研究论文

氢钼青铜对铂催化氧还原反应的促进作用

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摘要 采用循环伏安法在玻碳电极上和硫酸溶液中电沉积制备出铂催化剂(Pt)及铂-氢钼青铜复合催化剂(Pt- H_x MoO $_3$),用旋转圆盘电极研究并比较了它们对硫酸溶液中氧还原反应的催化活性。研究结果表明, H_x MoO $_3$ 能明显地提高Pt对氧还原反应的电催化活性。通过对静态电极上氧还原的峰电流与扫描速度的关系以及旋转圆盘电极上氧还原电流与旋转速度的关系的分析发现, H_x MoO $_3$ 提高了铂电极氧还原反应电荷传递步骤的传递系数,因此加快了氧还原的动力学过程。

关键词 氧还原反应 催化 铂 氢钼青铜

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Catalytic Activity Improvement of Platinum Toward Oxyge n Reduction Reaction Promoted by Hydrogen Molybdenum Bronze

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Abstract Platinum catalyst(Pt) and its composite catalyst with hydrogen molybdenum bronze(P t-H $_x$ MoO $_3$) were prepared on a glass carbon electrode with cyclic voltammetry, in 0.5 mol/L H $_2$ SO $_4$ solution containing 5 mmol/L H $_2$ PtCl $_6$ and 5 mmol/L H $_2$ PtCl $_6$ +5 mmol/L Na $_2$ MoO $_4$, respectively. Their catalytic activity toward the oxygen reduction reaction(ORR) in 0.5 mol/L H $_2$ SO $_4$ solution saturated with oxygen was studied with rotating disk electrode. The result shows that the catalytic activity of platinum toward ORR can be improved by H $_x$ MoO $_3$. The catalytic mechanism was understood, based on the analysis on the relationship between the peak current of ORR and scan rate on the quiescent electrodes and the relationship between the current of ORR and rotating speed on the rotating electrodes. The transfer coefficient of charge transfer step for ORR on Pt is improved by H $_x$ MoO $_3$, thus the kinetics of ORR is enhanced significantly. The transfer coefficient on Pt-H $_x$ MoO $_3$ for ORR is 1.2 times that on Pt, and the kinetic current on Pt-H $_x$ MoO $_3$ at 0.3 V(vs. SCE) is 5 times that on Pt.

Key words Oxygen reduction reaction Catalysis Platinum Hydrogen molybdenum bronze

DOI:

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