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## 南京一次强酸雾的化学特征和成因分析(PDF)

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Title: Chemical features and cause analysis of a strong acid fog event in Nanjing

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关键词: 南京; 平流雾; 化学特征; 强酸雾

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摘要: 利用2009年12月1-2日一次平流雾过程中采集到的10个雾水样本资料,结合微物理结构、边界层结构和污染气体等资料,分析了其雾水化学特征,并与2006和2007年冬季观测分析的平流辐射雾和辐射雾雾水化学特征进行了比较.结果表明:平流雾过程中阴离子 $\text{SO}_4^{2-}$ 浓度最高,其次为 $\text{NO}_3^-$ ,阳离子中 $\text{K}^+$ 最高;南京不同类型的雾水酸化类型均属于硫酸型,这次平流雾雾水酸性强, $\text{SO}_4^{2-}$ 和 $\text{NO}_3^-$ 的浓度均高于平流辐射雾和辐射雾,尤其是 $\text{NO}_3^-$ 浓度成倍增加;平流雾中雾滴平均直径、含水量与离子浓度呈反相关.对平流雾酸度来源进行了探讨发现,雾水酸度强的主要原因一是致酸物 $\text{SO}_4^{2-}$ 和 $\text{NO}_3^-$ 浓度高,其中 $\text{NO}_3^-$ 成倍增加;二是缓冲物 $\text{Ca}^{2+}$ 浓度特别低.局地排放和平流输送造成高浓度的 $\text{SO}_2$ 和 $\text{NO}_2$ 是导致强酸雾的重要原因,而平流和垂直输送使地面层的 $\text{Ca}^{2+}$ 浓度减小,从而使得雾水中碱性物质减少,进而增强了雾水酸性.

Abstract: Using ten samples of the fog water collected in an advection fog process on December 1-2, 2009 and the corresponding fog microphysical structure, boundary layer structure and air pollution data, the chemical characteristics of the advection fog waters and the cause of acid fog formation were analyzed and compared with that of the advection-radiation fog and the radiation fog

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occurred in winters of 2006 and 2007. Results show that the principal negative ion in the advection fog waters is  $\text{SO}_4^{2-}$  whose concentration is higher than  $\text{NO}_3^-$  and other negative ions, while the main positive ion is  $\text{K}^+$ . All types of Nanjing fogs are sulfuric acid. Comparing with the advection-radiation fog and the radiation fog, all of the advection fog water samples are acidic and have stronger acidity. The concentrations of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  in the advection fog are higher than those of the other two fog types, especially the concentration of  $\text{NO}_3^-$ . The average diameter of fog droplets and fog water concentration are anti-correlated to ions concentration. The source of the acidity is mainly due to high concentration of acidic  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$ , in which  $\text{NO}_3^-$  doubled, while the concentration of the  $\text{Ca}^{2+}$  is particularly low. The high concentrations of  $\text{SO}_2$  and  $\text{NO}_2$  caused by local exhaustion and advection transport are the important causes of strong acid fog. Meanwhile advection and vertical transmission decrease the concentration of  $\text{Ca}^{2+}$  on the near-surface floor, which reduce the alkaline substances in fog water, and thus enhance the acidity of fog water.

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#### 参考文献/REFERENCES

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