研究报告

水-氢同位素汽-气并流催化交换反应动力学研究

阮皓1;李金英1,2;胡石林1;张丽1;窦勤成1

1.中国原子能科学研究院 反应堆工程设计研究所, 北京 102413

2.中国核工业集团公司 综合计划部, 北京 100822

收稿日期 2007-9-19 修回日期 2008-2-25 网络版发布日期: 2008-8-6

摘要 在憎水催化剂的作用下,于固定反应床中研究了水-氢同位素汽-气并流催化交换反应的宏观动力学,讨论了该反应的速率方程和反应级数,比较了反应温度和所研制的3种憎水催化剂对反应速率常数的影响关系。研究结果表明,在本工作所拟订的实验条件下,该反应具有一级反应的动力学特征;温度对反应速率常数的影响服从阿仑尼乌斯公式,温度越高,反应的速率常数越大;Pt-SDB类催化剂的活化能小于Pt-C-PTFE类催化剂。

关键词 憎水催化剂;水蒸气-氢交换;反应级数;动力学 _

分类号 0643.14

Kinetics of Isotopic Exchange Reaction Between Hydroge n and Water Vapor Over Hydrophobic Catalyst in a Co-C urrent Bed

RUAN Hao¹; LI Jin-ying^{1, 2}; HU Shi-Iin¹; ZHANG Li¹; DOU Qin-cheng¹

- 1. China Institute of Atomic Energy, P. O. Box 275(53), Beijing 102413, China;
- 2. China National Nuclear Corporation, P. O. Box 2102, Beijing 100822, China

Abstract The kinetics of isotopic exchange reaction between hydrogen and water vapor over P t-SDB as a hydrophobic catalyst was investigated in a fixed co-current bed. The influence of v arious factors on the rate constant of water vapor-hydrogen co-current exchange reaction were st udied, including rate equation, order of reaction, temperature dependence of reaction and the spe cies of catalysts. The results show that the overall reaction is first order. The relation of apparent r ate constant with temperature accorded with Arrhenius and the apparent rate constant increase s with temperature rising. The apparent activation energy of Pt-SDB is lower than Pt-C-PTFE and the rate constant of water vapor—hydrogen co—current exchange reaction increases when the apparent activation energy of the hydrophobic catalyst decreases.

Key words <u>hydrophobic</u> <u>catalyst</u> <u>water</u> <u>vapor-hydrogen</u> <u>exchange</u> <u>order</u> <u>of</u> <u>re</u> <u>action</u> <u>kinetics</u>

DOI

扩展功能

本文信息

- ► Supporting info
- ▶<u>[PDF全文]</u>(141KB)
- ▶[HTML全文](0KB)
- ▶参考文献

服务与反馈

- ▶把本文推荐给朋友
- ▶文章反馈
- ▶ 浏览反馈信息

相关信息

- ▶ <u>本刊中 包含"憎水催化剂;水蒸</u> 气-氢交换;反应级数;动力学"的 相关文章
- ▶本文作者相关文章
 - 阮皓
 - 李金英
 - 胡石林
 - 张丽
 - 窦勤成