相转移催化反应研究 V: 利用 1 4C标记法研究取代苯乙烯及 α -芳代苯乙烯与二溴卡宾反应动力学 林守渊,吴毓玲,王执中,盛怀禹

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摘要 本文应用放射性计数进行反应物及产物的定量测定,比较了一些冠醚类相转移催化剂的性能. 发现新型氮杂冠醚在催化二溴卡宾与1,1-二苯乙烯加成反应时有最好的催化性能.本文还测定了取代苯乙烯、α-芳代苯乙烯类化合物与二溴卡宾反应的相对速率常数,并通过线性相关分析,

比较了两类化合物与二溴卡宾反应特性常数的差别,对 α -取代苯乙烯中取代基R的性质和相对速率常数作了比较. 关键词 定量分析 催化 反应动力学 氮杂环化合物 溴化合物 冠式化合物 标记化合物 相转移催化 碳烯 电子效应 加成反应 反应速度常数 苯乙烯 P 放射性 碳同位素

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Studies on phase transfer catalysis V: A kinetic study of the cycloaddition of dibromocarbene with substituted styrenes and α -arylstyrenes by 14C labelling method

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Abstract The relative rate constants krel of the PTC reaction of p-substituted styrenes I (R = H, Me, Cl, MeO, NO2) with dibromocarbene were measured by radiocounting method and the krel values were analyzed by linear correlation method. The r+ value of the PTC reaction of a-arylstyrenes II (R1 = 3-Me, 4-Me, H, 3-Cl, 4-MeO, 3-CF3) with dibromocarbene system was smaller than the one obtained from substituted styrene reaction. It was suggested that the aryl group of a-arylstyrene might disperse the pos. charge from a carbon atom. A variety of the crown compounds were used as PTC catalysts for the reaction of 1,1-diphenylethylene and dibromocarbene and it was found that N,N'-disubstituted azacrown possessed the highest catalytic activity for this reaction. Six 1-14C-1, 1-dibromo-2-aryl-2-phenyl-cyclopropanes were prepared and identified by comparison with the unlabeled authentic samples III which were characterized by elemental anal., NMR, MS, and TLC anal.

Key wordsQUANTITATIVE ANALYSISCATALYSISREACTION KINETICSNITROGENHETEROCYCLICSBROMINE COMPOUNDSCROWN ETHER COMPOUNDSLABELLED COMPOUNDSPHASE TRANSFER CATALYSISCARBENEELECTRONIC EFFECTADDITION REACTIONRATE CONSTANTSTYRENE PRADIACTIVITYCARBON ISOTOPES

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