REACTION KINETICS, CATALYSIS AND

环戊二烯加氢宏观动力学的研究

本区、一級国金を表現分子中間で、 整性点⁴D. 松本^{ME}. 夏春曜^A 周^E B. 房最^E ^a Shanghai Petrochemical Company Ltd., Shanghai 200540, China ^b College of Chemical Engineering, East China University of Science and Technology, Shanghai 200237, China 収載日期 第日日期 网络版及在日期 投与国民

摘要 The macrokinetics of hydrogenation of cyclopentadiene was investigated over Pdy-

AZIO actalates. Experimental results showed that the relationship between the constituents and reaction time was in agreement with the characteristic of consecutive irreversible first-order reaction. Analysis on the reaction mechanism of selective hydrogenation of cyclopentadiene indicated that it is reasonable to experimental data. Prom the statistic test and residual error distribution the kinetic model was proved to be adequate.

关键词 <u>环戊二烯</u> <u>环戊烯</u> <u>加氢反应 宏观动力学</u> <u>Pd/γ-Al2O3催化剂</u>

分类号

Kinetics on Hydrogenation of Cyclopentadiene over ${\rm Pd}/\gamma{\rm -Al}_2{\rm O}_3$ Catalyst

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Abstract The macrokinetics of hydrogenation of cyclopenatidne was investigated over Pdy-ADO3 catalyst. Experimental results showed that the relationship between the constituents and reaction time was in agreement with the characteristic of consecutive irreversible first-order reaction. Analysis on the reaction mechanism of selective hydrogenation of cyclopenation disclosed that it is reasonable to express the hydrogenation rate of cyclopenation in the power law form. Parameters of the kinetic model were obtained by the Gauss-Newton method based on the experimental data. From the statistic test and residual error distribution the kinetic model was proved to be adequate.

Key words cyclopentadiene; cyclopentene; kinetic; mechanism; hydrogenation

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