

论文

γ -环糊精与溴甲酚绿的包合作用

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摘要:

采用紫外-可见分光光度法和等摩尔连续变化法研究了 γ -环糊精与溴甲酚绿的包合作用, 确定了包合物形成的化学计量比为1:2; 采用热力学方法分析了温度与包合常数之间的关系, 计算了包合过程的焓变、熵变及自由能变化分别为-39.988 kJ/mol, 86.400 J/(K·mol)和-14.245 kJ/mol, 这表明疏水作用力为主要驱动力; 采用核磁共振、分子模拟和红外光谱法对包合物进行了研究, 确定了包合物的形成, 分析认为这可能是基团进入 γ -环糊精腔内导致增色效应。

关键词: γ -环糊精 溴甲酚绿 包合物

Complexation of γ -Cyclodextrin with Bromocresol Green

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Abstract:

UV-Vis and continuous variation plot method were used to study the complexation of γ -cyclodextrin with bromocresol green, and the stoichiometric ratio of the complexation was to be 1:2. Thermodynamic analysis results show an inverse relationship between temperature and complexation constant, and the change of enthalpy, entropy and free energy of the complexation were -39.988 kJ/mol, 86.400 J/(K·mol) and -14.245 kJ/mol, respectively, which indicating that hydrophobic effect was the main force to form the complexes. γ -Cyclodextrin and bromocresol green complexes were examined by nuclear magnetic resonance, infrared spectrum and molecular modeling analysis, and was may be the group included in γ -cyclodextrin.

Keywords: γ -Cyclodextrin Bromocresol green Complexation

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1. Xie H., Wang H. Y., Mal Y., *et al.*. Spectro. Chem., Acta Part A: Molecular and Biomolecular

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- Spectroscopy[J], 2005, 62(1—3): 197—202
2. Ishida T., Omata J., Nogami T.. Polyhedron[J], 2003, 22(14—17): 2133—2138
 3. DONG She-Ying(董社英), HUANG Yan-Lin(黄廷林), ZHENG Jian-Bin(郑建斌). Chem. J. Chinese Universities(高等学校化学学报)[J], 2005, 26(11): 2023—2026
 4. Ali S. M., Asmat F., Maheshwart A.. Farmaco[J], 2004, 59(10): 835—838
 5. SUN Wei-Xing(孙维星), LIU Zhi-Qiang(刘志强), SONG Feng-Rui(宋凤瑞). Chem. J. Chinese Universities(高等学校化学学报)[J], 1998, 19(5): 708—710
 6. ZHANG Qiang(张强), LIU Yu(刘育). Chem. J. Chinese Universities(高等学校化学学报)[J], 2004, 25(3): 458—461
 7. KANG Shu(康澍), CHEN Lao(陈涝), SHI Ju(史璐). Chem. J. Chinese Universities(高等学校化学学报)[J], 2007, 28(3): 458—461
 8. YONG Guo-Ping(雍国平), LI Guang-Shui(李光水), ZHENG Fei(郑飞). Chem. J. Chinese Universities(高等学校化学学报)[J], 2000, 21(7): 1124—1126
 9. JIN Zheng-Yu(金征宇). Carbohydrate Chemistry(碳水化合物化学)[M], Beijing: Chemical Industry Press, 2008
 10. Food Standards in Australia and New Zealand. Final Assessment Report, Application A438[S], 2003
 11. Kato T., Horikoshi K.. Analytical Chemistry Acta[J], 1984, 56: 1738—1741
 12. HE Hua(何华), TANG Yao(汤瑶), SUN Cheng(孙成). Acta Chimica Sinica(化学学报)[J], 2006, 64(2): 175—181
 13. ZHANG Jing(张境), LIU Wan-Yi(刘万毅), ZHANG Xia(张霞). Spectroscopy and Spectral Analysis(光谱学与光谱分析)[J], 2006, 26(3): 517—521
 14. YIN Kai-Liang(殷开梁), XU Duan-Jun(徐端钧), CHEN Zheng-Long(陈正隆). Chinese Journal of Inorganic Chemistry(无机化学学报)[J], 2003, 19(5): 480—484

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