

影响 TiO₂ 纳米管光电催化还原 Cr(VI) 的因素探讨

汪青, 尚静*, 宋寒

北京大学环境科学与工程学院环境模拟与污染控制国家重点联合实验室, 北京 100871

WANG Qing, SHANG Jing*, SONG Han

State Key Joint Laboratory of Environmental Simulation and Pollution Control, College of Environmental Sciences and Engineering, Peking University, Beijing 100871, China

- 摘要
- 参考文献
- 相关文章

Download: PDF (512KB) [HTML](#) (1KB) Export: BibTeX or EndNote (RIS) Supporting Info

摘要 采用阳极氧化法在 Ti 基底上生长 TiO₂ 纳米管 (TNT) 阵列薄膜, 利用扫描电镜和 X 射线衍射对样品的形貌和结构进行了表征. 以 TNT 薄膜为光阴极、Ti 网为光阴极, 在酸性条件下进行紫外光光电催化还原 Cr(VI), 探讨了外加偏压、NaCl 浓度、柠檬酸浓度以及 Cr(VI) 初始浓度对反应性能的影响. 结果表明, 外加电压有效地促进了电子和空穴的分离, 使 Cr(VI) 的光电催化还原效率明显高于光催化; NaCl 的引入增加了体系的导电性, 可在一定程度上提高 Cr(VI) 的光电还原反应速率. 另外, 柠檬酸作为空穴捕获剂可有效地捕获光生空穴, 在一定浓度范围内, Cr(VI) 的光电还原反应速率随柠檬酸浓度的增加而增加. 在 1.5 V 偏压, 1.0 mol/L NaCl, 0.5 mmol/L 柠檬酸条件下, 初始浓度为 17.7 mg/L 的 Cr(VI) 反应 60 min 时, 其转化率达到 98.6%.

关键词: 二氧化钛纳米管 六价铬 光电催化 还原

Abstract: TiO₂ nanotube (TNT) arrays present excellent light trapping and electronic conduction ability. TNT film was prepared by the anodic oxidation method using Ti slice as the substrate. The morphology and structure of the TNT film were characterized by scanning electron microscopy and X-ray diffraction, respectively. Photoelectrocatalytic (PEC) reduction of Cr(VI) was carried out using the TNT arrays as the photoanode and Ti mesh as the photocathode under acid conditions and ultraviolet light irradiation. The influencing factors such as applied voltage, NaCl concentration, citric acid concentration, and Cr(VI) initial concentration were investigated. The applied voltage facilitates the separation of electron-hole pairs, making the PEC reaction rate of Cr(VI) much higher than that of photocatalysis. The introduction of NaCl increases the electrical conductivity of solution, getting enhanced PEC reaction rate of Cr(VI) to some extent. Citric acid, behaving as the hole scavenger, can capture photogenerated holes effectively. The PEC reaction rate of Cr(VI) increases with increasing citric acid concentration in a certain range. Under the conditions of 1.5 V voltage, 1.0 mol/L NaCl, and 0.5 mmol/L citric acid, the conversion of 17.7 mg/L Cr(VI) achieved 98.6% at 60 min irradiation.


Keywords: titania nanotube, chromium (VI), photoelectrocatalysis, reduction


收稿日期: 2011-05-18; 出版日期: 2011-08-17


引用本文:
汪青, 尚静, 宋寒. 影响 TiO₂ 纳米管光电催化还原 Cr(VI) 的因素探讨[J]. 催化学报, 2011, V32(9): 1525-1530

WANG Qing, SHANG Jing, SONG Han. Factors Influencing the Photoelectrocatalytic Reduction of Cr(VI) over TiO₂ Nanotube Arrays[J]. Chinese Journal of Catalysis, 2011, V32(9): 1525-1530


链接本文:
<http://www.chxb.cn/CN/10.3724/SP.J.1088.2011.10522> 或 <http://www.chxb.cn/CN/Y2011/V32/I9/1525>


[1] osta M. Toxicol Appl Pharmacol, 2003, 188: 1 


[2] osta M, Klein C B. Crit Rev Toxicol, 2006, 36: 155 

[3] chrank S G, Jose H J, Moreira R F P M. J Photochem Photobiol A, 2002, 147: 71 

[4] 亮. 环境技术 (Liao L. Environ Technol), 2002, (1): 19

[5] u Y, Jung I L. Water Res, 2001, 35: 135 

[6] uxel A A, Gimenez S M N, Almeida F A S, Alfaya R V S, Alfaya A A S. Clean-Soil Air Water, 2011, 39: 289 

[7] ing Y Q, Chen X M, Wang D H. Environ Sci Technol, 2007, 41: 1439 

[8] iu A J, Ding S L, Zhao C C, Ren H J, Zhao Q Y. J Soc Leath Technol Chem, 2007, 91: 52

Service

- ▶ 把本文推荐给朋友
- ▶ 加入我的书架
- ▶ 加入引用管理器
- ▶ Email Alert
- ▶ RSS

作者相关文章

- ▶ 汪青
- ▶ 尚静
- ▶ 宋寒

- [9] ang L M, Wang N, Zhu L H, Yu H W, Tang H Q. *J Hazard Mater*, 2008, 152: 93 
- [10] Yu H B, Chen S, Quan X, Zhao H M, Zhang Y B. *Environ Sci Technol*, 2008, 42: 3791 
- [11] Yoon J, Shim E, Bae S, Joo H. *J Hazard Mater*, 2009, 161: 1069 
- [12] Testa J, Grela M A, Litter M I. *Environ Sci Technol*, 2004, 38: 1589 
- [13] Papadam T, Xekoukoulotakis N P, Poullos I, Mantzavinos D. *J Photochem Photobiol A*, 2007, 186: 308 
- [14] Liu Z, Zhang X, Nishimoto S, Jin M, Tryk D A, Murakami T, Fujishima A. *J Phys Chem C*, 2008, 112: 253 
- [15] Paschoal F M M, Anderson M A, Zanoni M V B. *J Braz Chem Soc*, 2008, 19: 803 
- [16] Paschoal F M M, Anderson M A, Zanoni M V B. *J Hazard Mater*, 2009, 166: 531 
- [17] Zhang Z H, Yuan Y, Shi G Y, Fang Y J, Liang L H, Ding H C, Jin L T. *Environ Sci Technol*, 2007, 41: 6259 
- [18] 魏凤玉, 桑蕾. *催化学报* (Wei F Y, Sang L. *Chin J Catal*), 2009, 30: 335
- [19] 温艳媛, 丁昆明. *催化学报* (Wen Y Y, Ding H M. *Chin J Catal*), 2011, 32: 36 
- [20] Wang Q, Shang J, Zhu T, Zhao F W. *J Mol Catal A*, 2011, 335: 242 
- [21] Yang L X, Xiao Y, Liu S H, Li Y, Cai Q Y, Luo S L, Zeng G M. *Appl Catal B*, 2010, 94: 142 
- [22] Shang J, Zhao F W, Zhu T, Wang Q, Song H, Zhang Y C. *Appl Catal B*, 2010, 96: 185 
- [23] Macedo L C, Zaia D A M, Moore G J, de Santana H. *J Photochem Photobiol A*, 2007, 185: 86 
- [1] 张海艳, 曹春晖, 赵健, 林瑞, 马建新. 燃料电池Pt 基核壳结构电催化剂的最新研究进展[J]. *催化学报*, 2012,33(2): 222-229
- [2] 闫朝阳, 兰丽, 陈山虎, 赵明, 龚茂初, 陈耀强*. 高性能 $Ce_{0.5}Zr_{0.5}O_2$ 稀土储氧材料的制备及其负载的单 Pd 三效催化剂[J]. *催化学报*, 2012,33(2): 336-341
- [3] 宫绪敏, 聂尧, 徐岩, 肖荣, Gaetano T. MONTELLIONE. 羰基还原酶辅酶结合域位点突变对其不对称催化性能的影响[J]. *催化学报*, 2012,33(2): 330-335
- [4] 任利敏, 张一波, 曾尚景, 朱龙凤, 孙琦, 张海燕, 杨承广, 孟祥举, 杨向光, 肖丰收. 由新型铜胺络合物模板剂设计合成活性优异的 Cu-SSZ-13 分子筛[J]. *催化学报*, 2012,33(1): 92-105
- [5] 刘彤, 于琴琴, 王卉, 蒋晓原, 郑小明. 等离子体与催化剂协同催化 CH_4 选择性还原脱硝反应[J]. *催化学报*, 2011,32(9): 1502-1507
- [6] 郑青¹, 李金花¹, 陈红冲¹, 陈全鹏¹, 周保学^{1,2,*}, 尚树川³, 蔡伟民¹. 基于薄层反应器的有机污染物光电催化氧化反应性能与机理[J]. *催化学报*, 2011,32(8): 135-1363