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N,N-二(4-(6-2,2'-联吡啶)-苯甲基)-2-(氨基)吡啶的合成、结构和光谱性质

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Synthesis, structure, and spectra studies of N,N-bis(4-(6-2,2'-bipyridine)-phenyl-methyl)-2-(aminomethyl)pyridine

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- 摘要
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全文: PDF (1380 KB) HTML (1 KB) 输出: BibTeX | EndNote (RIS) 背景资料

摘要 合成了一种多齿配体N,N-二(4-(6-2,2'-联吡啶)-苯甲基)-2-(氨基)吡啶(C₄₀H₃₂N₆),并用MS,EA,¹H NMR进行结构表征,测试了吸收光谱(UV)和荧光光谱(FL).随HBF₄,乙酸锌的加入,荧光光谱明显红移;随Cu²⁺浓度增加吸收峰强度逐渐增加,但发射峰逐渐消失,成为激发态荧光体的猝灭剂.测量并计算了摩尔消光系数和量子产率,并用密度泛函理论计算预测了其稳定的空间构型.

关键词: 联吡啶 多齿配体 结构特征 光谱性质 密度泛函

Abstract: Polydentate ligand N,N-bis(4-(6-(pyridine-2-yl)pyridin-2-yl)benzyl)(pyridin-2-yl)methanamine has been synthesized and characterized by MS,EA, and ¹H NMR. Photophysical behaviors of the ligand were investigated by UV-Vis absorption and fluorescence spectrometry. Addition of HBF₄ and zinc acetate to the solution of ligand all resulted in an obvious red-shift of fluorescence. Introduction of Cu²⁺ to the methanol solution of ligand caused a progressive increasing of absorption intensity and a gradual disappearing of emission bands, which indicated that Cu²⁺ was a quencher of the excited-state fluorescence. Molar extinction coefficient and quantum yield were measured and calculated. The stable space configuration of the ligand was predicted by density functional theory calculations.

Key words:

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