硫叶立德化合物优势构型和键结构的量子拓扑研究

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摘要 采用MP4(SDTQ)/6-311++G(d,p)和B3LYP/6-311++G(d,p)对所选四种化合 物进行构型优化,从量子拓扑学的角度对各稳定构型进行电子密度拓扑分析,讨论 了C-S键的特性。研究发现: (1) 类硫叶立德自由基(·CHSH_2)和硫叶立德(CH_2SH_2)基态的稳定构型都不具有C_s对称性; (2) 类硫叶立德自由基和硫叶立 德中C-S键的性质类似,硫叶立德中 π 键由两个电子形成,类硫叶立德自由基中 π 键由一个电子形成,所以前者的 π 键性质明显,后者的 π 键性质不明显; (3)类 硫叶立德自由基(·CHSH_2)中单电子 π 键中的电子主要在碳原子附近运动,属于 单电子 π (C \rightarrow S)配键,所以其C-S键的强度比相应的产物要弱。

关键词 硫叶立德 拓扑 化学键 电子密度 构型

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Topological Studies on the C-S Bond of S-Ylide Compounds

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Abstract MP4(SDTQ)/6-311++G(d,p) and B3LYP/6-311++G(d,p) calculations were carried out to optimize the structures of S-ylide compounds. The C-S bond characters were discussed by topological analysis of the electronic density. The following conclusions are drawn: (1) Neither the S-ylide nor the S-ylide-like radical has C_s symmetry. (2) The C-S bond character of S-ylide-like radical is similar to that of S-ylide, there exist π bonds in both types of ylide compounds. However, there are two electrons in the π bond of S-ylide, while there is only one electron in the π bond of S-ylide-like radical. (3) The electron in the π bond of S-ylide-like radical appears mainly near the C atom. It has $\pi(C \to S)$ bond character, therefore it is weaker than the C-S bond in the corresponding product.

Key words S-ylide TOPOLOGY CHEMICAL BONDS ELECTRON DENSITY CONFIGURATION

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