研究论文

双环氧乙烷对三类亚硝胺的羟基化过程的理论研究

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采用量子化学密度泛函(DFT)方法,在B3LYP/6-31G**水平下研究了双环氧乙烷(Dioxirane)、氧化二 甲基亚硝胺(NDMA)、吡咯烷亚硝胺(NPYR)和哌啶烷亚硝胺(NPIP)中的C—H键, 三类亚硝胺化合物均形成a-羟 基化产物的反应机理. 得到三类分子的羟基化反应有syn-和anti-两种进攻方式, 在气相和溶剂(CH₂Cl₂)中, Diox 相关信息 irane氧化三类亚硝胺分子有相对低的能垒,均容易进行a-羟基化.

氧化二甲基亚硝胺 吡咯烷亚硝胺 哌啶烷亚硝胺 密度泛函方法 α-羟基化 关键词 分类号 0641

Theoretical Study of Hydroxylation of Nitrosodimethylami ne, Nitrosopyrolidine, and Nitrosopiperidine by Dioxirane

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Abstract The hydroxylation reaction mechanisms of nitrosodimethylamine, nitrosopyrolidine and nitrosopiperidine by dioxirane were theoretically investigated at the B3LYP/6-31G** level. It is found that there are two paths (separately, syn- and anti-) to the hydroxylation reaction of the e three nitrosoamines that have the same hydroxylation reaction mechanisms. The study of t he potential surface shows that the hydroxylation of the three nitrosoamines by dioxirane ha s a relatively low energy barrier. The result of the theoretical study shows that the a-hydroxyl ation products of these nitrosoamines form easily by dioxirane.

Key words Nitrosodimethylamine(NDMA) Nitrosopyrotidine(NPYR) Nitrosopiperidine(NPIP) Den sity functional method α -hydroxylation

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