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## THERMODYNAMICS

一元契合物序下舱相对二甲来、乙来在国夜高限为303.15—353.15K下的密度、载度及某相关性质 标长生、3/84生、周琦 Key Laboratory for Green Chemical Technology of State Education Ministry, School of Chemical Engineering and Technology, Tianjin University, Tianjin 300072, China 收割订期 每月1期 阿格威克各订期 接受订期

保制用 Bmilital 用物な(socialise that many systems of sulfolane+ethylbenzene, sulfolane+ethylbenzene, sulfolane+p-xylene have been experimentally determined in temperature interval 303.15— 353.15 equal not atmospheric pressure for the whost open composition range per secure for the streeth of the size, shape and interaction of components on excess molar volumes and viscosity deviations were computed. The computed quantities have been fitted to Redlich-incompared per secure for the whost open and viscosity deviations show a systematic change with numerical numerical change with numerical change with numerical numerical

Densifies, Viscosities and Related Properties for Binary Mixtures of Saliblane+p-Xylene, Saliblane+p-Xylene, Saliblane+Ehlythenzene in the Temperature Range from 383.15K to 383.15K
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Key Laboratory for Green Charical Temples of Temperature Range from 383.15K to 383.15K
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Abstract Densities and viscosities of the basic yourself of the whole composition range. The excess nodar volumes and viscosity deviations were compated. The composed quantities have been fined to Redlich-Rich reposition cannot volume that whosely deviation for a systemic change with mercaling improveme. Two natures calable ragarities cases when which occurs appoint many at y = 0.5. The effect of the size, shape and interaction of components on excess nodar volumes and viscosity deviations is discounted.

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