

硼掺杂碳化硅负载Pt催化剂的甲醇电催化氧化性能

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Boron-doped silicon carbide supported Pt catalyst for methanol electrooxidation

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摘要 以硼掺杂碳化硅 ($B_{0.1}SiC$) 为载体, 采用循环伏安法在 $B_{0.1}SiC$ 载体上电沉积Pt纳米粒子制备了 $Pt/B_{0.1}SiC$ 催化剂。利用X射线光电子能谱、X射线衍射、氮气吸附-脱附、扫描电镜及透射电镜等测试方法对催化剂的晶型、表面性质及形貌进行了表征。结果表明, 硼原子掺杂进入SiC晶格并取代了Si位点, 使 $B_{0.1}SiC$ 载体的导电性增强; Pt纳米粒子均匀地分布在 $B_{0.1}SiC$ 载体上, 平均粒径为2.7 nm。与相同条件下制备的Pt/SiC催化剂相比, $Pt/B_{0.1}SiC$ 具有较大的电催化活性面积、更高的甲醇催化氧化活性和稳定性。

关键词: 甲醇电氧化 硼掺杂碳化硅 Pt催化剂

Abstract: Boron-doped silicon carbide ($B_{0.1}SiC$) synthesized by the carbothermal reduction method was used as support to prepare $Pt/B_{0.1}SiC$ catalyst by cyclic voltammetric deposition of Pt nanoparticles. The crystal structure, surface property and morphology of the catalysts were studied with X-ray diffraction, X-ray photoelectron spectroscopy, scanning electron microscopy and transmission electron microscopy techniques and N_2 adsorption-desorption experiment. It is shown that B atoms have been incorporated into the SiC lattice sites by substituting Si, which increases the electrical conductivity of SiC. Pt nanoparticles uniformly dispersed on the $B_{0.1}SiC$ support with an average size of 2.7 nm. The prepared $Pt/B_{0.1}SiC$ had a larger electrochemically active area and exhibited higher electrocatalytic activity and stability for methanol oxidation than the Pt/SiC synthesized by the same method. This shows that B-doped SiC is a promising support for preparing high-performance methanol oxidation electrocatalysts.

Key words: methanol electrooxidation B-doped SiC Pt catalyst

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- [1] WASMUS S, KUIVER A. Methanol oxidation and direct methanol fuel cells: A selective review[J]. *J Electroanal Chem*, 1999, 461(1/2): 14-31. 
- [2] LIU H S, SONG C J, ZHANG L, ZHANG J J, WANG H J, WILKINSON D P. A review of anode catalysis in the direct methanol fuel cell[J]. *J Power Sources*, 2006, 155(2): 95-110. 
- [3] CHEN A C, HOLT-HINDLE P. Platinum-based nanostructured materials: Synthesis, properties, and applications[J]. *Chem Rev*, 2010, 110(6): 3767-3804. 
- [4] ROEN L M, PAIK C H, JARVIC T D. Electrocatalytic corrosion of carbon support in PEMFC cathodes[J]. *Electrochem Solid-State Lett*, 2004, 7(1): A19-A22.
- [5] KANGASNIEMI K H, CONDIT D A, JARVI T D. Characterization of vulcan electrochemically oxidized under simulated PEM fuel cell conditions[J]. *J Electrochem Soc*, 2004, 151(4): E125-E132.
- [6] WANG Y J, WILKINSON D P, ZHANG J J. Noncarbon support materials for polymer electrolyte membrane fuel cell electrocatalysts[J]. *Chem Rev*, 2011, 111(12): 7625-7651. 
- [7] QIU Z, HUANG H, DU J, FENG T, ZHANG W K, GAN Y P, TAO X Y. NbC nanowire-supported Pt nanoparticles as a high performance catalyst for methanol electrooxidation[J]. *J Phys Chem C*, 2013, 117(27): 13770-13775. 
- [8] QIU Z, HUANG H, DU J, TAO X Y, XIA Y, FENG T, GAN Y P, ZHANG W K. Biotemplated synthesis of bark-structured TiC nanowires as Pt catalyst support with enhance electrocatalytic activity and durability for methanol oxidation[J]. *J Mater Chem A*, 2014, 2(21): 8003-8008. 
- [9] FANG L, HUANG X P, VIDAL-IGLESIAS F J, LIU Y P, WANG X L. Preparation, characterization and catalytic performance of a novel Pt/SiC [J]. *Electrochem Commun*, 2011, 13(12): 1309-1312. 
- [10] LV H F, MU S C, CHENG N C, PAN M. Nano-silicon carbide supported catalysts for PEM fuel cells with high electrochemical stability and improved performance by addition of carbon[J]. *Appl Catal B: Environ*, 2010, 100(1/2): 190-196. 
- [11] TONG X L, DONG L L, JIN G Q, WANG Y Y, GUO X Y. Electrocatalytic performance of Pd nanoparticles supported on SiC nanowires for methanol oxidation in alkaline media[J]. *Fuel Cells*, 2011, 11(6): 907-910. 
- [12] DHIMAN R, JOHNSON E, SKOU E M, MORGEN P, ANDERSEN S M. SiC nanocrystals as Pt catalyst supports for fuel cell applications[J]. *J Mater Chem A*, 2013, 1(19): 6030-6036. 
- [13] LIU Z W, SHI Q Q, PENG F, WANG H J, YU H, LI J C, WEI X Y. Enhanced methanol oxidation activity of Pt catalyst supported on the phosphorus-doped multiwalled carbon nanotubes in alkaline medium[J]. *Catal Commun*, 2012, 22: 34-38. 
- [14] LIU Z W, SHI Q Q, PENG F, WANG H J, ZHANG R F, YU H. Pt supported on phosphorus-doped carbon nanotube as an anode catalyst for direct methanol fuel cells[J]. *Electrochem Commun*, 2012, 16(1): 73-76. 
- [15] KRIENER M, MURANAKA T, KATO J, REN Z A, AKIMITSU J, MAENO Y. Superconductivity in heavily boron-doped silicon carbide[J]. *Sci Technol Adv Mater*, 2008, 9(4): 044205. 
- [16] 董莉莉, 王英勇, 童希立, 靳国强, 郭向云. 硼掺杂SiC的制备、表征及其可见光分解水产氢性能[J]. *物理化学学报*, 2014, 30(1): 135-140. (DONG Li-li, WANG Ying-yong, TONG Xi-li, JIN Guo-qiang, GUO Xiang-yun. Synthesis and characterization of boron-doped SiC for visible light driven hydrogen production[J]. *Acta Physico-Chimica Sinica*, 2014, 30(1): 135-140.)
- [17] DONG L L, TONG X L, WANG Y Y, GUO X N, JIN G Q, GUO X Y. Promoting performance and CO tolerance of Pt nanocatalyst for direct methanol fuel cells by supporting on high-surface-area silicon carbide[J]. *J Solid State Electrochem*, 2014, 18(4): 929-934. 
- [18] OSWALD S, WIRTH H. Core-level shifts at B- and Al-doped 6H-SiC studied by XPS[J]. *Surf Interface Anal*, 1999, 27(3): 136-141. 3.0.CO;2-Z target="_blank"> 
- [19] SEO W S, KOUMOTO K, ARAI S. Effects of boron, carbon, and iron content on the stacking fault formation during synthesis of beta-SiC particles in the system SiO₂-C-H₂[J]. *J Am Ceram Soc*, 1998, 81(5): 1255-1261.
- [20] AGATHOPOULOS S. Influence of synthesis process on the dielectric properties of B-doped SiC powders[J]. *Ceram Int*, 2012, 38(4): 3309-3315. 
- [21] XIN Y C, LIU J G, JIE X, LIU W M, LIU F Q, YIN Y, GU J, ZOU Z G. Preparation and electrochemical characterization of nitrogen doped graphene by microwave as supporting materials for fuel cell catalysts[J]. *Electrochim Acta*, 2012, 60: 354-358. 
- [22] RALPH T R, HARDS G A, KEATING J E, CAMPBELL S A, WILKINSON D P, DAVIS M, STPIERRE J, JOHNSON M C. Low cost electrodes for proton exchange membrane fuel cells-Performance in single cells and Ballard stacks[J]. *J Electrochem Soc*, 1997, 144(11): 3845-3857. 
- [23] PARK S J, PARK J M. Preparation and characteristic of platinum catalyst deposited on boron-doped carbon nanotubes[J]. *Curr Appl Phys*, 2012, 12(5): 1248-1251. 
- [24] JEHNG J M, LIU W J, PAN T C, DAI Y M. Preparation of Pt nanoparticles on different carbonaceous structure and their applications to methanol electro-oxidation[J]. *Appl Surf Sci*, 2013, 268: 425-431. 
- [25] MU Y Y, LIANG H P, HU J S, JIANG L, WAN L J. Controllable Pt nanoparticle deposition on carbon nanotubes as an anode catalyst for direct methanol fuel cells[J]. *J Phys Chem B*, 2005, 109(47): 22212-22216. 

[26] GUO S J, DONG S J, WANG E K. Three-dimensional Pt-on-Pd bimetallic nanodendrites supported on graphene nanosheet: facile synthesis and used as an advanced nanoelectrocatalyst for methanol oxidation[J]. *Acs Nano*, 2010, 4(1): 547-555. 

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