

掺杂Ce的TiO₂纳米粒子的光致光及其光催化活性

井立强,孙晓君,蔡伟民,李晓倩,付宏刚,侯海鸽,范乃英

黑龙江大学化学化工学院;哈尔滨工业大学环境科学与工程系

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摘要 采用sol-gel法制备了纯的和掺杂不同量Ce的TiO₂纳米粒子, 并利用XRD, TEM, BET, XPS和PL光谱对样品进行表征, 主要考察焙烧温度和含量对掺杂Ce的TiO₂纳米粒子性质以及光催化降解苯酚活性的影响, 并探讨了Ce的掺杂对TiO₂相变的作用机制以及PL光谱与光催化活性的关系, 结果表明, 掺杂的Ce⁴⁺没有进入到TiO₂晶格中, 而是以小团簇的CeO₂化学态均匀地弥散在TiO₂纳米粒子中, 这可能导致了Ce的掺杂对TiO₂的相变有很大的抑制作用; Ce的掺杂没有引起新的光致发光现象, 而适量Ce的掺杂能够降低TiO₂纳米粒子PL光谱的强度, 这是因为掺杂的Ce⁴⁺易于捕获光生电子而生成Ce³⁺;600℃处理的掺杂Ce的TiO₂纳米粒子表现出较高的光催化活性, 这说明600℃是比较合适的焙烧温度, 而掺杂不同量的Ce的TiO₂样品的光催化活性顺序是: 3 mol% > 4 mol% > 2 mol% > 5 mol% > 1 mol% > 0 mol%, 这与它们的PL光谱强度的顺序是相反的, 即PL光谱强度越低, 其光催化活性越高, 这说明PL光谱与其光催化活性间有着必然的联系, 这是因为掺杂剂Ce⁴⁺能够捕获光生电子, 在光致发光过程中使PL光谱强度下降, 而在光催化反应过程中使有机污染物加快氧化。

关键词 [二氧化钛](#) [纳米材料](#) [掺杂](#) [铈](#) [光催化](#) [苯酚](#) [溶胶-凝胶法](#)

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Photoluminescence of Ce Doped TiO₂ Nanoparticles and Their Photocatalytic Activity

Jing Liqiang, Sun Xiaojun, Cai Weimin, Li Xiaoqian, Fu Honggang, Hou Haige, Fan Naiying

School of Chemistry and Chemical Technology, Heilongjiang University; Department of Environmental Sciences and Engineering, Harbin Institute of Technology

Abstract In this paper, TiO₂ nanoparticles pure and doped with varying content of Ce were prepared by a sol-gel process using Ti(OC₄H₉)₄ as raw material and characterized by XRD, TEM, BET, XPS and PL spectra. We mainly investigated the effects of calcining temperature and Ce content on the properties and the photocatalytic activity for degrading phenol of the TiO₂ nanoparticles. The relationships between PL spectra and photocatalytic activity as well as the mechanisms of Ce doping on TiO₂ phase change were also discussed. The results show that Ce⁴⁺ did not enter into the crystal lattices of TiO₂ and was uniformly dispersed into TiO₂ as the form of CeO₂ with small size, which possibly made Ce dopant have a great inhibition on TiO₂ phase change; Ce dopant did not give rise to a new PL signal, but an appropriate content of Ce could make the intensity of PL spectra down, which was attributed to the ability of Ce⁴⁺ to easily capture the photoinduced electrons to form Ce³⁺; Ce doped TiO₂ nanoparticles calcined at 600 °C exhibited higher photocatalytic activity, indicating that 600 °C was an appropriate calcination temperature. The order of the photocatalytic activity of TiO₂ samples doped with varying content of Ce is as follows: 3 mol% > 4 mol% > 2 mol% > 5 mol% > 1 mol% > 0 mol%, which is opposite to the order of their PL intensity, namely, the weaker the PL intensity, the higher the photocatalytic activity, demonstrating that there are certain relationships between PL spectra and photocatalytic activity. This could be explained by the fact that the dopant Ce⁴⁺ can readily capture the photoinduced electrons, which not only makes the intensity of PL spectra down during the process of photoluminescence but also improves the oxidation speed of organic pollutants during the process of photocatalytic reaction.

Key words [TITANIUM DIOXIDE](#) [NANOPHASE MATERIALS](#) [DOPE](#) [CERIUM](#) [PHOTOCATALYSIS](#) [PHENOL](#) [SOL-GEL PROCESS](#)

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