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基金项目:

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中文摘要:

采用溶胶-凝胶方法制备了正极材料 $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$ 。XRD、XPS测试结果表明:  $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$ 中阳离子排列有序度较高, 层状结构明显; Co、Mn分别以+3、+4价形式存在, Ni以+2、+3价形式存在, 且 $\text{Ni}^{2+}$ 与 $\text{Ni}^{3+}$ 的含量之比约为1:1。SEM测试结果表明: 正极材料 $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$ 结晶粒径较均匀。充放电测试结果表明: 与 $\text{LiCoO}_2$ 相比, 尽管 $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$ 的放电电压平台较低, 但放电容量较高; 在恒流充电模式下, 当充电截止电压由4.35 V升高至4.75 V时, 首次放电容量由179  $\text{mAh} \cdot \text{g}^{-1}$ 增至201  $\text{mAh} \cdot \text{g}^{-1}$ , 50次循环后, 容量保持率由74.95%增至78.48%; 在先恒流再恒压的充电模式下, 电池首次放电容量为212  $\text{mAh} \cdot \text{g}^{-1}$ , 50次循环后, 容量保持率提高到87.71%。循环伏安测试表明: 在2.80~4.80 V扫描范围内, 该正极材料发生 $\text{Ni}^{2+}/\text{Ni}^{3+}$ ,  $\text{Co}^{3+}/\text{Co}^{4+}$ 两对电化学反应。EIS测试表明: 随着充电截止电压的增大, 该正极材料的传荷电阻变小。

英文摘要:

Cathode materials of  $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$  was synthesized by Sol-gel method in this study. XRD and XPS results reveal that the obtained  $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$  exhibits a typical layered structure composed of highly ordered  $\text{Co}^{3+}$ ,  $\text{Mn}^{4+}$ ,  $\text{Ni}^{2+}$  and  $\text{Ni}^{3+}$ ; and the atom ratio of  $\text{Ni}^{2+}$  to  $\text{Ni}^{3+}$  is 1:1. A morphology of homogeneously distributed and well crystallized is observed by SEM. Compared with  $\text{LiCoO}_2$ , the resulted materials possess larger capacity and lower voltage plateau. It is also found that the initial discharge capacity and capacity retention of 50<sup>th</sup> cycle of  $\text{LiNi}_{0.5}\text{Co}_{0.25}\text{Mn}_{0.25}\text{O}_2$  increase from 179  $\text{mAh} \cdot \text{g}^{-1}$  to 201  $\text{mAh} \cdot \text{g}^{-1}$ , from 74.95% to 78.48%, respectively, when the cutoff voltage is changed from 4.35 V to 4.75 V. Under the discharge-charge model of initial constant current and subsequent constant voltage, an initial discharge capacity of 212  $\text{mAh} \cdot \text{g}^{-1}$  can be achieved and the material keeps 87.71% of its initial capacity after 50<sup>th</sup> cycle. CV and EIS tests show that two redox couples of  $\text{Ni}^{2+}/\text{Ni}^{3+}$  and  $\text{Co}^{3+}/\text{Co}^{4+}$  are observed in the potential range of 2.80~4.80 V, and charge transfer resistance of the cathode decreases as the charge cutoff voltage increased.

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