研究论文

新型固体电解质Ce_{5.2}RE_{0.8}MoO₁₅₋ 的合成及电性质

周德凤 1,2 , 郭微 2 , 葛志敏 2 , 郝险峰 1 , 柴平 1 , 曹学强 1 , 邢献然 3 , 孟健 1

- 1. 中国科学院长春应用化学研究所稀土化学与物理重点实验室, 长春 130022;
- 2. 长春工业大学生物工程学院, 长春 130012;
- 3. 北京科技大学物理化学系, 北京100083

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摘要 采用溶胶-凝胶方法合成了系列新型氧化物 $Ce_{5.2}RE_{0.8}MoO_{15-\delta}$ (RE=Ce, Y, La, Sm, Gd, Dy, Ho, Er). 通过XRD, Raman和XPS等手段对氧化物的结构进行了表征, 采用交流阻抗谱测试其导电性能. 研究结果表明, RE³⁺的掺杂可增加氧离子的空位浓度, 改善母体电导率, 晶胞参数随RE³⁺半径的增大而增大. 掺杂离子Dy³⁺的 半径(0.0908 nm) 与母体基质离子Ce⁴⁺的半径(0.0920 nm) 相近, 形成的掺杂氧化物晶格弹性应变最小, RE³⁺与氧空位间的缔合焓(ΔH_A)最小, 因而氧化物 $Ce_{5.2}$ Dy_{0.8} $MoO_{15-\delta}$ 具有相对较高的电导率(7.02×10⁻³ S/cm) 和较低的激活能(1.056 eV).

 关键词
 固体电解质
 电导率
 溶胶-凝胶法
 固体氧化物燃料电池

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Synthesis and Electrical Properties of New Solid State Electrolyte Materials $Ce_{5.2}RE_{0.8}MoO_{15-\delta}$

ZHOU De-Feng^{1,2}, GUO Wei²,GE Zhi-Min², HAO Xian-Feng¹, CAI Ping¹, CAO Xue-Qiang¹,XING Xian-Yan³, MENG Jian¹

- 1. Key Laboratory of Rare Earth Chemistry and Physics, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, Changchun 130022, China;
- 2. School of Biological Engineering, Changchun University of Technology, Changchun 130012, China;
- 3. Department of Physical Chemistry, Beijing University of Scitnce & Technology, Beijing 100083, China

Abstract A series of solid state electrolytes, $Ce_{5.2}RE_{0.8}MoO_{15-\delta}(RE=Y, La, Sm, Gd, Dy, Ho, Er)$, were synthesized by sol-gel method. Their structures and electrical conductivities were charac terized by X-ray Diffraction(XRD), Raman and X-ray Photoelectron Spectroscopy(XPS) and AC i mpedance spectroscopy, respectively. The results show that the concentrations of oxygen vac ancy increased with increasing x and their conductivity were improved. And the cell parameter s increase as the radius of RE^{3+} increases. Because the ionic radius of doped RE^{3+} (0.0908 nm) is closed to that of RE^{3+} increases. Because the ionic radius of doped RE^{3+} increases as the least association enthalpy, thus the oxide RE^{3+} and oxygen vacancy, and the system has the least association enthalpy, thus the oxide RE^{3+} increases as the radius of RE^{3+} increases. Solve the ionic radius of doped RE^{3+} increases as the ionic radius of doped RE^{3+} increases. Because the

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通讯作者 孟健jmeng@ciac.jl.cn