

RESEARCH PAPERS

在管式填充床反应器中甲醇部分氧化蒸汽重整制氢的模拟研究

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摘要 Hydrogen production by partial oxidation steam reforming of methanol over a Cu/ZnO/Al₂O₃ catalyst has been paid more and more attention. The chemical equilibria involved in the methanol partial oxidation steam reforming reaction network such as methanol partial oxidation, methanol steam reforming, decomposition of methanol and water-gas shift reaction have been examined over the ranges of temperature 473-1073 K under normal pressure. Based on the detailed kinetics of these reactions over a Cu/ZnO/Al₂O₃ catalyst, and from the basic concept of the effectiveness factor, the intraparticle diffusion limitations were taken into account. The effectiveness factors for each reaction along the bed length were calculated. Then important results were offered for the simulation of this reaction process.

关键词 [methanol partial oxidation steam reforming](#) [chemical equilibria](#) [diffusional limitations](#) [effectiveness factor](#)

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Simulation Studies of the Hydrogen Production from Methanol Partial Oxidation Steam Reforming by a Tubular Packed-bed Catalytic Reactor

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Abstract Hydrogen production by partial oxidation steam reforming of methanol over a Cu/ZnO/Al₂O₃ catalyst has been paid more and more attention. The chemical equilibria involved in the methanol partial oxidation steam reforming reaction network such as methanol partial oxidation, methanol steam reforming, decomposition of methanol and water-gas shift reaction have been examined over the ranges of temperature 473-1073 K under normal pressure. Based on the detailed kinetics of these reactions over a Cu/ZnO/Al₂O₃ catalyst, and from the basic concept of the effectiveness factor, the intraparticle diffusion limitations were taken into account. The effectiveness factors for each reaction along the bed length were calculated. Then important results were offered for the simulation of this reaction process.

Key words [methanol partial oxidation steam reforming](#); [chemical equilibria](#); [diffusional limitations](#); [effectiveness factor](#)

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