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研究论文

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Alternative Use of Light Emitting Diodes in an Activated Charcoal-Supported Photocatalyst Reactor for the Control of Volatile Organic Compounds

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摘要 The applicability of ultraviolet-light emitting diodes (LEDs) as a light source for photocatalysis using granular activated charcoal (GAC) impregnated with transition metal-enhanced photocatalysts for the control of volatile organic compounds (VOCs) was investigated. Two target compounds (toluene and methyl mercaptan) were selected to evaluate the removal activities of the TiO₂/GAC composites. The photocatalysts were prepared by a sol-gel method. Methyl trimethoxy silane was added as a precursor sol solution to bind the photocatalyst with the GAC. Metal (Zn²⁺, Fe³⁺, Ag⁺, and Cu²⁺) enhanced TiO₂/GAC composites were prepared and tested for their photocatalytic activities under 400 nm LED irradiation. The specific surface area (SSA) and the surface chemical composition of the prepared composites were investigated. The SSAs of all the impregnated composites were similar to those of pure GAC. Both field emission-scanning electron microscopy and energy dispersive spectroscopic analysis confirmed that titanium and the impregnated metals were deposited on the surface of the adsorbent. The breakthrough time for GAC toward toluene or methyl mercaptan gas increased upon photocatalytic impregnation and LED illumination. Using different binders affected the breakthrough time of the TiO₂/GAC composite and the addition of zinc oxide to TiO2 increased the VOC removal capacity of the GAC composite.

关键词: granular activated charcoal titania sol-gel method light emitting diode surface chemical composition zinc oxide

Abstract: The applicability of ultraviolet-light emitting diodes (LEDs) as a light source for photocatalysis using granular activated charcoal (GAC) impregnated with transition metal-enhanced photocatalysts for the control of volatile organic compounds (VOCs) was investigated. Two target compounds (toluene and methyl mercaptan) were selected to evaluate the removal activities of the TiO₂/GAC composites. The photocatalysts were prepared by a sol-gel method. Methyl trimethoxy silane was added as a precursor sol solution to bind the photocatalyst with the GAC. Metal (Zn²⁺, Fe³⁺, Ag⁺, and Cu²⁺) enhanced TiO₂/GAC composites were prepared and tested for their photocatalytic activities under 400 nm LED irradiation. The specific surface area (SSA) and the surface chemical composition of the prepared composites were investigated. The SSAs of all the impregnated composites were similar to those of pure GAC. Both field emission-scanning electron microscopy and energy dispersive spectroscopic analysis confirmed that titanium and the impregnated metals were deposited on the surface of the adsorbent. The breakthrough time for GAC toward toluene or methyl mercaptan gas increased upon photocatalytic impregnation and LED illumination. Using different binders affected the breakthrough time of the TiO₂/GAC composite and the addition of zinc oxide to TiO2 increased the VOC removal capacity of the GAC composite.

Keywords: granular activated charcoal, titania, sol-gel method, light emitting diode, surface chemical composition, zinc oxide

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- etin E, Odabasi M, Seyfioglu R. Sci Total Environ, 2003, 312: 103
- iu Y, Shao M, Fu L, Lu S, Zheng L, Tang D. Atmos Environ, 2008, 42: 6247

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- KIM J. S

- JO W. K

- [3] oukos J, Riffault V, Locoge N, Plaisance H. Environ Poll, 2009, 157: 3001
- [4] EHHA (Office of Environmental Health Hazard Assessment) Proposition 65 Status Report Safe Harbor Levels: No Signifi-cant Risk Levels for Carcinogens and Maximum Allowable Dose Levels for Chemicals Causing Reproductive Toxicity. California Environmental Protection Agency, OEHHA, Sac-ramento, CA, 2003
- [5] tkinson R, Arey J. Atmos Environ, 2003, 37(suppl. 2): 197
- estallats H, Lunden M M, Singer B C, Coleman B K, Dodgson A T, Weschler C J, Nazaroff W W. Environ Sci Technol, 2006, 40: 4421
- [7] iu J, Huang Z H, Wang Z S, Kang F Y. J Environ Sci-China, 2004, 16: 53
- [8] iraudet S, Pre P, Tezel H, Le Cloirec P. Carbon, 2006, 44: 2413
- [9] asaki T, Matsumoto A, Yamashita Y. Colloid Surf A, 2008, 325: 166
- [10] Yu F D, Luo L G, Grevillot G. Chem Eng Proc, 2007, 46: 70
- [11] Deveau P A, Arsac F, Thivel P X, Ferronato C, Delpech F, Chovelon J M, Kaluzny P, Monnet C. J Hazard Mater, 2007, 144: 692
- [12] Qi H, Sun D Zh, Chi G Q. J Environ Sci-China, 2007, 19: 1136
- [13] Li D P, Qu J H. J Environ Sci-China, 2009, 21: 713
- [14] Shiraishi F, Yamaguchi S, Ohbuchi Y. Chem Eng Sci, 2003, 58: 929
- [15] Tao Y, Wu C-Y, Mazyck D W. Chemosphere, 2006, 65: 35
- [16] http://en.wikipedia.org/wiki/Light-emitting_diode, 2010
- [17] Ghosh J P, Langford C H, Achari G. J Phys Chem A, 2008, 112: 10310
- [18] Ghosh J P, Sui R, Langford C H, Achari G, Berlinguette C P. Water Res, 2009, 43: 4499
- [19] Tamai H, Nagoya H, Shiono T. J Colloid Interf Sci, 2006, 300: 814
- [20] Corrêa S M, Arbilla G. Atmos Environ, 2008, 42: 6721
- [21] Zhang X W, Zhou M H, Lei L C. Mater Chem Phys, 2005, 91: 73
- [22] Kim Y, Lee J, Jeong H, Lee Y, Umb M H, Jeong K M, Yeo M K, Kang M. J Ind Eng Chem, 2008, 14: 396
- [23] Araña J, Peña Alonso A, Doña Rodríguez J M, Herrera Melián J A, González Díaz O, Pérez Peña J. Appl Catal B, 2008, 78: 355 🛌
- [24] Melghit K, Bouziane K. J Alloy Comp, 2008, 453: 102
- [25] Ao Y H, Xu J J, Fu D G, Shen X W, Yuan C W. Colloid Surf A, 2008, 312: 125
- [26] Chen H W, Ku Y, Irawan A. Chemosphere, 2007, 69: 184
- [27] Matsumoto T, Iyi N, Kaneko Y, Kitamura K, Ishihara S, Takasu Y, Murakami Y. Catal Today, 2007, 120: 226 🚉
- [28] Yu J G, Xiang Q J, Zhou M H. Appl Catal B, 2009, 90: 595
- [29] Rehman S, Ullah R, Butt A M, Goher N D. J Hazard Mater, 2009, 170: 560
- [30] Yamashita H, Harada M, Misaka J, Takeuchi M, Ikeue K, Anpo M. J Photochem Photobiol A, 2002, 148: 257
- [31] Chen S F, Zhao W, Wei L, Zhang S J. Appl Surf Sci, 2008, 255: 2478
- [32] Zang X W, Lei L C. Mater Lett, 2008, 62: 895
- [33] Zhou M H, Yu J G, Liu S W, Zhai P C, Huang B B. Appl Catal B, 2009, 89: 160
- [34] Liu S W, Yu J G, Mann S. J Phys Chem C, 2009, 113: 10712
- [1] MENG Zada, ZHU Lei, CHOI Jong-geun, PARK Chong-yeon, OH Won-chun.Rare Earth Oxide-Treated Fullerene and Titania Composites with Enhanced Photocatalytic Activity for the Degradation of Methylene Blue[J]. 催化学报, 2011,32(9): 1457-1464
- Mouna AOUN, Aicha BENAMAR, Mohamed CHATER.Rh/ZnO-Al2O3 Catalysts for Selective Hydrogenation of Crotonaldehyde[J]. 催化学报, 2011,32(7): 1185-1190
- [3] K. KOCI, K. ZATLOUKALOVA, L. OBALOVA, S. KREJCIKOVA, Z. LACNY, L. CAPEK, A. HOSPODKOVA, O. SOLCOVA.Wavelength Effect on Photocatalytic Reduction of CO₂ by Ag/TiO₂ Catalyst[J]. 催化学报, 2011,32(5): 812-815
- S. SWETHA, R. GEETHA BALAKRISHNA.Preparation and Characterization of High Activity Zirconium-Doped Anatase Titania for Solar Photocatalytic Degradation of Ethidium Bromide[J]. 催化学报, 2011,32(5): 789-794
- [5] ZHANG Kan; MENG Zeda; OH Wonchun*.紫外光照射下 Fe-碳纳米管/TiO2 复合材料降解罗丹明 B[J]. 催化学报, 2010,31(7): 751-758
- Olga Z. DIDENKO; Gulnara R. KOSMAMBETOVA; Peter E. STRIZHAK*.纳米级ZnO/MgO固体的合成及其对CO的氧化活性[J]. 催化学报, 2008,29(11): 1079-1083
- G. R. MORADI*, F. KHOSRAVIAN, M. RAHMANZADEH.Effect of Partial Substitution of Ni by Cu in LaNiO₃ Perovskite Catalyst for Dry Methane Reforming[J]. 催化学报, 0,(): 0-

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