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EFFECT OF TEMPERATURE ON A FREE ENERGY AND EQUILIBRIUM CONSTANTS DURING DRY FLUE GAS DESULPHURISATION CHEMICAL REACTIONS

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ABSTRACT

During dry flue gas desulphurisation (FGD) dry particles of reagents are inserted (injected) in the stream of flue gas, where they bond SO2. As reagents, the most often are used compounds of calcium (CaCO3, CaO or Ca(OH)2). Knowledge of free energies and equilibrium constants of chemical reactions during dry FGD is necessary for understanding of influence of flue gas temperature to course of these chemical reactions as well as to SO2 bonding from flue gases.

KEYWORDS

sulphur-dioxid, free energy, equilibrium constant, temperature

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