研究简报

全氟二异丙基膦酸锂的合成及其电化学性能研究

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摘要 以合成的氯代二异丙基膦为原料,利用电化学全氟化方法,得到全氟二异丙基膦酸锂(Li[(C₃F₇)₂PF₄]),并对其物理和电化学性能进行了研究.

关键词 <u>Simons过程</u> <u>锂离子二次电池</u> <u>电解质</u>

分类号 0621

Synthesis and Electrochemical Properties of Lithium Bis(h eptafluoroisopropyl)tetrafluorophosphate

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Abstract Lithium-ion secondary cell has a high energy desity, stable and high working voltage, wide working temperature and long working term. It is a safe and clean energy resource with out pollution. At present, lithium hexafluorophosphate is used as a conducting electrolyte lithium salt in lithium-ion secondary batteries. But lithium hexafluorophosphate as conducting electrolyte lithium salt has some disadvantages such as hydrolysis and instability. Lithium bis(heptafluoroisopropyl) tetrafluorophosphate $\text{Li}[(C_3F_7)_2\text{PF}_4]$ was received by Simons process from disopropylchlorophosphane in this paper. As an electrolyte of Li ion secondary cell, $\text{Li}[(C_3F_7)_2\text{PF}_4]$ had lower vapor pressure than LiPF_6 in the solvent at the same temperature, comparable conductivity and oxidation stability in the same concentration at room temperature. It was worth mentioning that $\text{Li}[(C_3F_7)_2\text{PF}_4]$ has an excellent stability towards hydrolysis. The synthesis process is safe and easily controlled.

Key words Simons process Lithium ion secondary cell Electrolyte

DOI:

扩展功能

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