催化学报 » 2011, Vol. 32 » Issue (5):771-776 DOI: 10.1016/S1872-2067(10)60185-4

研究论文

最新目录 | 下期目录 | 过刊浏览 | 高级检索

<< ◀◀ 前一篇

一篇 >>

>>

Preparation of a Nickel Molybdenum Carbide Catalyst and Its Activity in the Dry Reforming of Methane

Taro HIROSE1, Yasushi OZAWA2, Masatoshi NAGAI1,*

1Graduate School of Bio-Applications and Systems Engineering, Tokyo University of Agriculture and Technology, 2-24 Nakamachi, Koganei, Tokyo 184-8588, Japan; 2Energy Engineering Research Laboratory, Central Research Institute of Electric Power Industry, 2-6-1 Nagasaka, Yokosuka, Kanagawa 240-0196, Japan

Taro HIROSE1, Yasushi OZAWA2, Masatoshi NAGAI1,*

1Graduate School of Bio-Applications and Systems Engineering, Tokyo University of Agriculture and Technology, 2-24 Nakamachi, Koganei, Tokyo 184-8588, Japan; 2Energy Engineering Research Laboratory, Central Research Institute of Electric Power Industry, 2-6-1 Nagasaka, Yokosuka, Kanagawa 240-0196, Japan

- 摘要
- 参考文献
- 相关文章

Download: PDF (247KB) HTML (1KB) Export: BibTeX or EndNote (RIS) Supporting Info

摘要 Nickel molybdenum carbide catalysts were prepared and their activities in the CO_2 reforming of methane at a low CO_2/CH_4 reactant ratio were investigated using a microreactor at atmospheric pressure and at 973 K. The effect of the catalyst preparation method and the Ni/Mo ratio on the increase in catalyst life and the promotion of catalytic activity were investigated using N_2 adsorption, X-ray diffraction, temperature-programmed carburization, temperature-programmed reaction, and a reforming reaction. The 25Ni75Mo catalyst that was carburized at 813 K exhibited the highest hydrogen formation ability and gave the least carbon deposition. The incomplete carburization of the Mo oxide species in the catalyst that was carburized at a lower temperature gradually gave a more active carburized species. The NiMoO_xC_y in the catalyst was more active in hydrogen formation during the dry reforming of methane while β-Mo₂C and η-Mo₃C₂ were less active.

关键词: catalyst activation catalyst selectivity fuel dry reforming methane

Abstract: Nickel molybdenum carbide catalysts were prepared and their activities in the ${\rm CO_2}$ reforming of methane at a low ${\rm CO_2/CH_4}$ reactant ratio were investigated using a microreactor at atmospheric pressure and at 973 K. The effect of the catalyst preparation method and the Ni/Mo ratio on the increase in catalyst life and the promotion of catalytic activity were investigated using ${\rm N_2}$ adsorption, X-ray diffraction, temperature-programmed carburization, temperature-programmed reaction, and a reforming reaction. The 25Ni75Mo catalyst that was carburized at 813 K exhibited the highest hydrogen formation ability and gave the least carbon deposition. The incomplete carburization of the Mo oxide species in the catalyst that was carburized at a lower temperature gradually gave a more active carburized species. The NiMoO_XC_y in the catalyst was more active in hydrogen formation during the dry reforming of methane while β -Mo₂C and η -Mo₃C₂ were less active.

Keywords: catalyst activation, catalyst selectivity, fuel, dry reforming, methane

收稿日期: 2010-10-14; 出版日期: 2011-01-11

引用本文:

Taro HIROSE, Yasushi OZAWA, Masatoshi NAGAI .Preparation of a Nickel Molybdenum Carbide Catalyst and Its Activity in the Dry Reforming of Methane[J] 催化学报, 2011,V32(5): 771-776

Taro HIROSE, Yasushi OZAWA, Masatoshi NAGAI .Preparation of a Nickel Molybdenum Carbide Catalyst and Its Activity in the Dry Reforming of Methane[J] Chinese Journal of Catalysis, 2011,V32(5): 771-776

链接本文:

http://www.chxb.cn/CN/10.1016/S1872-2067(10)60185-4 或 http://www.chxb.cn/CN/Y2011/V32/I5/771

- [1] shcroft A T, Cheetham A K, Green M L H, Vernon P D F. Nature, 1991, 352: 225
- [2] agai M, Nakahira K, Ozawa Y, Namiki Y, Suzuki Y. Chem Eng Sci, 2007, 62: 4998
- [3] hen X, Honda K, Zhang Z-G. Appl Catal A, 2005, 288: 86
- an-José-Alonso D, Juan-Juan J, Illán-Gómez M J, Román-Martínez M C. Appl Catal A, 2009, 371: 54
- [5] oh H, Jung Y, Koo K Y, Jung U H, Seo Y, Yoon W L. Appl Catal A, 2009, 383: 156
- [6] iu D P, Quek X Y, Cheo W N E, Lau R, Borgna A, Yang Y H. J Catal, 2009, 266: 380
- olymosi F, Cserényi J, Szöke A, Bánsági T, Oszkó A. J Catal, 1997, 165: 150

Service

- ▶ 把本文推荐给朋友
- ▶ 加入我的书架
- ▶ 加入引用管理器
- ▶ Email Alert
- ▶ RSS

作者相关文章

- ▶ Taro HI ROSE
- Yasushi OZAWA
- Masatoshi NAGAI

Shu Y, Ohnishi R, Ichikawa M. J Catal, 2002, 206: 134 [15] York A P E, Claridge J B, Brungs A J, Tsang S C, Green M L H. Chem Commun, 1997: 39 [16] Brungs A J, York A P E, Claridge J B, Márquez-Alvarez C, Green M L H. Catal Lett, 2000, 70: 117 [17] Lacheen H S, Iglesia E. J Catal, 2005, 230: 173 [18] Matsuo Y, Yoshinaga Y, Sekine Y, Tomishige K, Fujimoto K. Catal Today, 2000, 63: 439 [19] Sehested J, Jacobsen C J H, Rokni S, Rostrup-Nielsen J R. J Catal, 2001, 201: 206 [20] Xiao T, Suhartanto T, York A P E, Sloan J, Green M L H. Appl Catal A, 2003, 253: 225 [21] K. KUMBILIEVA1, L. PETROV2,*, Y. ALHAMED2, A. ALZAHRANI2. Reaction Mechanism and Deactivation Modes of Heterogeneous Catalytic [1] Systems[J]. 催化学报, 2011,32(3): 387-404 Beatriz FIDALGO, J. Angel MENENDEZ*. Carbon Materials as Catalysts for Decomposition and CO₂ Reforming of Methane: A Review[J]. 催化学报, [2] 2011,32(2): 207-216 Ahmed S. A. AL-FATESH, Anis H. FAKEEHA, Ahmed E. ABASAEED. Effects of Selected Promoters on Ni/γ-Al₂O₃ Catalyst Performance in Methane [3] Dry Reforming[J]. 催化学报, 2011,32(10): 1604-1609 Maryam ABDOLRAHMANI; Matin PARVARI*; Mehdi HABIBPOOR. Effect of Copper Substitution and Preparation Methods on the LaMnO3 Structure [4] and Catalysis of Methane Combustion and CO Oxidation[J]. 催化学报, 2010,31(4): 394-403 Alieh KHALESI; Hamid R. ARANDIYAN; Matin PARVARI*.La-Ni-AI钙铁矿氧化物中Sr和Ca取代La对甲醇重整的影响[J]. 催化学报, 2008,29(10): 960-968 [5] G. R. MORADI*, F. KHOSRAVIAN, M. RAHMANZADEH.Effect of Partial Substitution of Ni by Cu in LaNiO₃ Perovskite Catalyst for Dry Methane [6] Reforming[J]. 催化学报, 0,(): 0-Jose M. BERMUDEZ, Beatriz FIDALGO, Ana ARENILLAS, J. Angel MENENDEZ.Mixtures of Steel-Making Slag and Carbons as Catalyst for [7] Microwave-Assisted Dry Reforming of CH4[J]. 催化学报, 0,(): 0-

shikawa K, Nagai M, Omi S. J Phys Chem B, 2001, 105: 9124

Oshikawa K, Nagai M, Omi S. Jpn J Appl Phys, 2000, 39: 4456

Izhar S, Kanesugi H, Tominaga H, Nagai M. Appl Catal A, 2007, 317: 82

LaMont D C, Thomson W J. Appl Catal A, 2004, 274: 173

Nagai M, Kanesugi H. J Chem Eng Jpn, 2006, 39: 569

Nagai M, Matsuda K. J Chem Eng Jpn, 2006, 39: 575

ominaga H, Nagai M. Appl Catal A, 2007, 328: 35

[8] [9]

[10]

[11]

[12]

[13]

[14]