

[前一个](#)[后一个](#)[本期目录](#) | [下期目录](#) | [过刊浏览](#) | [高级检索](#)[\[打印本页\]](#) [\[关闭\]](#)**研究报告**

电化学方法研究锈层覆盖下碳钢的腐蚀规律

邹妍¹,王佳^{1,2},郑莹莹¹

1. 中国海洋大学化学化工学院 青岛 266100

2. 中国科学院金属研究所金属腐蚀与防护国家重点实验室 沈阳 110016

摘要: 分别采用失重法与电化学方法(包括极化曲线、线性极化法和电化学阻抗技术)研究了碳钢在海水中浸泡48周的腐蚀规律。结果发现,短期浸泡在海水中的碳钢,表面生成的腐蚀产物薄且疏松,长期浸泡后,锈层逐渐分为两层:外锈层薄且疏松,内锈层厚且致密。短期浸泡,失重法与电化学方法得到的腐蚀规律一致,腐蚀速度逐渐减小并且数值相近,此时电化学方法可以用来准确的估算碳钢的腐蚀速度。长期浸泡,电化学方法测定的腐蚀速度转为逐渐增大,偏离了失重结果,并且锈层越厚偏差越大,此时电化学方法会过高估算腐蚀速度。

关键词: 碳钢 海水 腐蚀形貌 复杂体系 偏差**ELECTROCHEMICAL STUDY ON CORROSION OF RUSTED CARBON STEEL**ZOU Yan¹, WANG Jia^{1,2}, ZHENG Yingying¹

1. College of Chemistry and Chemical Engineering, Ocean University of China, Qingdao 266100

2. State Key Laboratory for Corrosion and Protection, Institute of Metal Research, Chinese Academy of Sciences, Shenyang 110016

Abstract: Mass loss measurement and various electrochemical methods, including polarization curves, linear polarization resistance measurement (LPR) and electrochemical impedance spectra technique (EIS) were employed to evaluate the corrosion of rusted carbon steel immersed in seawater for 48 weeks. Results indicated that the initial corrosion product formed on the carbon steel was thin and loose. With prolonged immersion, the rust layer could be divided into two layers: the outer layer was thin and loose, while the inner layer was thick and compact. The corrosion rates calculated by electrochemistry measurement, were consistent with the mass loss. The electrochemical methods could be used to accurately estimate corrosion rate at initial period of immersion. After long-term immersion, the electrochemical corrosion rate turned to increase and deviated from mass loss gradually.

Keywords: carbon steel seawater corrosion morphology complex system deviation

收稿日期 2010-03-05 修回日期 2010-12-13 网络版发布日期 2011-04-14

DOI:**基金项目:**

国家自然科学基金项目(50971118)资助

通讯作者: 王佳**作者简介:** 邹妍,女,1980年生,博士生,研究方向为金属腐蚀与防护**通讯作者E-mail:** jwang@ouc.edu.cn**扩展功能****本文信息**

▶ Supporting info

▶ PDF(2345KB)

▶ [HTML] 下载

▶ 参考文献[PDF]

▶ 参考文献

服务与反馈

▶ 把本文推荐给朋友

▶ 加入我的书架

▶ 加入引用管理器

▶ 引用本文

▶ Email Alert

▶ 文章反馈

▶ 浏览反馈信息

本文关键词相关文章

▶ 碳钢

▶ 海水

▶ 腐蚀形貌

▶ 复杂体系

▶ 偏差

本文作者相关文章

▶ 邹妍

▶ 郑莹莹

▶ 王佳

PubMed

▶ Article by Zou,y

▶ Article by Zheng,Y.Y

▶ Article by Yu,j

参考文献:

- [1] Refait P, Memet J B, Bon C, et al. Formation of the Fe(II)--Fe(III) hydroxysulphate green rust during marine corrosion of steel [J]. Corros.Sci., 2003, 45(4): 833-845 
- [2] Dunnwald J, Otto A. An investigation of phase transitions in rust layers using raman spectroscopy [J]. Corros. Sci., 1989,29(9): 1167-1176 
- [3] Garcí K E, Morales A L, Barrero C A, et al. New contributions to the understanding of rust layer formation in steels exposed to a total immersion test [J]. Corros. Sci.,2006, 48(9): 2813-2842 
- [4] Xia L T, Huang G Q, Ding L P. Sea water corrosion properties of carbon steel and low alloy steel [J]. Res. Stud. Foundry Equip.,2002, 4: 14-17
- [5] Huang G Q. Corrosion behavior of carbon steels immersed in sea areas of China [J]. Corros. Sci. Prot. Technol.,2001, 13 (2): 81-88
黄桂桥. 碳钢在我国不同海域的海水腐蚀行为 [J]. 腐蚀科学与防护技术, 2001, 13(2): 81-88) [浏览](#)
- [6] Sawant S S, Wagh A B. Corrosion behaviour of metals and alloys in the waters of the Arabian Sea [J]. Corros. Prev.Contr., 1990, 37(6): 154-157
- [7] Melchers R E. Effect of small compositional changes on marine immersion corrosion of low alloy steels [J]. Corros. Sci., 2004,46(7): 1669-1691 
- [8] Melchers R E, Jeffrey R. Early corrosion of mild steel in seawater [J]. Corros. Sci., 2005, 47(7): 1678-1693 
- [9] Melchers R E. Mathematical modelling of the diffusion controlled phase in marine immersion corrosion of mild steel [J]. Corros. Sci., 2003, 45(5): 923-940 
- [10] Melchers R E, Wells T. Models for the anaerobic phases of marine immersion corrosion [J]. Corros. Sci., 2006,48 (7): 1791-1811 
- [11] Panda B, Balasubramaniam R, Dwivedi G. On the corrosion behaviour of novel high carbon rail steels in simulated cyclic wet-dry salt fog conditions [J]. Corros. Sci.,2008, 50(6): 1684-1692 
- [12] Bousselmi L, Fiaud C, Tribollets B, et al. The Characterisation of the coated layer at the interface carbon steel-natural salt water by impedance spectroscopy [J].Corros. Sci., 1997, 39(9): 1711-1724 
- [13] Yadav A P, Nishikata A, Tsuru T. Electrochemical impedance study on galvanized steel corrosion under cyclic wet--dry conditions--influence of time of wetness [J].Corros. Sci., 2004, 46(1): 169-181 
- [14] Bousselmi L, Fiaud C, Tribollets B, et al. Impedance spectroscopic study of a steel electrode in condition of scaling and corrosion: Interphase model [J]. Electrochim.Acta, 1999, 44(24): 4357-4363 
- [15] Ma Y T, Li Y, Wang F H. The effect of β -FeOOH on the corrosion behavior of low carbon steel exposed in tropic marine environment [J]. Mater. Chem. Phys.,2008, 112(3): 844-852 
- [16] Song G L. Theoretical analysis of the measurementof polarisation resistance in reinforced concrete [J].Cem. Concr. Comp., 2000, 22(6): 407-415\par 

本刊中的类似文章

1. 萧彧星, 吴光海, 孙宁, 王树宗, 萧以德.BP神经网络在碳钢和低合金钢大气腐蚀数据预测中的应用[J]. 中国腐蚀与防护学报, 2011,23(2): 171-174
2. 向红亮,刘东,何福善,黄利光.固溶温度对超级双相不锈钢在海水中耐蚀性的影响[J]. 中国腐蚀与防护学报, 2011,23(1): 25-28
3. 郑莹莹,邹妍,王佳.海洋环境中锈层下碳钢腐蚀行为的研究进展[J]. 中国腐蚀与防护学报, 2011,23(1): 93-98
4. 李发根,魏斌,邵晓东,蔡锐.双金属复合管技术经济性分析[J]. 中国腐蚀与防护学报, 2011,23(1): 86-88
5. 唐俊文,邵亚薇,郭金彪, 张涛, 孟国哲, 王福会.碳钢在90℃、 $H_2S-HCl-H_2O$ 环境下的腐蚀行为 I - H_2S 浓度对碳钢腐蚀行为的影响[J]. 中国腐蚀与防护学报, 2011,31(1): 28-33
6. 唐俊文,邵亚薇,陈阵, 张涛, 孟国哲, 王福会.碳钢在90℃、 $H_2S-HCl-H_2O$ 环境下的腐蚀行为 II - H_2S 溶液中HCl浓度对碳钢腐蚀行为的影响[J]. 中国腐蚀与防护学报, 2011,31(1): 34-39
7. 曹国良 李国明 陈珊 常万顺 陈学群.海水飞溅区Ni-Cu-P钢的锈层和耐点蚀性能研究[J]. 中国腐蚀与防护学报, 2011,47(02): 145-151

8. 阳靖峰 董俊华 柯伟 陈楠.硼酸缓冲溶液中pH值和腐蚀产物对低碳钢活化/钝化敏感性的影响[J].中国腐蚀与防护学报, 2011,47(02): 152-156
9. 钮晓博, 王毅, 钟庆东, 王超, 盛敏奇, 周琼宇.碳钢/环氧涂层在5%H₂SO₄溶液中的半导体行为[J].中国腐蚀与防护学报, 2010,22(6): 499-503
10. 王毅, 盛敏奇, 钟庆东, 钮晓博, 林海, 范成诚.表面沉积镍的低碳钢在3.5%NaCl溶液中的电化学行为[J].中国腐蚀与防护学报, 2010,22(6): 490-494

Copyright by 中国腐蚀与防护学报