

## 氢化可的松催化超氧阴离子歧化及反应动力学

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**摘要** 在 $0.1\text{ mol}\cdot\text{L}^{-1}\text{NaHCO}_3$ 介质中,用伏安法研究了超氧阴离子 $\text{O}_2^{\cdot-}$ 与糖皮质类甾体氢化可的松的化学反应。实验表明,氢化可的松清除 $\text{O}_2^{\cdot-}$ 的化学作用机制为氢化可的松催化 $\text{O}_2^{\cdot-}$ 的歧化反应,氢化可的松是 $\text{O}_2^{\cdot-}$ 的清除剂。氢化可的松催化 $\text{O}_2^{\cdot-}$ 歧化反应的速率对 $\text{O}_2^{\cdot-}$ 为零级表现反应,对氢化可的松则为二级表现反应,求得 $20^\circ\text{C}$ 时氢化可的松催化 $\text{O}_2^{\cdot-}$ 歧化反应表现速率常数 $k$ 为 $8.76\times 10^5\text{ L}\cdot\text{mol}^{-1}\cdot\text{s}^{-1}$ 。

本结果为医学组织研究结果提供了新的实验证据。在抗炎作用方面,氢化可的松除抑制磷脂酶A2的活性从而间接阻止 $\text{O}_2^{\cdot-}$ 的产生外,还能直接化学清除产生的 $\text{O}_2^{\cdot-}$ ,认为氢化可的松的抗炎作用应是这种生物和化学的综合作用结果。

**关键词** [氢化可的松](#) [表现](#) [伏安法](#) [糖皮质激素](#) [超氧阴离子](#) [歧化反应](#) [反应动力学](#) [自由基清除](#)

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## Dismutation of superoxide anion catalyzed by hydrocortisone and its reaction kinetics

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**Abstract** In  $0.1\text{ mol}\cdot\text{L}^{-1}\text{NaHCO}_3$  medium, the chemical reaction of hydrocortisone with superoxide anion  $\text{O}_2^{\cdot-}$  is studied by voltammetry. Experimental results prove that hydrocortisone, as a scavenger of  $\text{O}_2^{\cdot-}$ , can catalyze the dismutation reaction of  $\text{O}_2^{\cdot-}$ . The apparent reaction progression of the catalytic dismutation is zero for  $\text{O}_2^{\cdot-}$ , but two for hydrocortisone. The equation of rate constant for the catalytic dismutation reaction is deduced, and the apparent rate constant obtained is  $8.76\times 10^5\text{ L}\cdot\text{mol}^{-1}\cdot\text{s}^{-1}$ . The result obtained gives new evidence for biomedical research. Besides the indirect inhibition of  $\text{O}_2^{\cdot-}$  generation by such synthesized glucocorticoids as hydrocortisone suppressing the activity of phospholipase A2, chemically hydrocortisone can also directly scavenge  $\text{O}_2^{\cdot-}$  produced. Therefore, the activity of hydrocortisone as an anti-inflammatory drug is ascribed to the combination of its biologic effectiveness and chemical scavenging for  $\text{O}_2^{\cdot-}$ .

**Key words** [HYDROCORTISONE](#) [APPARENT VOLTAMMETRY](#) [GLUCOCORTICOID DISPROPORTIONATION REACTION](#) [REACTION KINETICS](#) [RADICALS SCAVENGING](#)

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