

[Home](#) > [Journal](#) > [Earth & Environmental Sciences](#) > [JWARP](#)
[Indexing](#) | [View Papers](#) | [Aims & Scope](#) | [Editorial Board](#) | [Guideline](#) | [Article Processing Charges](#)
[JWARP](#) > Vol.2 No.11, November 2010



Removal of Copper Ions from Acid Mine Drainage Wastewater Using Ion Exchange Technique: Factorial Design Analysis

PDF (Size: 414KB) PP. 984-989 DOI : 10.4236/jwarp.2010.211117

Author(s)

R. W. Gaikwad, R. S. Sapkal, V. S. Sapkal

ABSTRACT

A factorial experimental design method was used to examine the "Cu²⁺" removal from acid mine drainage wastewater by ion exchange technique. Ion Exchange technique is preferred because of reduced sludge generation compared to conventional treatment techniques and better decontamination efficiency from highly diluted solutions. Factorial design of experiments is employed to study the effect of four factors pH (3, 5, and 6), flow rate (5, 10, 15 L/hr), resin bed height (20, 40 and 60 cm) and initial concentration of the metal (100, 150 and 200 mg/l) at three levels. The efficiency of metal removal was determined after 100 min of treatment. Main effects and interaction effects of the four factors were analyzed using statistical techniques. A regression model was recommended and it was found to fit the experimental data very well. The results were analyzed statistically using the Student's t-test, analysis of variance, F-test and lack of fit to define most important process variables affecting the percentage "Cu²⁺" removal. In this study, pH was thus found to be the most important variable.

KEYWORDS

Acid Mine Drainage, Ion Exchange, Heavy Metal, Copper, Removal, Factorial Design of Experiments

Cite this paper

R. Gaikwad, R. Sapkal and V. Sapkal, "Removal of Copper Ions from Acid Mine Drainage Wastewater Using Ion Exchange Technique: Factorial Design Analysis," *Journal of Water Resource and Protection*, Vol. 2 No. 11, 2010, pp. 984-989. doi: 10.4236/jwarp.2010.211117.

References

- [1] Kleinman, R.L.P., Acid mine drainage, US bureau of mines researches and develops: control methods for both coal and metal mines. *Environmental Science and Technology*, 24 (9), 1990, pp. 1278-1285.
- [2] Fyson, A., Kalin, M., Adrian, L.W., Arsenic and nickel removal by wetland sediments. In: *Proceedings of the International Land Reclamation and Mine Drainage Conference and Third International Conference on the Abatement of Acidic Drainage*, Vol. 1. Pittsburgh, PA, April 1994, pp. 109-118.
- [3] Clarke, L., Coal mining and water quality. *Journal of Mines Metals and Fuels* 44, 1996, pp. 181-183.
- [4] Kuyucak, N., Acid mine drainage prevention and control options. *CIM Bulletin* 95 (1060), 2002, pp. 96-102.
- [5] Filipek, L.H., Hatton, C., Gusek, J., Tsukamoto, T., Passive treatment of acid rock drainage (ARD): state of the practice. In: *Proceedings of the Tenth International Conference on Tailings and Mine Waste*, October, 2003, Colorado, USA, 2003, pp. 293-303.
- [6] Modis, K., Adam, K., Panagopoulos, K., Komtopoulos, A., Development and Validation of a geostatistical model for prediction of acid mine drainage in underground sulphide mines. *J. Trans. Instn. Min. Metall. (Sect A: Min. Industry)*, 1998, pp. A102-A107.
- [7] Gaikwad R.W. and Gupta D.V., Review On Removal Of Heavy Metals From Acid Mine Drainage, *Applied Ecology And Environmental Research* 6(3): 2008, pp. 79-96.

- [Open Special Issues](#)
- [Published Special Issues](#)
- [Special Issues Guideline](#)

[JWARP Subscription](#)
[Most popular papers in JWARP](#)
[About JWARP News](#)
[Frequently Asked Questions](#)
[Recommend to Peers](#)
[Recommend to Library](#)
[Contact Us](#)

Downloads:	402,262
------------	---------

Visits:	1,010,707
---------	-----------

[Sponsors, Associates, and Links >>](#)

- [8] Kim, J. S., Chah, S. and Yi, J., " Preparation of Modified Silica for Heavy Metal Removal," Korean J. Chem. Eng., 17, 2000, pp. 118-121.
- [9] Kim, Y., Lee, B. and Yi, J., " Preparation of Functionalized Mesoporous Silica Containing Magnetite (MSM) for the Removal of Copper Ions in Aqueous Solutions and its Magnetic Separation," Separ. Sci. Technol., Vol. 38, 2003, pp. 2533-2548.
- [10] Lee, B., Kim, Y., Lee, H. and Yi, J., " Synthesis of Functionalized Porous Silicas via Templating Method as Heavy Metal Ion Adsorbents: The Introduction of Surface Hydrophilicity onto the Surface of Adsorbents," Micropor. Mesopor. Mat., 50(1), 2001, pp. 77-90.
- [11] Kim, S. J., Lim, K. H., Joo, K.H., Lee, M. J., Kil, S.G. and Cho, S.Y., " Removal of Heavy Metal Cyanide Complexes by Ion Exchange," Korean J. Chem. Eng., 19(6), 2002, pp. 1078-1084.
- [12] Rengaraj, S. and Moon, S.H., " Kinetics of Adsorption of Co(II) Removal from Water and Wastewater by Ion Exchange Resins," Water Res., 36, 2002, pp. 1783-1793
- [13] Rengaraj, S., Joo, C., Kim, Y. and Yi, J., " Kinetics of Removal of Chromium from Water and Electronic Process Wastewater by Ion Exchange Resins: 1200H, 1500H and IRN97H," J. Hazard. Mater., B102, 2003, pp. 257- 275.