

## 直接和间接电合成制备出的高铁酸钾的物理和电化学性能

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摘要 K<sub>2</sub>FeO<sub>4</sub> powders were synthesized by the ex-situ and in-situ electrochemical methods, respectively, and characterized by infrared spectrum (IR), scanning electron microscopy (SEM), X-ray powder diffraction (XRD) and BET. Their electrochemical performances were investigated by means of galvanostatic discharge and electrochemical impedance spectroscopy (EIS). The results of physical characterization showed that the two samples have similar structural features, but their surface morphologies and oriented growth of the crystals are different, which results in smaller specific surface area and lower solubility of the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> sample. The results of discharge experiments indicated that the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode has much larger discharge capacity and lower electrode polarization than the in-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode. It was found from the results of EIS that lower electrochemical polarization might be responsible for the improvement on the discharge performance of the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode.

关键词 [K<sub>2</sub>FeO<sub>4</sub>](#) [electrosynthesis](#) [discharge performance](#) [electrochemical impedance spectroscopy \(EIS\)](#)

分类号

Physical properties and electrochemical performance of solid K<sub>2</sub>FeO<sub>4</sub> samples prepared by ex-situ and in-situ electrochemical methodsXU ,Zhihua<sup>a</sup>, ,WANG ,Jianming<sup>a</sup>, SHAO ,Haibo<sup>a</sup>, ZHANG Jianqing<sup>a,b</sup><sup>a</sup> Department of Chemistry, Zhejiang University, Hangzhou 310027, China<sup>b</sup> Chinese State Key Laboratory for Corrosion and Protection, Shenyang 110015, China

## Abstract

K<sub>2</sub>FeO<sub>4</sub> powders were synthesized by the ex-situ and in-situ electrochemical methods, respectively, and characterized by infrared spectrum (IR), scanning electron microscopy (SEM), X-ray powder diffraction (XRD) and BET. Their electrochemical performances were investigated by means of galvanostatic discharge and electrochemical impedance spectroscopy (EIS). The results of physical characterization showed that the two samples have similar structural features, but their surface morphologies and oriented growth of the crystals are different, which results in smaller specific surface area and lower solubility of the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> sample. The results of discharge experiments indicated that the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode has much larger discharge capacity and lower electrode polarization than the in-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode. It was found from the results of EIS that lower electrochemical polarization might be responsible for the improvement on the discharge performance of the ex-situ electro synthesized K<sub>2</sub>FeO<sub>4</sub> electrode.

Key words [K<sub>2</sub>FeO<sub>4</sub>](#) [electrosynthesis](#) [discharge performance](#) [electrochemical impedance spectroscopy \(EIS\)](#)

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