研究论文

新型四酰胺杯[4]氮杂冠醚的合成及对阴离子的光化学传感性质研究

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摘要 合成了一类新型四酰胺杯[4]氮杂冠醚(**5a**, **5b**),

用紫外和荧光光谱研究了主体与阴离子客体之间的识别性能. 主体分子 $\mathbf{5a}$ 或 $\mathbf{5b}$ 加入到对硝基酚氧阴离子的 \mathbf{CHCl}_3 溶液中, 主客体间形成超分子复合物使得客体阴离子 $\mathbf{423}$ nm处吸光度降低且溶液从黄色变为无色. 受体 $\mathbf{5a}$, $\mathbf{5b}$ 与 $\mathbf{p-NO}_2\mathbf{C}_6\mathbf{H}_4\mathbf{O}^-$, \mathbf{F}^- , 相互作用, 荧光光谱发生变化, 并且对于 $\mathbf{p-NO}_2\mathbf{C}_6\mathbf{H}_4\mathbf{O}^-$ 具有较好的选择性识别能力,

其荧光光谱强度改变强弱的顺序为p-NO $_2$ C $_6$ H $_4$ O $^-$ >F $^-$ >H $_2$ PO $_4$ $^-$ >>Cl $^-$, Br $^-$, I $^-$.

荧光光谱滴定的结果经非线性拟合,结果表明主客体间形成1:1络合物.

同 四酰胺杯[4]氮杂冠醚 阴离子识别 荧光 传感 合成

分类号

Synthesis and Anion Recognition Properties of Two Novel Tetraamide Calix[4](aza)crowns for Optical Anion Sensor

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Abstract Two new tetraamide calix[4](aza)crowns (**5a**, **5b**) were synthesized and used for optical anion sensors. Addition of host **5a** or **5b** to the $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^-$ anion in CHCl $_3$ resulted in decreasing absorption of band at 423 nm and the yellow solution of $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^-$ anion became colorless because of the formation of host-guest complexes between host **5a** or **5b** and $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^-$. The fluorescence response was observed in the presence of anions such as $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^-$, F⁻,H $_2\text{PO}_4^-$, and the receptors **5a** and **5b** were highly selective toward $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^-$. The degree of change in the emission intensity was in the order of $p\text{-NO}_2\text{C}_6\text{H}_4\text{O}^- > \text{F}^- > \text{H}_2\text{PO}_4^- >> \text{Cl}^-$, Br⁻, I⁻. The fluorescence spectroscopy titration experiments indicated the formation of a 1: 1 stoichiometry complex between receptor **5a** (or **5b**) and anion by nonlinear fitting curve.

Key words tetraamide calix[4](aza)crown anion recognition fluorescence sensing synthesis

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