



Light-induced redox cycling of iron in circumneutral lakes

Emmenegger, Lukas, René Schönenberger, Laura Sigg, Barbara Sulzberger

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ABSTRACT: The light-induced redox cycling of Fe^{II}/Fe^{III} was studied both in laboratory experiments and in the field in two circumneutral Swiss lakes: Greifensee, a eutrophic, natural water body, and Melchsee, an oligotrophic, artificial mountain lake. To determine Fe^{II} at the nanomolar level, an automated flow-injection analysis system was used. Irradiation by simulated sunlight leads to pH dependent (pH 6.9-9.1) steady-state Fe^{II} concentrations which are similar in samples from both lakes. However, the kinetics of Fe^{III} reduction and of apparent Fe^{II} oxidation are considerably faster in Melchsee. On the basis of experimental results and on modeling that uses literature values of known chemical transformation processes, we suggest that superoxide may be a key parameter for light-induced iron redox cycling in these lakes. Field measurements of [Fe^{II}] in Greifensee and Melchsee show a pronounced day/night cycle, with Fe^{II} concentrations of ~0.1-0.2 nM at night and up to 0.9 nM near the surface during the day (pH 8.0-8.5). Depth profiles of [Fe^{II}] have two maxima: one at the surface and the second one at a depth of 5-10 m. Empirical rates and measured physical parameters were included in a model to simulate [Fe^{II}] as a function of time and depth. The model results indicate that Fe^{II} at the surface of both lakes is produced by light-induced processes, whereas the deeper Fe^{II} maxima at depths with maximal chlorophyll a concentrations are probably due to a combination of biologically and photochemically induced processes.

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