶剂, TMS 为内标。质谱仪用 MAT-711型。

### 二甲氨基甲酸-(3-羟基-4-叔丁基)苯酯(IV)

取-4-叔丁基间苯二酚溶于苯中, 在吡啶为去酸剂条件下与二甲胺基甲酰氯反应制备<sup>(1)</sup>。mp 191~193°C, 产率 40.2%。MS m/z 237 (M<sup>+</sup>)。IR cm<sup>-1</sup> 3320, 1690, 1395, 1370。元素分析 C<sub>13</sub>H<sub>19</sub>NO<sub>3</sub>, 计算值 %C 65.80, H 8.07, N 5.90; 实测值 %C 65.91, H 8.16, N 5.88。

### 5-硝基-3-(烷氨基)烷氧基叔丁基苯(VI)

取 5-硝基-3-叔丁基苯酚 (V)<sup>(4,5)</sup> 溶于甲苯中, 加入等摩尔的 80%NaH, 搅拌回流 1.5 h 后, 滴加烷氨基氯代烷(与 V 的摩尔比为 1.3:1), 继续回流 2~5 h。用 10% 盐酸溶液提取, 以浓氨水碱化酸提取液, 乙醚提取, 无水 MgSO<sub>4</sub> 干燥, 减压蒸馏即得中间体 VI。结果见表 1。

### 3-(烷氨基)烷氧基-5-叔丁基苯胺(VII)

取中间体 VI 溶于无水乙醇中, 加入 Ti-Ni 于常压下室温催化氢化至吸氢量接近于理论值, 然后按常法处理即得。结果见表 1。

### 3-(烷氨基)烷氧基-5-叔丁基苯酚(VIII)

参照文献<sup>(5)</sup>, 将中间体 VII 经重氮化水解制备。结果见表 1。

### 二甲氨基甲酸-[3-(烷氨基)烷氧基-4-叔丁基]苯酯(I)

以中间体 IV 为原料, 仿制备中间体 VI 的方法制取。结果见表 2。

### 二甲氨基甲酸-[3-(烷氨基)烷氧基-5-叔丁基]苯酯(II)

取相应的中间体 VIII 溶于甲苯中, 先与等摩尔的 80%NaH 反应 1.5 h, 再与过量的二甲氨基甲酰氯反应 2 h。所得游离碱未经蒸馏, 直接做成盐酸盐。结果见表 2。

**关键词** 胆碱酯酶抑制剂; 二甲氨基甲酸-[3-(烷氨基)烷氧基-4-叔丁基]苯酯; 二甲氨基甲酸-[3-(烷氨基)烷氧基-5-叔丁基]苯酯

**致谢** 光谱和元素分析由本所金素琴、王好山及本院仪器中心王小琴、金人慈、缪振春、魏同太、贾进山和武力民等同志测定; 氢化实验由苏俊和李荣立两位同志帮助完成。

## 参 考 文 献

- 董永明, 等. 可逆性胆碱酯酶抑制剂:二甲氨基甲酸-[3-烷基-4-(烷氨基)烷氧基]苯酯的合成. 药学学报 1984; 19: 538.
- 姚霞君、董永明. 可逆性胆碱酯酶抑制剂:二甲氨基甲酸-[2-(2-二甲氨基)乙氧基-4(或5)-特丁基]苯酯的合成. 药学学报 1984; 19: 622.
- Stephens FF and Sharpe CJ. Ortho-Tertiary butyl phenoethers. *Brit* 922, 600; *CA* 1963; 59:9880 f.
- Hoefnagel AJ, et al. The nitration of 2-nitro-1, 4-di-t-butylbenzene. Synthesis of the three dinitro-1,4-di-t-butyl benzene and some related compounds. *Rec Trav Chim* 1969; 88:386.
- Musso H and Bormann D. Über orceinfarbstoffe. XXII. Die autoxydation des 5-tert-butylresorcin. *Chem Ber* 1965; 98:2774.

## SYNTHESIS OF 3-(ALKYLAMINO) ALKOXY-4(5)-t-BUTYLPHENYL N, N-DIMETHYLCARBAMATES AS REVERSIBLE CHOLINESTERASE INHIBITORS

L Wang and YM Dong

(Institute of Pharmacology and Toxicology, Academy of Military Medical Sciences, Beijing)

**ABSTRACT** A series of 3-(dialkylamino) alkoxy-4(5)-t-butylphenyl N, N-dimethylcarbamates(I and II), which are the nuclear alkylated derivatives of 3-(2-dimethylamino) ethoxyphenyl N, N-dimethylcarbamate (CUI XING AN) was prepared as potential anticholinesterase agents. Preliminary screening tests showed that the introduction of t-butyl group in CUI XING AN and its homologs decreased the toxicity and anticholinesterase activity. Compounds of type I have increased protective activity against organophosphorus poisoning.

**Key words** Cholinesterase inhibitor; 3-(Alkylamino)alkoxy-4-t-butylphenyl N,N-dimethylcarbamates; 3-(Alkylamino)alkoxy-5-t-butylphenyl N,N-dimethylcarbamates