水-乙醇混合溶剂中弱酸电离的溶剂效应

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摘要 本文用LKB2277Bioactivity monitor微量流动热量计测定卤代乙酸. 脂肪酸在水-乙醇混合溶剂中的标准电离焓,计算了热力学函数.结合文献数据,

计算各酸根离子从水迁移到混合溶剂的迁移自由能,并讨论溶剂效应对弱酸电离的影响1.

#### Solvent effects on the dissociation of weak acids in water-etoh mixtures at 298.15 K

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**Abstract** The ionization enthalpies of acetic acid, chloroacetic acid, bromoacetic acid, iodoacetic acid, propionic acid, and n-butyric acid in water-EtOH mixtures were measured at 298.15K with an LKB 2277 bioactivity monitor. The values of free energy and entropy of ionization were then calculated On the base of experimental data, the free energy changes on transfer of carboxylate ions from water to water-EtOH mixtures were obtained. The solvent effect on the ionization of the six weak acids was discussed.

**Key words** ETHANOL WATER FATTY ACID THERMODYNAMIC FUNCTION SOLVENT EFFECT MIX SOLVENT .

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#### 扩展功能

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