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OPEN@ACCESS The Factors on Removal of Zinc Cation from Aqueous Solution by			NR Subscription	
Bentonite PDF (Size: 339KB) PP. 107-113 DOI: 10.4236/nr.2011.22015 Author(s) Shuli Ding, Juanjuan Shen, Bohui Xu, Qinfu Liu, Yuzhuang Sun		Most popular papers in NR About NR News		
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		The re	RACT moving zinc cation from aqueous solution by Ca-bentonite and Na-exchanged bentonite was studied.	Recommend to Peers
The factors such as the initial concentration of Zn2+, the liquid-to-solid ratio, pH, adsorption time, stirring speed, coexisting ions, temperature and bentonite particle size were investigated. The results show that the adsorption process of bentonite accorded with the Freundlich isotherm model, the removal of Zn2+ by Ca-bentonite and Na-exchanged bentonite reached equilibrium in 2 h, and adsorption of Na- bentonite was superior to Ca-bentonite. The adsorption rate of zinc increased with increasing pH, temperature, stirring speed, time span and with decreasing bentonite particle, the initial concentration of Zn2+ and the liquid-to-			Recommend to Library	
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solid ratio. In mixed solution which contains Pb2+and Cr6+, Pb2+ has no influ- ence on the removal of $Zn2+$ by both the bentonites while Cr6+can decease it.		Downloads:	62,915	
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