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## 软阳离子钾改性蒙脱石矿物对硝基苯的强化吸附实验研究

## Experimental research on intensified adsorption of soft $K^+$ cation\_modified montmorillonite mineral for nitrobenzene

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中文关键词:蒙脱石 硝基苯 强化吸附 钾阳离子

英文关键词:montmorillonite nitrobenzene intensified adsorption potassium cation

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## 中文摘要:

考察软阳离子钾改性蒙脱石在K<sup>+</sup>存在的条件下对硝基苯的强化吸附行为,详细分析比较了Henry和Freundlich方程对硝基苯吸附等温线的拟合结果,同时还进行了吸附热力学研究。实验结果表明,软阳离子钾对蒙脱石吸附硝基苯具有强化作用;硝基苯在蒙脱石上的吸附符合Freundlich 和Henry方程,两模型方程线性拟合结果的相关系数均在0.96以上。热力学分析结果表明,硝基苯在蒙脱石上的吸附过程是一个自发的放热过程,说明蒙脱石对硝基苯的吸附主要由分配与EDA作用机制所控制。钾离子存在可为硝基苯吸附营造有利场所,不仅改善准入条件,同时增加某些吸附点位,从而促进并强化蒙脱石对硝基苯的吸附。

## 英文摘要:

The sorption behaviors of nitrobenzene (NB) onto the soft  $K^+$  cation modified montmorillonite (K-Mt) under the conditions of adding of  $K^+$  ion and varying reaction temperatures were investigated in this paper. The sorption kinetic experiments showed that the sorption equilibrium of NB could be reached within 8 h. A comparative analysis of the Henry and Freundlich equations was made to evaluate the sorption isotherms of NB onto the K-Mt, and some thermodynamic experiments were conducted to explore the sorption mechanism. The results show that soft cation of  $K^+$  can strengthen the sorption of K-Mt for NB. The adsorption isotherms of NB by the K-Mt can be fitted well with Henry and Freundlich equation ( $R_2$ >0.96). Moreover, the research results of thermodynamic sorption experiments demonstrate that the sorption of NB onto the K-Mt is a spontaneous exothermic process mainly controlled by both the hydrophobic bonding and  $n-\pi$  electron-donor-acceptor (EDA) interaction between clay mineral surface and nitrobenzene. The presence of  $K^+$  ion is likely to be a favorable factor due to increasing some sorption sites and improving the accessing condition for sorption, leading to the promotion of sorption for NB onto the K-Mt.

江强明, 张姚娜, 吴宏海, 2011, 软阳离子钾改性蒙脱石矿物对硝基苯的强化吸附实验研究[J]. 岩石矿物学杂志, 30(6):1074~1080.

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