Atmospheric Chemistry and Physics An Interactive Open Access Journal of the European Geosciences Union

Home

Online Library ACP

- Recent Final Revised Papers
- Volumes and Issues
- Special Issues
- Library Search
- Title and Author Search

Online Library ACPD

Alerts & RSS Feeds

General Information

Submission

Production

Subscription

Comment on a Paper





Volumes and Issues Contents of Issue 16 Special Issue Atmos. Chem. Phys., 7, 4171-4185, 2007 www.atmos-chem-phys.net/7/4171/2007/ © Author(s) 2007. This work is licensed under a Creative Commons License.

Observations of OH and HO₂ radicals in coastal Antarctica

W. J. Bloss^{1,*}, J. D. Lee^{1,**}, D. E. Heard¹, R. A. Salmon², S. J.-B. Bauguitte², H. K. Roscoe², and A. E. Jones² ¹School of Chemistry, University of Leeds, Woodhouse Lane, Leeds, LS2 9JT, UK ²British Antarctic Survey, Madingley Road, Cambridge, CB3 0ET, UK * now at: School of Geography, Earth & Environmental Sciences, University of Birmingham, Edgbaston, Birmingham, B15 2TT, UK ** now at: Department of Chemistry, University of York, Heslington, York, YO10 5DD, UK Abstract. OH and HO₂ radical concentrations have been measured in the boundary layer of coastal Antarctica for a six-week period during the austral summer of 2005. The measurements were performed at the British Antarctic Survey's Halley Research Station (75° 35' S, 26° 19' W), using the technique of on-resonance laser-induced fluorescence to detect OH, with HO₂ measured following chemical conversion through addition of NO. The

mean radical levels were 3.9×10^5 molecule cm⁻³ for OH, and 0.76 ppt for HO₂ (ppt denotes parts per trillion, by volume). Typical maximum (local

noontime) levels were 7.9×10^5 molecule cm⁻³ and 1.50 ppt for OH and HO₂ respectively. The main sources of HO_x were photolysis of O₃ and HCHO, with potentially important but uncertain contributions from HONO and higher aldehydes. Of the measured OH sinks, reaction with CO and CH₄ dominated, however comparison of the observed OH concentrations with those calculated via the steady state approximation indicated that additional co-reactants were likely to have been present. Elevated levels of NO_x resulting from snowpack photochemistry contributed to HO_x cycling and enhanced levels of OH, however the halogen oxides IO and BrO dominated the $CH_3O_2 - HO_2 - OH$ conversion in this environment, with associated ozone destruction.

■ Final Revised Paper (PDF, 1169 KB) ■ Discussion Paper (ACPD)

Citation: Bloss, W. J., Lee, J. D., Heard, D. E., Salmon, R. A., Bauguitte, S. J.-B., Roscoe, H. K., and Jones, A. E.: Observations of OH and HO₂ radicals in coastal Antarctica, Atmos. Chem. Phys., 7, 4171-4185, 2007. Dibtex Distance EndNote Reference Manager

| EGU Journals | Contact



Library Search Author Search

- Sister Journals AMT & GMD
- Financial Support for Authors
- Journal Impact Factor
- Public Relations & **Background Information**

Recent Papers

01 | ACP, 17 Dec 2008: Characterizing ozone production and response under different meteorological conditions in Mexico City

02 | ACP, 17 Dec 2008: Significant impact of the East Asia monsoon on ozone seasonal behavior in the boundary layer of Eastern China and the west Pacific region

03 | ACP, 17 Dec 2008: Carbonyl sulfide in air extracted from a South Pole ice core: a 2000 year record