#### 研究论文

层状锰酸锂衍生物的阳离子价态与锂电池的电化学性能研究

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收稿日期 2004-12-8 修回日期 2005-3-4 网络版发布日期 接受日期

摘要 采用溶胶-凝胶法,通过锂盐、镍盐、钴盐与锰盐生成锂镍钴锰氧化合物的前驱体,

随后采用高温固相法合成了 $\text{Li}(\text{Mn}_{1/3}\text{Co}_{1/3}\text{Ni}_{1/3})\text{O}_2$ ·借助于X射线光电子能谱(XPS)、X射线衍射(XRD)、循环伏安(CV)及充放电测试等现代测试手段研究了材料的晶型结构、离子价态及电化学性能. 前驱体经950  $^{\circ}$ 

煅烧可获得晶体结构完整、晶胞参数为a=0.2864 nm, c=1.4235 nm的六方层状Li(Mn $_{1/3}$ Co $_{1/3}$ Ni $_{1/3}$ )O<sub>2</sub>化合物;

XPS结果表明Li(Mn<sub>1/3</sub>Co<sub>1/3</sub>Ni<sub>1/3</sub>)O<sub>2</sub>化合物表面上的Mn, Ni和Co分别以Mn<sup>4+</sup>, Ni<sup>2+</sup>和Co<sup>3+</sup>存在; 材料的高温放电比容量比室温要高, 在55 ℃下, 在2.5~4.6 V电压范围内, 电流密度为28 mA/g时材料首次放电容量 195 mAh/g, 循环10次后容量保持在170 mAh/g; 循环伏安曲线上3.7 V和4.4 V的氧化还原过程对应于Ni<sup>2+/4+</sup>和Co<sup>3+/4+</sup>氧化还原电对的反应.

关键词  $\underline{\text{Li}(\text{Mn}_{1/3}\text{Co}_{1/3}\text{Ni}_{1/3})\text{O}_2}$  <u>高温固相合成</u> <u>晶体结构</u> <u>表面分析</u> <u>电化学性能</u>

分类号

# Cationic Ion Value State of Layered Lithium Manganate Derivative and Electrochemical Properties of Lithium Battery

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**Abstract** Abstract The cathode material for lithium batteries  $\text{Li}(\text{Mn}_{1/3}\text{Co}_{1/3}\text{Ni}_{1/3})\text{O}_2$  has been synthesized by sol-gel technique followed by high temperature solid state reaction. The crystalline structure, ionic value and electrochemical properties of the material were investigated via X-ray photoelectron spectra, X-ray diffraction, cyclic voltammetry and charge-discharge cycling. The hexagonal lattice parameters of the compound by calcining the precursor at 950 °C are: a = 0.2864 nm, c = 1.4235 nm. XPS studies indicate that the oxidation states of manganese, nickel and cobalt in the Li  $(\text{Mn}_{1/3}\text{Co}_{1/3}\text{Ni}_{1/3})\text{O}_2$  surface region are largely 4+, 2+ and 3+ respectively with small content of  $\text{Ni}^{2+}$  ion. Initial discharge capacity of 195 mAh/g at 55 °C was obtained in the range  $2.5\sim4.6 \text{ V}$  and at a specific current of 28 mA/g and under the same conditions 170 mAh/g was retained at the end of the 10th cycles. From the voltage profile and cyclic voltammetry, the redox processes occurring at  $\sim3.7 \text{ and} \sim4.4 \text{ V}$  were assigned to the  $\text{Ni}^{2+/4+}$  and  $\text{Co}^{3+/4+}$  couples, respectively.

**Key words** <u>Li(Mn<sub>1/3</sub>Co<sub>1/3</sub>Ni<sub>1/3</sub>)O<sub>2</sub></u> <u>high temperature solid state reaction</u> <u>crystal structure</u> <u>surface analysis</u> electrochemical property

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