化学

N-甲基甲异羟肟酸的合成及其与Np(IV)、Pu(IV)配合物稳定常数的测定

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收稿日期 修回日期 2007-12-28 网络版发布日期:

摘要 以甲酸乙酯和N-甲基盐酸羟胺为主要原料,在乙醇-水体系中合成N-甲基甲异羟肟酸(NMFHA),并通过元素分析、红外光谱、质谱分析和核磁共振波谱等方法对其结构进行表征。TTA萃取法测定结果表明,在 $1.0~\text{mol/L}~\text{HNO}_3$ 体系中,Np(IV)、Pu(IV)与NMFHA形成稳定的1:2的配合物,其累积稳定常数分别为: β_1 (Np (IV))= 8.83×10^9 , β_2 (Np(IV))= 1.01×10^{19} ; β_1 (Pu(IV))= 7.78×10^{10} , β_2 (Pu(IV))= 5.80×10^{19} 。

 关键词
 N-甲基甲异羟肟酸
 合成; Np(IV)
 Pu(IV)
 配合
 稳定常数

 分类号

Synthesis of N-Methylformohydroxamic Acid and Determination

of Stability Constants of Its Complexes With Pu(IV) and Np(IV)

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Abstract The title compound N-methylformohydroxamic acid (NMFHA) was synthesized by the reaction of N-methyl hydroxylamine hydrochloride with ethyl formate in alcohol medium. It was also characterized by element analysis, IR spectrum, mass spectrum and NMR spectrum. In 1.0 mol/L HNO3 solution, stable 1:2 complexes of NMFHA complexes with Np(IV) or Pu (IV) were determined by TTA extraction method, and the stability constants are $\beta_1(\text{Np}(\text{IV}))$ =

8.83×10⁻⁹, $\beta_2(\text{Np}(\text{IV})) = 1.01 \times 10^{19}$, and $\beta_1(\text{Pu}(\text{IV})) = 7.78 \times 10^{10}$, $\beta_2(\text{Pu}(\text{IV})) = 5.80 \times 10^{19}$, respectively.

Key words N-methylformohydroxamic acid synthesis Np(IV) Pu (IV) complex stability constant

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