

镍促进 CuO-CeO₂ 催化剂的结构表征及低温 CO 氧化活性

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摘要 制备了一系列 CO 低温氧化的 Ce₂₀Cu₅Ni_yO_x 催化剂, 并采用氮气低温物理吸附、X 射线衍射、程序升温还原、X 射线光电子能谱以及拉曼光谱等手段对催化剂进行表征。结果表明, Ce₂₀Cu₅Ni_{0.4}O_x 催化剂活性最高。NiO 的添加可以使较多的 Cu 物种掺杂到 CeO₂ 晶格中, 通过形成铈镍固溶体产生更多的氧空位。表征结果显示, Ce₂₀Cu₅Ni_{0.4}O_x 催化剂中存在大量的 Cu⁺, Ce³⁺ 及晶格氧, 催化剂中的 Cu⁺ 很容易进入到氧化铈晶格, 形成 Cu-O-Ce 固溶体, 从而增强了在还原气氛下晶格氧的释放能力。Ce₂₀Cu₅Ni_{0.4}O_x 催化剂高的催化活性主要归因于大量 Cu⁺ 以及形成的 Cu-O-Ce 和 Ni-O-Ce 固溶体。

关键词: 铜 镍 一氧化碳 氧化 镍 固溶体

Abstract: A series of Ce₂₀Cu₅Ni_yO_x catalysts for CO oxidation at low temperature were prepared and characterized by N₂ adsorption, X-ray diffraction, temperature-program reduction by H₂, X-ray photoelectron spectroscopy (XPS), and Raman spectroscopy. Ce₂₀Cu₅Ni_{0.4}O_x exhibited the highest catalytic activity. The addition of NiO increased the amount of copper ions doped into the CeO₂ matrix and gave more oxygen vacancies in ceria by the formation of a Ni-O-Ce solid solution. XPS results showed that large quantities of Cu⁺, Ce³⁺, and lattice oxygen existed in the fresh Ce₂₀Cu₅Ni_{0.4}O_x catalyst. Cu⁺ ions in the catalyst can easily migrate to the ceria lattice to form a Cu-O-Ce solid solution, which enhanced the release of the lattice oxygen of the oxides under a reducing atmosphere. The high catalytic activity of Ce₂₀Cu₅Ni_{0.4}O_x is due to the promoter giving increased amounts of Cu⁺ in the catalyst and the formation of solid solutions of both Cu-O-Ce and Ni-O-Ce.

Keywords: copper, cerium, carbon monoxide, oxidation, nickel, solid solution

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