

研究论文

全氟烷基磺酸酯C—O键断裂的同面S<sub>N</sub>2反应

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**摘要** 用密度泛函理论研究了CF<sub>3</sub>SO<sub>3</sub>CF<sub>2</sub>CF<sub>3</sub>+F<sup>-</sup>的碳氧键断裂反应的机理. 首先,用DFT方法优化了反应物、中间体、过渡态、产物的平衡构型,分析了碳氧键断裂反应的势能面变化.发现在S<sub>N</sub>2反应机理中,除了S—O断裂S<sub>N</sub>2反应外,引起C—O键断裂的同面进攻也是一个可能的反应途径.理论计算表明,最终反应的产物是受热力学控制的,S—O键的断裂绝对地优于C—O的断裂.因此,C—O断裂的同面机理虽然是可能的,但却难以被实验观察到.本文还讨论了端基—F<sub>3</sub>在同面S<sub>N</sub>2反应中的邻位效应,以及基组对这个效应的影响.

**关键词** [反应机理](#) [DFT方法](#) [全氟磺酸酯](#) [同面S<sub>N</sub>2反应](#)

分类号

**Front-Side S<sub>N</sub>2 Reaction, CF<sub>3</sub>SO<sub>3</sub>CF<sub>2</sub>CF<sub>3</sub> + F<sup>-</sup>, Leading to the C—O Cleavage**

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**Abstract** The front-side S<sub>N</sub>2 reaction, CF<sub>3</sub>SO<sub>3</sub>CF<sub>2</sub>CF<sub>3</sub>+F<sup>-</sup>, leading to the C—O cleavage was studied using DFT. The reactant complex, transition state, product complex and products, including their vibration frequencies, were optimized and calculated at B3LYP/6-311G\*\* level. The potential energy surface for the front-side S<sub>N</sub>2 reaction was investigated, showing that the front-side S<sub>N</sub>2 mechanism is possible. However, as shown by computation, the products are thermodynamically controlled, and the S—O cleavage is predominates over the C—O cleavage. As a result, it is difficult for the products resulting from the C—O cleavage to be detected experimentally. Meanwhile, the neighboring effect of the —CF<sub>3</sub> group on the front-side S<sub>N</sub>2 mechanism was found.

**Key words** [reaction mechanism](#) [DFT method](#) [perfluorosulfonate](#) [the front-side S<sub>N</sub>2 reaction](#)

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