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Kinetics of CO oxidation over Pt-CeO_x supported on air-oxidized activated carbon

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Keywords Authors Abstract: Kinetics of low-temperature CO oxidation was studied under atmospheric pressure at 383 K in the initial rates region over 1wt%Pt-1wt%CeO $_{\rm x}$ supported on air-oxidized activated carbon (AC2). Feed concentrations of 1-5 mol% CO and 1-2.5 mol% $\rm O_2$ with balance He were used for CO oxidation in the absence of hydrogen. A simple power-function rate expression was obtained with reaction orders of -0.24 in CO and 0.98 in $\rm O_2$, and a plausible LHHW expression compatible with mechanisms reported in the literature was proposed. Feed concentrations of 5 mol% CO and 2.5 mol% $\rm O_2$ were used for investigating the effect of the presence of $\rm CO_2$ or $\rm H_2$ on CO oxidation rates by partially replacing inert He with 2-4 mol% $\rm CO_2$ or 10-60 mol% $\rm H_2$.



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