

ISSN: 2175-9146

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Abstract of Published Article

Study of the thermal decomposition of 2,2',4,

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Abstract:

2,2', 4,4', 6,6'- hexanitrostilbene (HNS) is an energetic material. Its thermal and chemical stabilities, which is employed in the Arrhenius parameters (activation energy and pre-exponential factor) were studied and the results were compared with the Kissinger method, applied to DSC's non-isothermal data. The Kissinger method was chosen for this study. The activation energy determined for HNS revealed values from 428 kJ.mol^{-1} to 477 kJ.mol^{-1} were employed.

Keywords:

Explosives, HNS, DSC, FT-IR.



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