

分子印迹聚合物负载纳米金催化剂的制备及其底物识别性能

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摘要 以 4-硝基苯甲醇与氯金酸的络合物为模板, 利用聚合物空腔内胺基捕获 NaBH_4 还原的纳米粒子, 设计和制备了一种具有底物识别性能分子印迹聚合物负载纳米 Au 催化剂 (Au/MIP). 运用红外光谱、紫外-可见光谱和扫描电镜等方法对催化剂进行了表征. 同时以水为溶剂, 过氧化氢为氧化剂, 考察了催化剂在取代苯甲醇氧化反应中的催化性能. 结果表明, 以 Au/MIP 为催化剂时, 4-硝基苯甲醇转化率为 75.6%, 而以非印迹聚合物负载的纳米 Au (Au/NIP) 为催化剂时, 4-硝基苯甲醇转化率仅为 41.5%. 以其它取代苯甲醇为底物时, Au/MIP 与 Au/NIP 的催化活性差别不大. 这说明 Au/MIP 催化剂活性与反应底物结构有关, 脱除模板剂后它具有与 4-硝基苯甲醇相匹配的空腔结构和识别位点, 对反应底物表现出专一的识别性, 因而提高了催化剂活性.

关键词: 分子印迹聚合物 金纳米粒子 取代苯甲醇 催化氧化 底物识别性

Abstract: A molecularly imprinted polymer-supported gold nanoparticle (Au/MIP) catalyst, which has the characteristics of specific substrate recognition, was prepared by the template complex of 4-nitrobenzyl alcohol (4-NBA) and hydrogen tetrachloroaurate(III), where Au nanoparticles were formed by the reduction with NaBH_4 solution and captured by amino groups ($-\text{NH}_2$) in the cavities of the MIP. The obtained samples were characterized with FT-IR spectroscopy, UV-Vis spectroscopy, and scanning electron microscopy. The catalytic activity and substrate recognition of the Au/MIP were investigated by the oxidation of substituted benzyl alcohol using H_2O_2 as the oxidant in water. It was found that the conversion of 4-NBA was up to 75.6% over Au/MIP when using the template molecule of 4-NBA as the substrate. However, the conversion of 4-NBA was only 41.5% over non-imprinted polymer-supported gold nanoparticle (Au/NIP) because no template molecule of 4-NBA was used in the preparation of catalyst. Furthermore, no significant difference of the catalytic activity between the catalysts Au/MIP and Au/NIP was observed when other substituted benzyl alcohols were used as the substrate. These results indicated that the catalytic activity of Au/MIP was related to the structure of substrates. The Au/MIP after removal of the template had molecular recognition shape and sites in the cavities matching to the substrate of 4-NBA molecule. The special recognizable cave of the Au/MIP exhibited unique substrate recognition and therefore improved the catalytic activity.

Keywords: molecular imprinted polymer, gold nanoparticle, substituted benzyl alcohol, catalytic oxidation, substrate recognition

收稿日期: 2011-05-14; 出版日期: 2011-07-27







引用本文:

祝贞科, 谭蓉, 孙文庆等. 分子印迹聚合物负载纳米金催化剂的制备及其底物识别性能[J]. 催化学报, 2011, V32(9): 1508-1512

ZHU Zhen-Ke, TAN Rong, SUN Wen-Qing etc. Preparation of Molecularly Imprinted Polymer-Supported Gold Nanoparticles and Their Ability for Specific Substrate Recognition[J]. Chinese Journal of Catalysis, 2011, V32(9): 1508-1512

链接本文:

<http://www.chxb.cn/CN/10.3724/SP.J.1088.2011.10514> 或 <http://www.chxb.cn/CN/Y2011/V32/I9/1508>


















- [1] auling L. J Am Chem Soc, 1940, 62: 2643 
- [2] ulff G, Gross T, Schonfeld R. Angew Chem, Int Ed, 1997, 36: 1962 
- [3] ocatelli F, Gamez P, Lemaire M. J Mol Catal A, 1998, 135: 89 
- [4] trikovsky A G, Kasper D, Grün M, Green B S, Hradil J, Wulff G. J Am Chem Soc, 2000, 122: 6295 
- [5] iu J Q, Wulff G. J Am Chem Soc, 2004, 126: 7452 
- [6] isnjevski A E, Yilmaz E, Brüggemann O. Appl Catal A, 2004, 260: 169 

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- [7] Urri E, Öhm M, Daguinet C, Severin K. *Chem Eur J*, 2005, 11: 5055 
- [8] Li S J, Ge Y, Tiwar A, Wang S Q, Turner A P F, Piletsky S A. *J Catal*, 2011, 278: 173 
- [9] Zhang H, Piacham T, Drew M, Patek M, Mosbach K, Ye L. *J Am Chem Soc*, 2006, 128: 4178 
- [10] Brunkan N M, Gagné M R. *J Am Chem Soc*, 2000, 122: 6217 
- [11] Li Y Z, Fan Y N, Yang H P, Xu B L, Feng L Y, Yang M F, Chen Y. *Chem Phys Lett*, 2003, 372: 160 
- [12] Tang Z C, Li Q Y, Lu G X. *Carbon*, 2007, 45: 41 
- [13] Li S J, Gong S Q. *Adv Funct Mater*, 2009, 19: 2601 
- [14] Viton F, White P S, Gagne M R. *Chem Commun*, 2003, 3040
- [15] Hashm A S K. *Chem Rev*, 2007, 107: 3180 
- [16] Yadav G D, Mistry C K. *J Mol Catal A*, 2001, 172: 135 
- [17] Liu C, Tan R, Yu N Y, Yin D H. *Microporous Mesoporous Mater*, 2010, 131: 162 
- [18] Li S J, Liao C, Li W K, Chen Y F, Hao X. *Macromol Bio-sci*, 2007, 7: 1112 
- [19] Zhang D L, Li S J, Li W K, Chen Y F. *Catal Lett*, 2007, 115: 169 
- [20] Park J E, Momma T, Osaka T. *Electrochim Acta*, 2007, 52: 5914 
- [21] Liu X H, Hong Y, Li S J, Li W K. *J Inorg Organomet Polym*, 2009, 19: 335 
- [22] Itoh H, Naka K, Chujo Y. *J Am Chem Soc*, 2004, 126: 3026 
- [23] Luo L R, Yu N Y, Tan R, Jin Y, Yin D L, Yin D H. *Catal Lett*, 2009, 130: 489 
- [24] 庄大英, 金勇, 喻宁亚, 秦亮生, 刘建福, 银董红, 杨翠清. *催化学报* (Zhuang D Y, Jin Y, Yu N Y, Qin L Sh, Liu J F, Yin D H, Yang C Q. *Chin J Catal*), 2009, 30: 896
- [25] Hoshino Y, Koide H, Urakami T, Kanazawa H, Kodama T, Oku N, Shea K J. *J Am Chem Soc*, 2010, 132: 6644 
- [26] Wu X Y, Goswami K, Shimizu K D. *J Mol Recognit*, 2008, 21: 410 
- [27] Zayats M, Kanwar M, Ostermeier M, Searson P C. *Macromolecules*, 2011, 44: 3966 

- [1] 宋磊, 陈天虎, 李云霞, 刘海波, 孔德军, 陈冬. 凹凸棒石负载的 Cu-Mn-Ce 催化剂上甲苯氧化反应性能[J]. *催化学报*, 2011,32(4): 652-656
- [2] 马丁. 《物理化学学报》2010 年刊发催化领域文章的述评[J]. *催化学报*, 2011,32(12): 1880-1884
- [3] 刘雪松;鲁继青;王晓霞;罗孟飞.OMS-2 的制备及其负载 PdO 对 CO 氧化的催化活性[J]. *催化学报*, 2010,31(2): 181-185
- [4] 庄大英;金勇;喻宁亚;秦亮生;刘建福;银董红;杨翠清 .有机官能化介孔硅基材料负载纳米金催化剂的制备及其催化性能[J]. *催化学报*, 2009,30(9): 896-900
- [5] 夏峰;刘肇敏;范彬彬;李瑞丰.碱金属离子对 Fe(phen)3/Y 催化氧化性能的影响[J]. *催化学报*, 2009,30(7): 601-605
- [6] 陈敏;马莹;宋萃;张婷;郑小明.Ce-Pt-Pd/不锈钢丝网催化剂的制备与催化性能[J]. *催化学报*, 2009,30(7): 649-653
- [7] 梁镇海;孙红艳;崔玉青.环己醇在Ti/Ni/NiO电极上的催化氧化[J]. *催化学报*, 2009,30(3): 254-258
- [8] 李美超;汪伍洋;马淳安.一种提高铂电催化氧化甲酸性能的简单方法[J]. *催化学报*, 2009,30(11): 1073-1075
- [9] 谢芳;伏再辉;袁晨;叶正培;周小平;刘凤兰;卢春丽;荣春英;毛利秋;尹笃林.六齿八羟基喹啉锰类配合物催化二甲亚砷的氧化消除[J]. *催化学报*, 2009,30(10): 981-985
- [10] 高立达;薛青松;路勇;何鸣元.汽油中正戊硫醇的催化空气氧化脱除 I . CuZnAl复合氧化物催化剂的催化活性[J]. *催化学报*, 2008,29(7): 633-637
- [11] 王新葵;张万生;王爱琴;王晓东;杨学锋;张涛.Au/Fe2O3/Al2O3催化剂上丙烯选择催化还原NO[J]. *催化学报*, 2008,29(6): 503-505
- [12] 皮展;蔡黎;钟俊波;龚茂初;陈耀强.Ce0.5-xZr0.5-xBa2xO2负载TiO2光催化降解气相苯[J]. *催化学报*, 2008,29(5): 453-457
- [13] 刘依农;许志明;孙学文;赵锁奇;王仁安;龙军.四(五氟化苯基)-六, 七溴-吡啶氯化铁选择性催化氧化超临界丙烷的反应机理[J]. *催化学报*, 2007,28(5): 423-428
- [14] 关业军;李灿.氧化铈的氧化还原性能对VOx/CeO2催化剂催化氯苯氧化性能的影响[J]. *催化学报*, 2007,28(5): 392-394
- [15] 郭婷;白志鹏;吴灿;朱坦.环境湿度对TiO2/活性炭纤维气-固光催化氧化甲苯的影响[J]. *催化学报*, 2007,28(12): 1089-1095