

[本期目录](#) | [下期目录](#) | [过刊浏览](#) | [高级检索](#)[\[打印本页\]](#) [\[关闭\]](#)**论文****PAMP SLi/P(MMA-VAc)为基体的聚合物电解质**牛丽丹^{1,2}, 傅相锴^{1,2,3*}, 刘素娟^{1,2}, 杜秋亮^{1,2}

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摘要:

将聚(2-丙烯酰胺-2-甲基丙磺酸锂)(PAMP SLi)和聚(甲基丙烯酸甲酯-醋酸乙烯酯)[P(MMA-VAc)]与LiClO₄共混, 制备了复合聚合物电解质膜。用FTIR, TG, XRD, DMA, SEM及电化学交流阻抗和机械性能测试对聚合物及其电解质膜的结构和性能进行了表征。结果表明, PAMP SLi与P(MMA-VAc)共混后结晶状态发生变化, 交联网络变得密实, 提高了热稳定性和拉伸强度, 聚合物电解质膜含有较多微孔结构, 孔径为5~10 μm; 在20 °C时, 当n(MMA):n(VAc)=2:8, m(PAMP SLi):m[P(MMA-VAc)]=5:95, m(LiClO₄):m(copolymer)=15:85时, 聚合物电解质膜电导率可达到 1.68×10^{-3} S/cm, 且电导率未出现随LiClO₄含量的进一步增加而下降的现象。将此电解质用于全固态电致变色显示器件表现出了优良的性能。对加入PAMP SLi后聚合物电解质膜电导率和热稳定性提高的原因进行了初步探讨。

关键词: 聚合物电解质; 聚(甲基丙烯酸甲酯-醋酸乙烯酯); 聚(2-丙烯酰胺-2-甲基丙磺酸); 离子电导率; 电致变色器件

Polymer Electrolytes Based on PAMP SLi/P(MMA-VAc)NIU Li-Dan^{1,2}, FU Xiang-Kai^{1,2,3*}, LIU Su-Juan^{1,2}, DU Qiu-Liang^{1,2}

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Abstract:

Poly-2-acrylamido-2-methylpropanesulfonic acid lithium(PAMP SLi) and poly-vinylacetate-methyl methacrylate[P(MMA-VAc)] were synthesized. And the polymer electrolytes were prepared by blending PAMP SLi, P(MMA-VAc) and LiClO₄. Performances of the copolymer and polymer electrolyte were studied by FTIR, TG, XRD, DMA, SEM, alternating current(AC) impedance and mechanical testing. The results showed that: the crystallization state of the polymer electrolyte changed after blending PAMP SLi with P(MMA-VAc); the cross-linked network changed dense; the thermal stability and tensile strength were improved; the polymer electrolyte membrane had a great number of micro-pores with dimension of 5—10 μm; the ionic conductivity value of the copolymer membrane reached 1.68×10^{-3} S/cm when n(MMA):n(VAc)=2:8, m(PAMP SLi):m[P(MMA-VAc)]=5:95 and m(LiClO₄):m(copolymer)=15:85 at 20 °C, which did not appear drop by the increase of the LiClO₄. And its excellent performance promised the usage of the polymer electrolytes as the ionic conductor material in electrochromic devices(ECD). The mechanism of the conductivity and the thermal stability of polymer electrolyte membrane with the increase of PAMP SLi was preliminary studied.

Keywords: Polymer electrolyte; Poly-2-acrylamido-2-methylpropanesulfonic[P(MMA-VAc)]; Poly-vinylacetate-methyl methacrylate acid(PAMP SLi); Ionic conductivity; Electrochromic device

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参考文献:

- [1]Fenton D. E., Parker J. M., Wright P. V.. Polymer[J], 1973, 14(11): 589—591
[2]Armand M. B., Chabagno J. B., Duclot M. J.. Fast Ion Transport in Solids[M], Amsterdam: North Holland Publishing Co., 1979: 131—136
[3]GU Ning-Yu(古宁宇), QIAN Xin-Ming(钱新明). Chem. J. Chinese Universities(高等学校化学学报)[J], 2001, 22(8): 1403—1405
[4]QI Li(齐力), DONG Shao-Jun(董绍俊). Chem. J. Chinese Universities(高等学校化学学报)[J], 2005, 26(11): 2165—2169
[5]JIANG Jing(蒋晶), GAO De-Shu(高德淑), LI Chao-Hui(李朝辉), et al.. Chem. J. Chinese Universities(高等学校化学学报)[J], 2006, 27(7): 1319—1322
[6]REN Xü-Mei(任旭梅), GU Hui(顾辉). Chem. J. Chinese Universities(高等学校化学学报)[J], 2002, 23(7): 1383—1385
[7]Travas-Sejdic J., Easteal A.. J. Appl. Polym. Sci.[J], 2000, 75: 619—628
[8]Shim W. S., Lee D. S.. J. Appl. Polym. Sci.[J], 1999, 74: 311—321
[9]Durmaz S., Okay O.. Polymer[J], 2000, 41(10): 3693—3704
[10]Travas-Sejdic J., Steiner R., Silvestro J. D., et al.. Electrochim. Acta[J], 2001, 46(10/11): 1461—1466
[11]Harris C. S., Rukavina T. G.. Electrochim. Acta[J], 1995, 40(13/14): 2315—2320
[12]Sun J., MacFarlane D. R., Forsyth M.. Solid State Ionics[J], 2002, 147(3/4): 333—339
[13]Karlsson L. E., Wesslen B., Jannasch P.. Electrochim. Acta[J], 2002, 47: 3269—3275
[14]Karlsson L. E., Jannasch P., Wesslen B.. Macromol. Chem. Phys.[J], 2002, 203: 686—694
[15]Walker C. R.. J. Power Sources[J], 2002, 110: 144
[16]Sakai Y., Matsuguchi M., Yonesato N.. Electrochim. Acta[J], 2001, 46(10/11): 1509—1514
[17]Shi W., Watson C. J., Palmer C. P.. J. Chromatogr. A[J], 2001, 905(1/2): 281—290
[18]YANG Dao-Jun(杨道均), FU Xiang-Kai(傅相锴), JIANG Qing-Long(蒋庆龙), et al.. Chem. J. Chinese Universities(高等学校化学学报)[J], 2007, 28(9): 1781—1786
[19]TAN Ying(谭颖), LI Zhi-Qiang(李志强), ZHANG Zhi-Cheng(张志成), et al.. Advances in Fine Petrochemicals(精细石油化工进展)[J], 2003, 4(3): 47—49
[20]Jiang Qing-Long, Fu Xiang-Kai. Chinese Chemical Letters[J], 2006, 17: 1447—1449
[21]HONG Zhang-Chuan(洪璋传). China Synthetic Fiber Industry(合成纤维工业)[J], 2001, 24(2): 38—39
[22]Baskaran R., Selvasekarpandian S., Kuwata N., et al.. Solid State Ionics[J], 2006, 177: 2679—2682

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