RESEARCH PAPERS

丙烯腈与N-乙烯基吡咯烷酮在H₂O/DMSO混合试剂中共聚反应动力学研究

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摘要 Kinetics of copolymerization of acrylonitrile(AN) with N-vinylpyrrolidone(NVP) initiated by azodiisobutyronitrile(AIBN) was investigated in H2O/dimethyl sulphoxide(DMSO) mixture between 50°C and 70°C under N2 atmosphere. The rate of copolymerization and particle size were measured respectively. The kinetic equation 0.651 ± 0.123 1.59 ± 0.35 1.17 ± 0.14 of copolymerization system is obtained as Rp \propto cAIBN cAN cNVP at 60°C. The overall activation energy for copolymerization system was computed as 87.3 kJ.mol-1. Effect of additives of hydroquinone and dioxane on copolymerization was discussed. The polymerization is retarded

by hydroquinone and accelerated by dioxane, which confirms the free radical copolymerization of AN with NVP.

关键词 <u>acrylonitrile</u> <u>N-vinylpyrrolidone</u> <u>copolymerization kinetics</u>

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Kinetics of Copolymerization of Acrylonitrile with N-Vinyl pyrrolidone in $\rm H_2O/Dimethyl$ Sulphoxide Mixture

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Abstract Kinetics of copolymerization of acrylonitrile(AN) with N-vinylpyrrolidone(NVP) initiated by azodiisobutyronitrile(AIBN) was investigated in H2O/dimethyl sulphoxide(DMSO) mixture between 50°C and 70°C under N2 atmosphere. The rate of copolymerization and particle size were measured respectively. The kinetic equation 0.651 ± 0.123 1.59 ± 0.35 1.17 ± 0.14 of copolymerization system is obtained as Rp \propto cAIBN cAN cNVP at 60°C. The overall activation energy for copolymerization system was computed as 87.3 kJ.mol-1. Effect of additives of hydroquinone and dioxane on copolymerization was discussed. The polymerization is retarded by hydroquinone and accelerated by dioxane, which confirms the free radical copolymerization of AN with NVP.

Key words acrylonitrile; N-vinylpyrrolidone; copolymerization kinetics

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