

铁酸镍用于热化学循环反应CO₂分解制CO的研究

 康敏^{1,2}, 张军³, 赵宁¹, 魏伟^{1,4}, 孙予罕^{1,3}

1. 中国科学院山西煤炭化学研究所 煤转化国家重点实验室, 山西 太原 030001;
2. 中国科学院大学, 北京 100049;
3. 中国科学院上海高等研究院 低碳转化中心, 上海 201210;
4. 中国科学院上海高等研究院 温室气体与环境工程研究中心, 上海 201210

CO production via thermochemical CO₂ splitting over Ni ferrite-based catalysts

 KANG Min^{1,2}, ZHANG Jun³, ZHAO Ning¹, WEI Wei^{1,4}, SUN Yu-han^{1,3}

1. State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese Academy of Sciences, Taiyuan 030001, China;
2. University of Chinese Academy of Sciences, Beijing 100049, China;
3. Low Carbon Conversion Center, Shanghai Advanced Research Institute, Chinese Academy of Sciences, Shanghai 201210, China;
4. Center for Greenhouse Gas and Environmental Engineering, Shanghai Advanced Research Institute, Chinese Academy of Sciences, Shanghai 201210, China

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摘要 采用共沉淀法制备了NiFe₂O₄和NiFe₂O₄/ZrO₂催化剂,用TGA考察了其热化学法,CO₂高温分解反应性能。通过对反应前后催化剂的表征发现,反应高温使两种催化剂都发生了明显的烧结,导致在热还原反应中形成的还原态氧化物不能完全被CO₂氧化从而降低了催化剂的反应性能;ZrO₂的加入对于提高催化剂的热稳定性以及循环反应稳定性具有重要的作用。在高温反应炉中考察了NiFe₂O₄/ZrO₂的CO₂分解实验,结果表明,提高热还原温度可以提高CO产量,然而,随着循环次数的增加CO的产量降低得更明显。

关键词: 热化学循环反应 二氧化碳分解 一氧化碳制备 铁酸镍

Abstract: The thermochemical CO₂ splitting activity of NiFe₂O₄ and NiFe₂O₄/ZrO₂ prepared by the conventional co-precipitation method was investigated with thermogravimetric analysis (TGA) technique. Significant sintering was observed over the two samples during cyclic reactions because of the high reaction temperature. This would lead to an incomplete re-oxidation of the reduced sample in the CO₂ splitting reaction. Introduction of ZrO₂ could greatly enhance the thermal stability of NiFe₂O₄, and hence, the cycling behavior in repeated cycles. The catalytic results of NiFe₂O₄/ZrO₂ for cyclic splitting of CO₂ in a high-temperature furnace indicate that CO productivity increased with the thermal reduction temperature, while the cycling stability severely decreased with the cyclic number.

Key words: thermochemical cyclic reactions CO₂ splitting CO production nickel ferrite

收稿日期: 2013-07-11;

基金资助:

科技部国家科技支撑计划(2013BAC11B02); 国家自然科学基金(21301186)。

通讯作者: 赵宁, Tel: 0351-4049612, E-mail: zhaoning@sxicc.ac.cn. E-mail: zhaoning@sxicc.ac.cn

引用本文:

康敏,张军,赵宁等. 铁酸镍用于热化学循环反应CO₂分解制CO的研究[J]. 燃料化学学报, 2014, 42(01): 68-73.

KANG Min, ZHANG Jun, ZHAO Ning et al. CO production via thermochemical CO₂ splitting over Ni ferrite-based catalysts[J]. J Fuel Chem Technol, 2014, 42(01): 68-73.













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