

铁铈复合氧化物催化剂SCR脱硝反应动力学研究

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Kinetic study on the selective catalytic reduction of NO with NH₃ over iron-cerium mixed oxide

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摘要 利用共沉淀法制备了铁铈复合氧化物催化剂,在积分实验系统上考察了NO初始浓度、NH₃/NO比及O₂浓度对其SCR脱硝活性的影响;并借助微分系统探讨了其SCR脱硝的催化反应动力学,构建了铁铈复合氧化物催化剂的催化脱硝反应动力学模型.实验结果表明,NO初始浓度越高,每克催化剂的NO转化率越高;随着NH₃/NO比的增加,NO转化率先迅速增加后趋势减缓,最终趋于稳定;O₂在NH₃-SCR反应中起着重要的作用;在175~225 °C下,Fe_{0.95}Ce_{0.05}O₂催化剂的NO和NH₃反应级数分别为1级和0级,O₂的反应级数接近0.5级,该反应的表现活化能为42.6 kJ/mol.

关键词: 铁铈复合氧化物催化剂 选择性催化还原脱硝 NO 反应动力学

Abstract: Iron-cerium mixed metal oxide (Fe_{0.95}Ce_{0.05}O₂) was prepared through co-precipitation method and used as the catalyst in the selective catalytic reduction of NO with NH₃ (NH₃-SCR); the effects of initial concentration of NO, molar ratio of NH₃/NO and O₂ concentration on its NH₃-SCR activity were investigated in an integral reactor at steady-state and the kinetics were considered in a differential reactor. The results indicated that the amount of NO converted over per unit catalyst increases with the initial concentration of NO; with the increase of NH₃/NO mol ratio, NO conversion is firstly increased sharply and then gradually reaches a stable value; O₂ in the flue gas also plays an important role in the NH₃-SCR reaction. The NH₃-SCR reaction over Fe_{0.95}Ce_{0.05}O₂ exhibits first-order with respect to NO, zero-order to NH₃, and nearly 0.5-order to O₂ at 175~275 °C; the apparent activation energy is 42.6 kJ/mol.

Key words: iron-cerium mixed oxide catalyst selective catalytic reduction of NO (SCR) NO kinetics

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