

Ti/SiO₂ 催化 H₂O₂ 氧化苯甲醇制苯甲醛反应机理的理论研究

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摘要 采用密度泛函理论 B3LYP 方法研究了非贵金属 Ti 催化 H₂O₂ 氧化苯甲醇为苯甲醛的反应机理. 考察了 6 条可能的反应途径, 优化得到了各个途径的过渡态和中间体结构, 计算了气相和液相中的反应势垒. 结果表明, 无催化剂时, H₂O₂ 氧化苯甲醇的反应途径具有非常高的反应势垒, 反应不能进行; 以 Ti/SiO₂ 为催化剂时, 标题反应在乙腈溶液中的反应势垒为 123.8 kJ/mol, 反应可在约 353 K 下发生. 结果还表明, 标题反应在极性较大的溶剂中有较高的反应势垒, 而在气相或者极性较小的溶剂中的反应势垒较低.

关键词: 密度泛函理论 苯甲醇 氧化 反应机理 苯甲醛 钛 二氧化硅 负载型催化剂

Abstract: The mechanism of benzyl alcohol oxidation to benzaldehyde catalyzed by Ti/SiO₂ in different solvents was studied by the density function theory B3LYP method. Total six possible reaction pathways and the corresponding optimal structures of reactants, transition states, intermediates, and products were located at the B3LYP/6-31G* level of theory. The reaction barriers were then calculated at the B3LYP/6-311++G(d,p) level of theory both in gas phase and in solvents. The calculation results show that a very high reaction barrier exists for the pathways where no catalyst is involved, suggesting that the oxidation cannot occur, while when Ti/SiO₂ is involved, a reaction barrier about 123.8 kJ/mol is found for that in the acetonitrile solvent, indicating that the oxidation of benzyl alcohol by H₂O₂ to benzyl aldehyde can happen at about 353 K in the acetonitrile solvent. It is also found that the title reaction has a relative lower barrier in non-polar or low-polar solvent but relative higher barrier in a high-polar solvent.

Keywords: density function theory, benzyl alcohol, oxidation, reaction mechanism, benzaldehyde, titanium, silica, supported catalyst

收稿日期: 2011-09-13; 出版日期: 2012-02-08

引用本文:
张岩, 黄翠英, 王俊芳等. Ti/SiO₂ 催化 H₂O₂ 氧化苯甲醇制苯甲醛反应机理的理论研究[J] 催化学报, 2012, V33(2): 360-366

ZHANG Yan, HUANG Cui-Ying, WANG Jun-Fang etc. Theoretical Study on the Mechanism of Benzyl Alcohol Oxidation to Benzaldehyde Catalyzed by Ti/SiO₂[J] Chinese Journal of Catalysis, 2012, V33(2): 360-366

链接本文:
<http://www.chxb.cn/CN/10.3724/SP.J.1088.2012.10924> 或 <http://www.chxb.cn/CN/Y2012/V33/I2/360>

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