#### H2O/NaY吸附体系的热化学研究

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摘要 用精密自动绝热量热计测定了在220-320K范围内,不同含水量的H2O/NaY吸附体系的热容. 结果表明,在这些吸附体系的Cp-T曲线上均没有水的固-液相变峰. 这说明即使在饱和吸附的情况下,

水分子仍以单分子层的形态存在于NaY表面上,它们没有形成聚集态.此外,

还测定了往饱和吸附的H2O/NaY中再加入不同量水后所组成的H2O/NaY体系的热容. 在这些Cp-T曲线上都出现了明显的相变峰. 所加之水一旦脱出, 则相变峰又消失.

这些水存在于分子筛颗粒之间只与外表面接触. 但仍受分子筛表面的影响. 所以它们的熔化热、熔化温度均比正常水的

关键词 吸附 分子筛 表面化学 比热 热化学 超分子结构

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## Thermochemical study on H2O/NaY-Zeolite adsorption system

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Abstract An automatic adiabatic colorimeter was used to measure the heat capacity at 220--320 K for the H2O/NaY zeolite adsorption system with various amts. of adsorbed water. The results obtained from CP- T measurements show that there is no peak for the solid-liq. phase transition of adsorbed water. This means that no aggregate state of water is formed on the surface under these conditions and no solid-liq. phase transition can occur. Heat capacity measurements for H2O/NaY zeolite systems were made on samples prepared by mixing mech. the saturated H2O/NaY adsorption system with fixed amts. of water. There are clear peaks on the CP-T curves. If the water mixed in the saturated adsorption system is evacuated at once, the peak for the phase transition on the CP-T curves disappears. These water mols. bind to the external surface of the NaY particles. Interaction of water mols. and zeolite lowers surface of the heat, temperature, and entropy of fusion for these adsorbed H2O mols. compared to bulk water. Water adsorbed on the solid surface differs from normal bulk water.

Key words ADSORPTION MOLECULAR SIEVE SURFACE CHEMISTRY SPECIFIC HEAT THERMOCHEMISTRY SUPERMOLECULAR STRUCTURE

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